



The metabolic activity of denitrifying microorganisms accumulating polyphosphate in response to addition of fusel oil

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Abstract

The effect of distillery waste product (fusel oil) as an alternative external organic carbon source (EOCS) was investigated in terms of the metabolic properties of denitrifying polyphosphate accumulating organisms (DPAOs). Samples of the non-acclimated biomass were collected from a local full-scale wastewater treatment plant employing A²/O type bioreactors. The acclimated biomass was obtained after cultivation (with fusel oil added) in a bench-scale reactor with a process configuration similar to the full-scale bioreactor. Changes in the functional properties of the biomass were investigated by measuring the phosphate release/uptake rates (PRRs and PURs), and nitrate utilization rates (NURs) with fusel oil in anaerobic-anoxic batch tests. Furthermore, a validated extended Activated Sludge Model no 2d (ASM2d) was used as a supporting tool to analyze the experimental results and estimate the contribution of DPAOs to the overall denitrification. In the non-acclimated biomass with fusel oil, the PRRs, PURs and NURs were low and close to the rates obtained in a reference test without adding EOCS. With the acclimated biomass, the PUR and NUR increased significantly, i.e., 3.5 and 2.7 times, respectively. In the non-acclimated biomass, approximately 60.0 ± 3.6% and 20.0 ± 2.2% of the total NUR was attributed to the utilization of endogenous carbon and examined EOCS, respectively. The remaining portion (20% of the total NUR) was attributed to PHA utilization (linked to PO₄-P uptake) by DPAOs. With the acclimated biomass, the contribution of the EOCS to the NUR increased to approximately 60%, while the contribution of the endogenous carbon source decreased accordingly. Very accurate predictions of PURs and NURs ($R^2 = 0.97-1.00$) were obtained with the extended ASM2d. Based on model simulations, it was estimated that the activity of DPAOs and denitrifying ordinary heterotrophic organisms corresponded to approximately 20% and 80% of the total NUR, respectively.

Keywords Alternative carbon source · Fusel oil · Activated sludge · Anoxic N and P removal

Introduction

The successful operation of existing biological nutrient removal (BNR) wastewater treatment plants (WWTPs) depends to a great extent on the availability of sufficient amounts of appropriate (biodegradable) organic compounds (carbon sources). In the combined N and P systems, these compounds should satisfy the needs of three microbial groups, including denitrifying “ordinary” heterotrophs, “ordinary” polyphosphate accumulating organisms (PAOs)—performing enhanced biological P removal (EBPR) in the traditional anaerobic/aerobic cycle—and denitrifying PAOs (DPAOs)—performing EBPR and denitrification in the provision of the anaerobically stored polyhydroxy-alkanoates (PHA). The combination of the two processes by DPAOs is a sustainable solution as it can save aeration and carbon source required for denitrification,

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resulting in the reduction of energy consumption and operational costs [1–4].

Due to the limited amounts of biodegradable organic compounds often encountered in the influent wastewater, the three groups of microorganisms have to compete for the same substrate. To overcome the limitation for denitrification, either “conventional” or “alternative” external organic carbon sources (EOCS). The first group includes commercial products, such as methanol, ethanol, acetic acid, sodium acetate or glucose. Despite their proved efficiency, several recent studies emphasized economical barriers for the practical use in WWTPs [5–8]. Due to high costs of the commercial compounds, various industrial byproducts or waste materials have received more attention as the “alternative” EOCS. Moreover, such an approach is in accordance with the principle of “zero waste” strategy implemented in the European Union to stimulate the development of new waste reuse technologies [9].

Results of numerous studies e.g., [10–18] have indeed confirmed that the characteristics of various branches of the agro-food industry are beneficial for enhancing denitrification in WWTPs in terms of the process rate and efficiency. Gu and Onnis-Hayden (2010) [19] presented a practical evaluation protocol of the alternative EOCS for denitrification. The protocol, recommended by the Water Environment Research Foundation (WERF) for the WWTP operators, provides a guidance to evaluate the feasibility of using specific EOCS in full-scale WWTPs.

However, one of the most important deficiencies of that protocol is that the document does not consider the effect of dosing EOCS on interactions between denitrification and EBPR occurring in the combined N and P systems. The aim of this study was to investigate changes in the metabolic activity of the DPAOs under anaerobic/anoxic conditions in response to addition of a distillery waste product (fusel oil). This compound is potentially a very favorable carbon source for denitrification due to the advantageous composition, i.e., high chemical oxygen demand (COD) concentrations and high COD/N ratios, which has been confirmed in a few earlier studies [20, 21]. However, the anoxic interactions between denitrification and EBPR in the presence of fusel oil have not been reported so far. Furthermore, a complex activated sludge model was used as a supporting tool to analyze the experimental results and estimate the contribution of DPAOs to the overall denitrification under different environmental conditions. The investigations were extended with a microbial analysis using the 16S rRNA PCR-DGGE (Polymerase Chain Reaction-Denaturing Gradient Gel Electrophoresis) technique. To characterize the dominant heterotrophic denitrifying bacteria and DPAOs in the acclimated biomass, clone libraries of *nirS* and *nirK* genes were prepared and analyzed with bioinformatics tools.

Methodology

Origin of the biomass

Non-acclimated biomass

The activated sludge samples were collected from a large municipal WWTP “Wschod” (600,000 PE) in the city of Gdansk (northern Poland). A modified A²/O process configuration is currently employed for BNR in the biological stage. The WWTP meets the European Union effluent criteria for large WWTPs, i.e., effluent concentrations of total N (TN)=10 mgN·L⁻¹ and total P (TP)=1 mgP·L⁻¹.

Acclimation of biomass to fusel oil

The acclimation of processed biomass to fusel oil was conducted in a bench-scale continuous flow reactor with the process configuration similar to the full-scale A²/O bioreactor. The total working volume of the reactor was 30.0 L with the volumetric ratios of 4:9:14 L of the subsequent compartments (anaerobic : anoxic : aerobic). The reactor was fed with the primary effluent from the “Wschod” WWTP and the wastewater inflow rate to the reactor was kept constant at 1.13 L·h⁻¹ to obtain a similar hydraulic retention time (HRT) to the average HRT in the full-scale bioreactor. Fusel oil with the total COD concentration of 1,690,000 mgO₂·L⁻¹ was added to the anoxic compartment in the total amount of 1.5 mL·day⁻¹. The metered amount of fusel oil was mixed with water in proportion of 1:30 and dispensed periodically using a peristaltic pump Heidolph PD 5001 (Schwabach, Germany). During the acclimation period, the return activated sludge and mixed liquor recirculations were, respectively, set to 150% and 500% of the influent flow rate. The solids retention time (SRT) was controlled at a constant level of 20 days based on the mixed liquor suspended solids (MLSS) concentrations in the reactor. Those concentrations varied in the range of approximately 3000–3300 mg·L⁻¹. The set point for dissolved oxygen concentration in the aerobic compartment was 2.0 mgO₂·L⁻¹. In total, the system was operated for over 100 days, but the acclimation period of approximately of 50 days was needed to achieve the maximum NURs. More information about the system performance can be found elsewhere [22]. For the batch experiments with the acclimated biomass, described in the present study, the processed biomass was used after 53 days of acclimation.

External organic carbon sources

The EOCSs examined in the batch experiments comprised the principal substrate for PAOs (acetate) and a fermentation byproduct from distilleries (fusel oil).

The basic characteristics of acetate and fusel oil used in the present study (10 samples from 7 different distilleries) are outlined in Table 1. The composition of the fusel oil revealed a high content of organic compounds (COD) and high value of the COD/TN ratio (approximately 1800). Moreover, the detailed composition of fusel oil was investigated with the gas chromatography with respect to the content of selected 29 organic compounds.

Experimental procedure

A series of two-phase anaerobic-anoxic batch experiments were carried out in plexiglass batch reactors (the working volume of 4.0 L), which were equipped with electrodes and probes (WTW, Germany) for a continuous monitoring of pH (SenTix 21) and temperature (CellOx 325). The temperature set point was 20.0 ± 0.5 °C, ensured by a circulation pump with a controlled cooling water valve. The pH was controlled at 7.0 ± 0.5 which is optimal for the dominance of PAOs [23]. Each experiment lasted 7.5 h, including 2.5 h of the anaerobic phase followed by 5.0 h of the anoxic phase.

The batch experiments were carried out with non-acclimated (Tests 1–5) and acclimated to fusel oil-biomass (Test 6 and 7). Apart from the reference test (Test 1, performed without any EOCS), different combinations of the examined EOCS were dosed at the beginning of both anaerobic and anoxic phase as shown in Table 2. The dosage of acetate (A) and fusel oil (FO) was 0.46 mL and 0.16–0.25 mL, respectively. The dosages of EOCS increased COD concentrations in the reactors in the range of approximately 120 to 200 $\text{mgO}_2\cdot\text{L}^{-1}$.

In each test, at the beginning of the anoxic phase, potassium nitrate (KNO_3) was added in the amount of 525 mg (for the non-acclimated biomass) and 1315 mg (for the acclimated biomass) to increase the nitrate concentration in the reactor by approximately 19 ± 1 and 48 ± 2 $\text{mgN}\cdot\text{L}^{-1}$, respectively. In the latter case, the higher concentrations applied resulted from much higher denitrification capabilities of the acclimated biomass. Consequently, the COD dosages had to be adjusted to obtain the initial COD:N ratios > 4 to avoid the limitation of the denitrification process. This resulted in the range of 120–200 $\text{mg COD}\cdot\text{L}^{-1}$ of the initial COD in the batch experiments. Therefore, the doses of fusel oil varied from 0.16 to 0.25 mL. In both cases, the $\text{NO}_3\text{-N}$ concentrations were sufficient to promote the optimal anoxic $\text{PO}_4\text{-P}$ uptake activity. The characteristics of the initial conditions—mixed liquor of wastewater and activated sludge—in the two-phase anaerobic-anoxic batch experiments are presented in Table 3.

To evaluate the contribution of DPAOs to the increased denitrification rates and efficiency, chemical precipitation of $\text{PO}_4\text{-P}$ was performed in three tests (Table 2), including Tests 4 and 5 (with non-acclimated biomass) and Test 7 (with acclimated biomass). At the end of the anaerobic phase, mixing was turned off to allow for sedimentation of the activated sludge flocs. The liquid supernatant was decanted for precipitation in a separate beaker. The precipitation process was carried out using ferric chloride (III) as a commercial product PIX 113 (Kemira, Finland). The PIX with 35% of ferric chloride solution dose amounted 2.5 mL. After the precipitation, the supernatant

Table 1 Basic characteristics of the different electron donors used in the lab-scale experiments

Parameter	COD ($\text{mgO}_2\cdot\text{L}^{-1}$)	TOC ($\text{mgC}\cdot\text{L}^{-1}$)	TN ($\text{mgN}\cdot\text{L}^{-1}$)	$\text{NH}_4\text{-N}$ ($\text{mgN}\cdot\text{L}^{-1}$)	TP ($\text{mgP}\cdot\text{L}^{-1}$)
Fusel oil	1,690,000–1,890,000	412,000–415,000	960–1040	1.9–2.5	0.2
Acetate	913,000	–	–	–	–

Table 2 Examined EOCS and their dosages (mL) in the two-phase anaerobic-anoxic batch experiment

	Test 1 RT (na)	Test 2 A-FO (na)	Test 3 FO-FO (na)	Test 4 A-FO/PIX (na)	Test 5 FO-FO/PIX (na)	Test 6 FO-FO (a)	Test 7 FO-FO/PIX (a)
Biomass	Non-acclimated biomass (na)					Acclimated biomass (a)	
The dosages of EOCS (mL)							
Anaerobic phase ^a	Reference test (RT) (mixed liquor of wastewater and activated sludge) without EOCS	Acetate (0.46)	Fusel oil (0.16–0.18)	Acetate (0.46)	Fusel oil (0.16–0.18)	Fusel oil (0.25)	Fusel oil (0.25)
Anoxic phase ^b		Fusel oil (0.25)	Fusel oil (0.16–0.22)	Fusel oil (0.25) PIX 113 (2.50)	Fusel oil (0.16–0.18) PIX 113 (2.50)	Fusel oil (0.25)	Fusel oil (0.25) PIX 113 (2.50)

^aThe dosage of EOCS at the beginning of the anaerobic phase

^bThe dosage of EOCS at the beginning of the anoxic phase

Table 3 Characteristics of the initial conditions (mixed liquor of wastewater and activated sludge) in the two-phase anaerobic-anoxic batch experiments (average concentrations \pm standard deviations, $n=3-4$ repetitions for each experiment)

Parameter	Test 1 RT (na)	Test 2 A-FO (na)	Test 3 FO-FO (na)	Test 4 A-FO/PIX (na)	Test 5 FO-FO/PIX (na)	Test 6 FO-FO (a)	Test 7 FO-FO/PIX (a)
Biomass	Non-acclimated biomass (n-ab)					Acclimated biomass (ab)	
COD, (mgO ₂ ·L ⁻¹)	47.0 \pm 3.0	35.0 \pm 3.0	35.0 \pm 2.0	37.0 \pm 4.0	38.0 \pm 3.0	40.0 \pm 5.0	41.0 \pm 4.0
TP (mgP·L ⁻¹)	9.1 \pm 1.3	7.5 \pm 0.9	6.9 \pm 1.1	9.4 \pm 1.5	8.3 \pm 1.4	4.2 \pm 0.8	2.5 \pm 0.5
NO ₃ -N (mgN·L ⁻¹)	0.3 \pm 0.1	0.2 \pm 0.1	0.4 \pm 0.1	0.3 \pm 0.1	0.30 \pm 0.1	0.5 \pm 0.1	0.4 \pm 0.1
NO ₂ -N (mgN·L ⁻¹)	0.08 \pm 0.03	0.02 \pm 0.01	0.03 \pm 0.01	0.02 \pm 0.01	0.03 \pm 0.01	0.01 \pm 0.01	0.03 \pm 0.01
MLSS (mg·L ⁻¹)	2829 \pm 47	3117 \pm 55	3010 \pm 65	3335 \pm 83	2929 \pm 77	3128 \pm 98	3208 \pm 71
pH	7.0 \pm 1.1	7.1 \pm 1.5	7.3 \pm 1.0	7.2 \pm 1.5	6.9 \pm 1.3	7.3 \pm 0.8	7.2 \pm 0.9

was decanted and transferred back to the batch reactor to continue the measurements under anoxic conditions.

Altogether, seven kinds of the experiments (repeated 3–4 times) were carried out, including five and two kinds with the non-acclimated and acclimated biomass, respectively.

Analytical methods

Before each analysis, the withdrawn samples of the mixed liquor were filtered under vacuum pressure through a 1.2 μ m pore size nitrocellulose filter (Millipore, USA). Total nitrogen (TN) concentrations were determined using a TOC/TN analyzer (Shimadzu Corp., Japan). Concentrations of nitrate (NO₃-N) and phosphate (PO₄-P) were determined using cuvette tests in Xion 500 spectrophotometer (Dr Lange GmbH, Germany). The analytical procedures, which were adopted by Dr Lange and Shimadzu, followed the Standard Methods [24]. Total and volatile suspended solids (TSS and VSS) were measured by the gravimetric methods in accordance with the Standard Methods [24].

Model simulations

Model simulations were carried out to estimate the contributions of DPAOs and denitrifying ordinary heterotrophic organisms (DOHOs) to nitrate utilization in each experiment. The GPS-X ver. 5.0.2 program was used as a simulation platform (Hydromantis, Canada). The model used in the present study was an extension of the Activated Sludge Model No. 2d (ASM2d) as proposed by Swinarski et al. (2012) [25]. In the extended model, fusel oil was incorporated as a new state variable ($S_{A,1}$) termed “external readily biodegradable substrate”. The new kind of substrate was introduced to differentiate it from the original ASM2d “fermentation products” (assumed to be acetate), S_A , and to denote that $S_{A,1}$ is not available for PAOs under anaerobic conditions, but it can be consumed by PAOs under anoxic and aerobic conditions. The rationale for introducing that new-state variable was comprehensively discussed in [22,

25]. The expanded ASM2d was implemented in GPS-X using a special utility called “Model Developer”, and subsequently calibrated and validated based on results of the comprehensive laboratory experiments and 96-h measurement campaign in the full-scale bioreactor [25]. In the present study, values of the kinetic and stoichiometric coefficients were adopted from [25] with only minor adjustments for PO₄-P release and uptake. It should be noted that the activity of GAOs was negligible at the studied plant as no significant acetate utilization was observed after polyphosphate depletion in the PAO biomass [22]. Therefore, the GAOs’ metabolism was not incorporated in the model. In addition, in the experiments with the non-acclimated biomass (Tests 1–5), the initial concentrations of PAO (17% of MLVSS) and OHO (32% of MLVSS) were assumed based the simulation results of the full-scale bioreactor. The same values were also assumed for the experiments with the acclimated biomass (Tests 6–7) as the PAO/OHO proportions did not change substantially during the acclimation period as shown by Hu et al. [22].

Calculations

The phosphate release rate (PRR) (–) under anaerobic conditions and the phosphate uptake rate (PUR) (+) under anoxic conditions were calculated from the following equation:

$$\text{PRR or PUR} = \text{slope} \left[(\text{PO}_4 - \text{P})_i \right] \cdot X^{-1} \cdot t^{-1}, \text{ mgP} \cdot g_{\text{VSS}}^{-1} \cdot \text{h}^{-1} \quad (1)$$

where, $(\text{PO}_4 - \text{P})_i$ – PO₄-P concentration in time t_i , mgP·L⁻¹, X -VSS concentration, $g_{\text{VSS}} \cdot \text{L}^{-1}$, t duration of the experiment, h the nitrate utilization rate under anoxic conditions was calculated from the following equation:

$$\text{NUR} = \left[(\text{NO}_3 - \text{N})_i \right] \cdot X^{-1} \cdot t^{-1}, \text{ mgN} \cdot g_{\text{VSS}}^{-1} \cdot \text{h}^{-1} \quad (2)$$

where: $(\text{NO}_3 - \text{N})_i$ – NO₃-N concentration in time t_i , mgN·L⁻¹, contributions to the observed NURs (NUR_{EOCS}, and NUR_{PHA}, and NUR_{Endogenous}) in terms of the examined

SI (Figure SI-1, Table SI-1). For comparison, in the study of Mayer et al. (2015) [33], the composition of fusel oil was very similar and the specific contributions were as follows: $C_5H_{12}O$ (63%), $C_4H_{10}O$ (22%), CH_3CH_2OH (8%), and C_3H_7OH (3%).

Effect of the examined EOCS on EBPR and denitrification

The full PO_4 -P and NO_3 -N profiles observed during the anaerobic-anoxic experiments with the non-acclimated and acclimated biomass in response to dosing the different EOCS (in both anaerobic and anoxic phase) are shown in Fig. 2. The relevant calculations of the process rates and removal efficiencies are summarized in Table 4. A comparison of the PO_4 -P behaviors during the experiments with the non-acclimated and acclimated biomass is presented in Fig. 3.

Phosphate anaerobic release and anoxic uptake rates

The PRR was dependent on the specific EOCS added at the beginning of the anaerobic phase. In the tests with the non-acclimated biomass and acetate dosed (Tests 2 and 4), PO_4 -P was immediately released by the average COD consumption of $68.7 \pm 2.5\%$ (Table 4). The maximum PO_4 -P concentration and PRR reached $38.8 \pm 2.6 \text{ mgP}\cdot\text{L}^{-1}$ (Fig. 2) and $4.4 \pm 0.4 \text{ mgP}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$ (Table 4), respectively. Similar results were recently reported by Zhang et al. (2016) [34] who found that after addition of acetate in the anaerobic phase, the released PO_4 -P concentrations reached 25.0–34.0 $\text{mgP}\cdot\text{L}^{-1}$ by the COD consumption of $84.0 \pm 2.0\%$. Furthermore, the observed PRRs were similar with results presented by Monclus et al. (2010) (2.1 – $5.0 \text{ mgP}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$) [35]. The present results revealed that PAOs in the non-acclimated biomass released up to $10.8 \text{ mgPO}_4\text{-P}\cdot\text{g}^{-1}\text{VSS}$ in response to adding acetate in the anaerobic phase (Fig. 3). These ratios were approximately 2.5 times higher than presented by Coats et al. [36], but 2–3 times lower than obtained by Onnis-Hayden et al. (2008) [37] for similar experiments with dosing of acetate. Wang et al. (2015) [38] showed in experiments with acetate that when the influent P concentration was $5.0 \text{ mgP}\cdot\text{L}^{-1}$, the amounts of anaerobic P release and anoxic P uptake were 4.7 and $5.8 \text{ mgP}\cdot\text{g}^{-1}\text{SS}$, respectively. On the contrary, when fusel oil was dosed in the present study (Test 3), the PO_4 -P profile followed the same low trend as in the reference test (Fig. 3). The addition of fusel oil had only a minor effect on PO_4 -P release and COD consumption in the anaerobic phase with the non-acclimated biomass (Test 3 and Test 5) (Fig. 2; Table 4). The average COD consumption was only $24.7 \pm 2.5\%$. The average $\text{PRR} = 0.65 \pm 0.08 \text{ mgP}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$ was only slightly higher than the rate obtained in the reference test ($0.40 \pm 0.03 \text{ mgP}$.

$\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$) (Table 4). Therefore, non-acclimated PAOs are not implicitly capable of using fusel oil, and the slight P release observed could also be related to the use of their intracellular storage compounds. For comparison, the PO_4 -P profile followed a similar trend as in the experiments with another distillery product (ethanol) as presented by Puig et al. [23] and Swinarski et al. [25]. Swinarski et al. [25] found that ethanol insignificantly induced the anaerobic P release. For comparison, Puig et al. [23] reported that the specific PRR in non-acclimated ethanol biomass from an bench-scale SBR reactor was close to the reference test (1.5 and $0.9 \text{ mgP}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$, respectively).

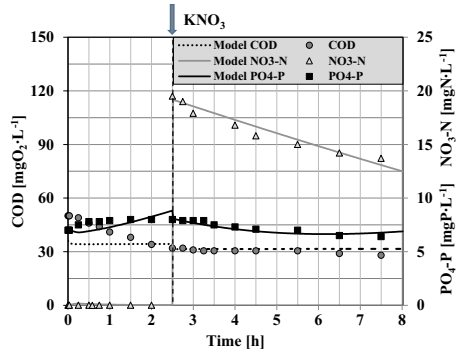
It should be noted that the behavior of PAOs under anoxic conditions was significantly different when adding acetate or fusel oil as the EOCS. When fusel oil was dosed in both phases (Test 3), the anoxic PUR and its removal efficiency was small ($0.60 \pm 0.04 \text{ mgP}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$ and $15.8 \pm 1.5\%$, respectively) and similar to the rate observed in the reference test (Table 4). The low P-release and -uptake rates would confirm that fusel oil was not a suitable carbon source for EBPR in the non-acclimated biomass. On the contrary, the maximum PUR ($2.6 \pm 0.1 \text{ mgP}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$) by high PO_4 -P removal efficiency ($41.8 \pm 1.9\%$) was found in Test 2 after dosing acetate in the anaerobic phase. These observations confirmed the conclusion of Guerrero et al. (2011) [39] that the availability of VFA is the key factor in triggering the EBPR activity and the complex compound (e.g., distillery products) must be fermented to VFAs to maintain the EBPR activity.

An acclimation period of approximately 50 days was required to obtain maximum NURs in the bench-scale bioreactor [22]. During that period, the population of the activated sludge explicitly evolved also towards a more efficient use of fusel oil for the EBPR process as shown in the experiments with the acclimated biomass (Fig. 2). Similar results have been reported in studies with another distillery product [23]. These authors revealed that only after a period of biomass acclimation to ethanol (30–140 days), the population dynamics of the activated sludge evolved to an efficient phosphorus removal process. The present results revealed that PAOs in the acclimated biomass released up to $11.1 \text{ mgPO}_4\text{-P}\cdot\text{g}^{-1}\text{VSS}$ (Test 6) (Fig. 3). After the addition of fusel oil at the beginning of the anaerobic phase (Test 6 and Test 7), a rapid release of PO_4 -P was observed (average value = $39.1 \pm 2.1 \text{ mgP}\cdot\text{L}^{-1}$) (Fig. 2). This ratio is much higher in comparison with the values from the corresponding experiments with the non-acclimated biomass. The maximum PRR was observed during the first 45 min when the readily biodegradable fraction of fusel oil was almost completely utilized (the average COD consumption = $84.3 \pm 2.2\%$) and PUR was similar with experiments when acetate in the anaerobic phase was dosed (Test 2) (Table 4). In comparison with the results with the non-acclimated biomass, the PRRs and PURs increased

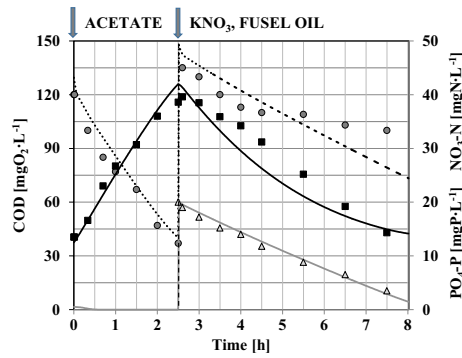
Fig. 2 Behaviors of COD, PO₄-P, and NO₃-N in the two-phase experiments with non-acclimated and acclimated biomass in response to dosing different EOCS

NON-ACCLIMATED BIOMASS

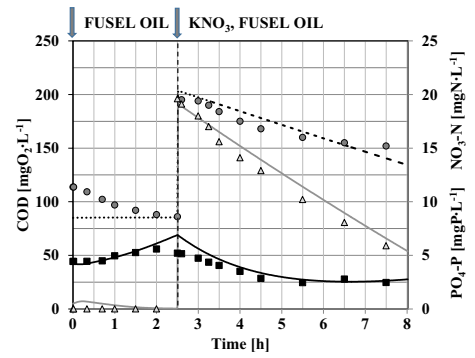
Test 1. RT (na)



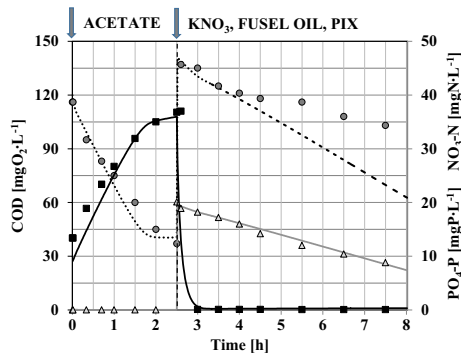
Test 2. A-FO (na)



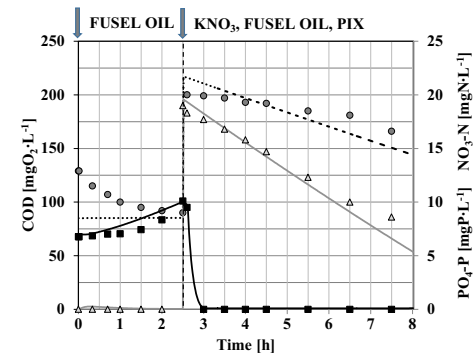
Test 3. FO-FO (na)



Test 4. A-FO/PIX (na)

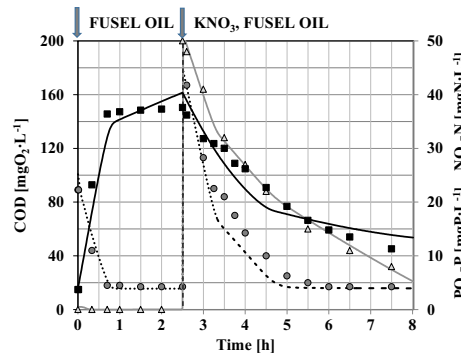


Test 5. FO-FO/PIX (na)



ACCLIMATED BIOMASS

Test 6. FO-FO (a)



Test 7. FO-FO/PIX (a)

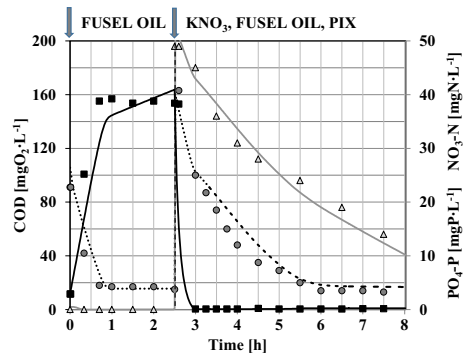
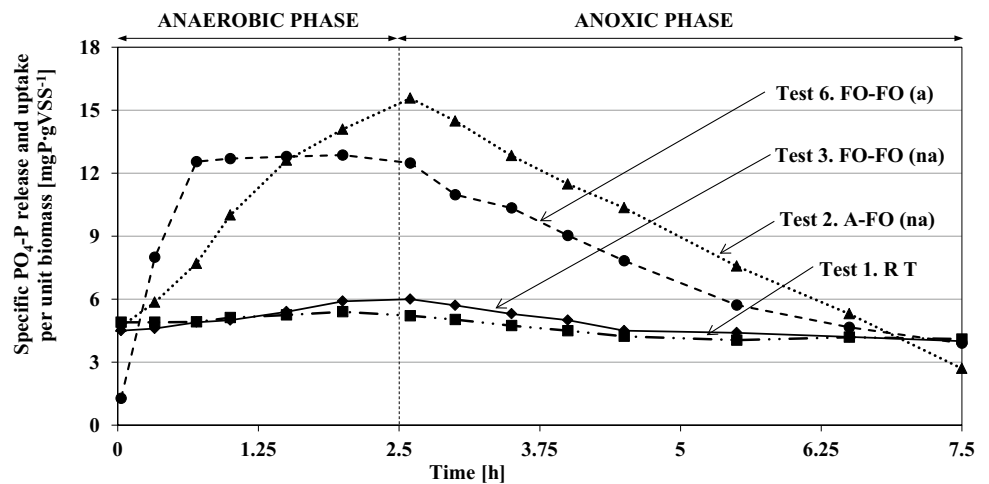


Table 4 Summary of the PRR, PUR, and NUR (average values \pm standard deviations, $n=3-4$ repetitions for each experiment) and removal efficiencies for the non- and acclimated biomass in response to dosing different EOCS

Parameter	Non-acclimated biomass					Acclimated biomass	
	Test 1 RT (na)	Test 2 A-FO (na)	Test 3 FO-FO (na)	Test 4 A-FO/PIX (na)	Test 5 FO-FO/PIX (na)	Test 6 FO-FO (a)	Test 7 FO-FO/PIX (a)
EOCS dosed in the anaerobic phase	Not added	Acetate	Fusel oil	Acetate	Fusel oil	Fusel oil	Fusel oil
PRR, $\text{mgP}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$	0.40 ± 0.03	4.4 ± 0.4	0.70 ± 0.04	4.3 ± 0.6	0.60 ± 0.05	18.3 ± 1.1	17.9 ± 1.3
COD removal efficiency in the anaerobic phase, %	27.2 ± 2.5	69.4 ± 5.5	25.5 ± 3.1	68.1 ± 5.1	23.9 ± 1.7	85.1 ± 3.9	83.5 ± 1.6
$\Delta\text{Prelease}/\Delta\text{CODuptake}$	0.06 ± 0.01	0.27 ± 0.02	0.07 ± 0.01	0.29 ± 0.01	0.08 ± 0.01	0.47 ± 0.02	0.46 ± 0.02
EOCS dosed in the anoxic phase/ chemical $\text{PO}_4\text{-P}$ precipitation	Not added	Fusel oil	Fusel oil	Fusel oil/PIX 113	Fusel oil/PIX 113	Fusel oil	Fusel oil
PUR, $\text{mgP}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$	0.50 ± 0.05	2.6 ± 0.1	0.60 ± 0.04	–	–	2.10 ± 0.06	–
NUR, $\text{mgN}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$	0.70 ± 0.09	1.30 ± 0.06	1.40 ± 0.07	1.00 ± 0.04	1.10 ± 0.09	3.7 ± 0.1	3.0 ± 0.1
$\text{PO}_4\text{-P}$ removal efficiency in the anoxic phase, %	12.3 ± 1.1	41.8 ± 1.9	15.8 ± 1.5	–	–	–	–
$\text{NO}_3\text{-N}$ removal efficiency in the anoxic phase, %	32.5 ± 1.2	82.3 ± 2.0	68.1 ± 1.1	56.0 ± 2.2	44.6 ± 1.5	86.4 ± 1.7	66.2 ± 1.3
COD removal efficiency in the anoxic phase, %	15.5 ± 3.1	25.8 ± 4.2	17.5 ± 1.8	22.2 ± 3.3	15.2 ± 2.1	84.9 ± 2.3	79.2 ± 5.3

Fig. 3 Observed behavior of $\text{PO}_4\text{-P}$ in the two-phase experiments with the non-acclimated biomass (Test 2 and 3) and acclimated biomass (Test 6) in response to dosing different EOCS (without chemical precipitation of $\text{PO}_4\text{-P}$ at the beginning of the anoxic phase)


from 0.70 ± 0.04 to 18.3 ± 1.1 $\text{mgP}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$ and from 0.60 ± 0.04 to 2.10 ± 0.06 $\text{mgP}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$, respectively.

Nitrate utilization rates

The maximum NUR was observed in the experiment with the addition fusel oil in the anoxic phase (followed the addition of acetate in the anaerobic phase). The rate reached 1.40 ± 0.07 $\text{mgN}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$ (Test 3) (Table 4) by the average net $\text{NO}_3\text{-N}$ removal -16.4 ± 0.7 $\text{mgN}\cdot\text{L}^{-1}$ (Fig. 2). Furthermore, the NURs increased in comparison with the reference test by 0.60 ± 0.08 $\text{mgN}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$ (without chemical precipitation of $\text{PO}_4\text{-P}$) (Test 2 and Test 3) and 0.30 ± 0.06 $\text{mgN}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$ (after chemical precipitation of $\text{PO}_4\text{-P}$)

(Test 4 and Test 5) (Table 4). Due to chemical precipitation of $\text{PO}_4\text{-P}$, DPAOs reduced their metabolism, resulting in lower NURs which was necessary for respiration and intracellular transformation. Accordingly, it can be assumed that the difference between the N removal values before and after chemical precipitation of $\text{PO}_4\text{-P}$ could be related to the activity of DPAOs (storage of $\text{PO}_4\text{-P}$ at the expense of PHA), which was approximately 20% of the total NUR.

The results obtained with the acclimated biomass also revealed that $\text{NO}_3\text{-N}$ removal was significantly higher in comparison with the results obtained with the non-acclimated biomass. The observed NURs were apparently increasing from the initial value of less than 1.5 $\text{mgN}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$ [22] and reached the maximum of 3.7 ± 0.1 $\text{mgN}\cdot\text{g}_{\text{VSS}}^{-1}\cdot\text{h}^{-1}$.

h^{-1} on day 53 (Table 4). For comparison, similar acclimation periods (30–50 days) were reported for ethanol by other authors [23, 40]. When chemical precipitation of $\text{PO}_4\text{-P}$ was applied (Test 7), the NUR decreased to $3.0 \pm 0.1 \text{ mgN} \cdot \text{g}_{\text{VSS}}^{-1} \cdot \text{h}^{-1}$, respectively (Fig. 2; Table 4). Therefore, it can be assumed that the difference between the NURs before and after P precipitation was related to the activity of DPAOs ($\text{PO}_4\text{-P}$ uptake at the expense of stored PHA) and contributed to approximately 20% of the total NUR.

During the anoxic phase in Test 2 when acetate was dosed in the anaerobic phase, 1.28 mg $\text{PO}_4\text{-P}$ was consumed to reduce 1 mg $\text{NO}_3\text{-N}$. These values remain in the range of the reported ratios of 0.60–1.31 mg $\cdot \text{mg}^{-1}$ in the EBPR processes with acetate or propionate as a sole EOCs [41–44]. In the present study, acclimated to fusel oil-biomass was characterized by a similar value, i.e., 1.11 mg $\text{PO}_4\text{-P}/1 \text{ mg NO}_3\text{-N}$. For Test 3, with non-acclimated biomass when fusel oil was added in both phases, the consumption of $\text{PO}_4\text{-P}$ to reduce 1 mg $\text{NO}_3\text{-N}$ was similar with the result obtained in the reference test (0.34 mg $\text{PO}_4\text{-P}$ vs. 0.31 mg $\text{PO}_4\text{-P}$). These results revealed that only after acclimation, the population dynamics of the activated sludge evolved to an efficient phosphorus and nitrogen removal process.

Model-based evaluation of the contributions of DPAOs and DOHOs to the NURs

Simulations of the specific experiments

Model simulations of the individual batch experiments and predicted contributions of DPAOs and DOHOs to the NURs are shown in Figure SI-2. Comparison of the observed and predicted PURs and NURs of the two-phase batch experiments are shown in Fig. 4. The observed NURs and PURs

were very accurately predicted ($R^2 = 0.97\text{--}1.00$). Estimated contributions of the specific carbon sources and microbial activities on to the NURs are shown in Fig. 5.

Only two kinetic parameters, including the rate constants for storage of PHA (q_{PHA}) and polyphosphate (q_{PP}), in the original set of model parameters [25] were modified to

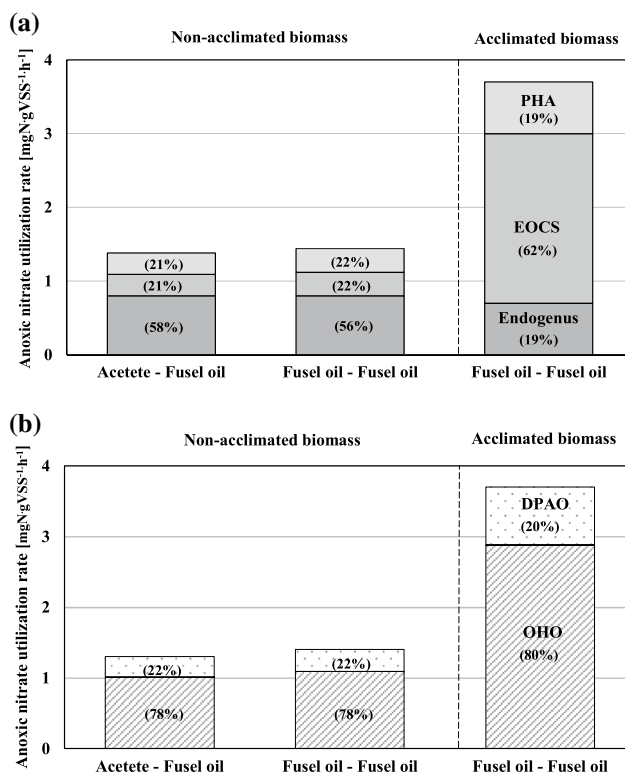
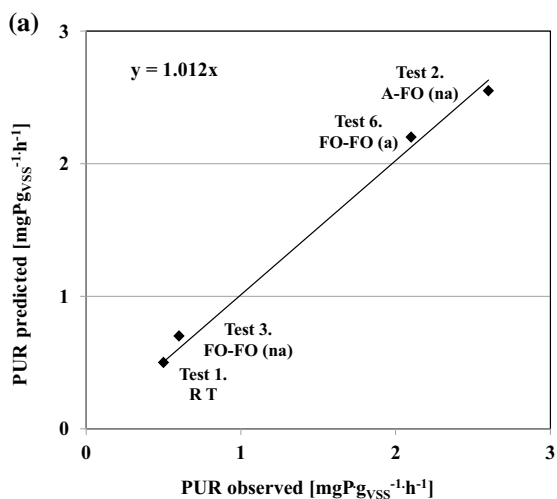


Fig. 5 Estimated contributions of the specific carbon sources (a) and microbial activities (b) on to the NURs

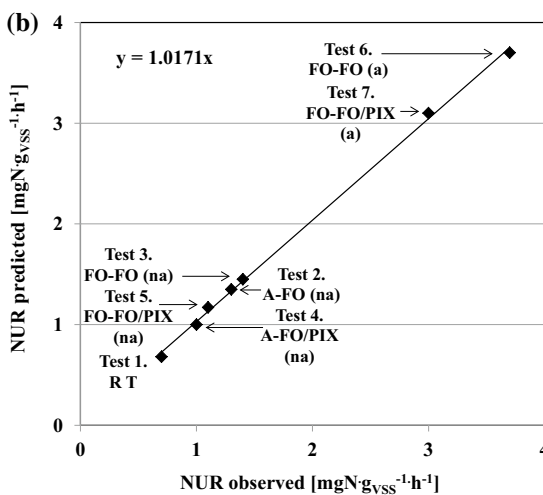


Fig. 4 Comparison of the observed and predicted PURs (a) and NURs (b) during the anoxic phase of the two-phase batch experiments

the values of 2.7 and 2.0 day⁻¹, respectively, for the non-acclimated biomass. The effect of acclimation to fusel oil by PAOs was reflected by significant increases in the observed PRRs. Consequently, the value of q_{PHA} had to be increased to 7 day⁻¹ to accurately predict that effect (Fig. 4).

The highest PAO contributions (> 30%) were predicted for the cases when acetate was added in the anaerobic phase (Figure SI-2). In contrast, the PAO contributions decreased below 20% in the experiments with the fusel oil added in the anaerobic phase (no stored PHA in that phase) as well as in the experiments with the acclimated biomass (better acclimation of OHO to fusel oil). For comparison, in the experiments with real wastewater, Hu et al. [45] and Makinia et al. [46] estimated that the NURs associated with the anoxic activities of DPAO (storage of phosphate and growth) constituted approximately 20% of the NURs associated with the anoxic activity of DOHOs.

Contributions of the specific carbon sources and microbial activities to NURs

In the non-acclimated biomass, approximately 60% and 20% of the total NUR was attributed to the utilization of endogenous carbon sources and examined EOCS, respectively (Fig. 5a). The remaining portion (approximately 20% of the total NUR) could be attributed to PHA utilization (linked to PO₄-P uptake) by DPAOs. For the acclimated biomass, the contribution of the EOCS to the total NUR increased to approximately 60% and the contribution of the endogenous carbon sources decreased accordingly (Fig. 5a). The model simulations revealed that the acclimation rate was similar to both DPAO and DOHO (Fig. 5b). In both non-acclimated and acclimated biomass, the activity of DPAOs and DOHOs corresponded to approximately 20% and 80% of the total NUR, respectively.

Microbial analysis

Microbial community structure stability during acclimation to fusel oil

Potential shifts in the structure of the microbial communities during acclimation to fusel oil were analyzed based on band patterns obtained using the 16S rRNA PCR-DGGE technique (Figure SI-3). In the inoculum sample, 14 unique bands were detected. During the cultivation, only one new band appeared in the sample from the 17th day of the operation. Due to the observed PUR increase in the system, the newly detected microorganism potentially plays a role of flanking species capable of transforming some fusel oil components to VFA. The remaining band patterns were stable during the entire cultivation period. This finding suggests that the applied operational conditions and fusel oil doses

did not reflect selective pressure on activated sludge microorganisms. The increased NURs and PURs resulted from modulation of the bacterial cell physiological activity rather than quality shifts in the microbial community structure.

Phylogenetic analysis of denitrifying bacterial population based on *nirS* and *nirK* genes

The phylogenetic affiliation of denitrifying bacteria in the acclimated biomass from the studied bench-scale reactor with addition of fusel oil were analyzed by the clone libraries approach. As molecular markers, *nirS* and *nirK* genes encoding alternative forms of nitrite reductase were used. Results of the phylogenetic analysis are presented in Figure SI-4 and SI-5. As shown in Table 5, the analyzed PCR amplicons of *nirS* and *nirK* genes formed 28 OTUs (Operational Taxonomic Units > 98% sequence identity) grouped into 8 clades and 23 OTUs grouped in 5 clades, respectively. In the case of both genes, it was revealed that the main role in denitrification is played by members of *Alpha* and *Betaproteobacteria*. A subpopulation of the denitrifying bacteria which contained *nirS* gene was more diverse in comparison with the *nirK* gene holders. Potentially, bacteria which are able to synthesize cd 1-containing nitrite reductase *nirS* may acquire evolutionary advantage in wastewater treatment systems over bacteria harboring *nirK* gene. In terms of the *nirS* gene clone library, the most abundant clades belonged to *Azoarcus* genera and not-yet-classified members of the *Proteobacteria* phylum, each accounted for 22% of the clone library. Members of *Acidovorax sp.*, *Alicyclophilus sp.*, and *Thauera sp.* were also abundant. The clone library of *nirK* gene was outcompeted by the two clades of not-yet-specified bacteria genera; however, they grouped closely with the distinct members of *Rhizobiales* order. OTUs belonging to those clades constituted together more than 60% of the total *nirK* gene clone library. A significant content (23%) of denitrifiers harboring *nirK* gene were assigned to the *Nitrosomonas sp.* genus. Among the analyzed *nirS* and *nirK* gene variants, none reflected similarity to those specified for the “typical” PAO affiliated to *Accumulibacter sp.* Lv et al. (2014) [47] noted that *Dechloromonas sp.*-related microorganisms could potentially play a key role of DPAO in phosphorus removal systems in anaerobic/anoxic conditions. In the present study, their presence was confirmed by a single OTU in the clone library of *nirS* gene.

Conclusions

Fusel oil was proved to be a viable EOCS for the combined nitrogen and phosphorus removal, but an acclimation of microbial community to fusel oil period (> 50 days) was required to effectively utilize that distillery product for

Table 5 Phylogenetic analysis of nirK and nirS genes

Target gene	Cluster	OTU numbers	Unique sequences numbers	Percentage of clone library (%)	Species affiliation
nirS	I	5	7	21.9%	Betaproteobacteria, Rhodocyclales, Zoogloeaceae, Azoarcus sp
	III	6	6	18.8%	Betaproteobacteria, Burkholderiales, Comamonadaceae, Acidovorax sp
	IV	1	1	3.1%	Betaproteobacteria, Burkholderiales, Burkholderiaceae, Cupriavidus sp
	V	1	1	3.1%	Alphaproteobacteria, Rhizobiales, Bradyrhizobiaceae, Bradyrhizobium sp
	VII	5	7	21.9%	Uncultured proteobacteria
	VIII	2	2	6.3%	Alphaproteobacteria, Rhodobacterales, Rhodobacteraceae, Paracoccus sp
	IX	4	4	12.5%	Betaproteobacteria, Burkholderiales, Comamonadaceae, Alicyclophilus sp
	X	1	1	3.1%	Betaproteobacteria, Rhodocyclales, Azonexaceae, Dechloromonas sp
		3	3	9.4%	Betaproteobacteria, Rhodocyclales, Zoogloeaceae, Thauera sp
		Total	28	32	
nirK	I	5	6	23.1%	Betaproteobacteria, Nitrosomonadales, Nitrosomonadaceae, Nitrosomonas sp
	III	8	8	30.8%	Uncultured proteobacteria
	IV	1	1	3.8%	Alphaproteobacteria, Rhizobiales, Rhizobiaceae, Rhizobium sp
	V	3	3	11.5%	Alphaproteobacteria, Rhizobiales, Phyllobacteriaceae, Mesorhizobium sp
	VI	6	8	30.8%	Uncultured alphaproteobacteria
		Total	23	26	

denitrification and EBPR. In comparison with results with the non-acclimated biomass, the PRRs, PURs, and NURs increased from 0.70 ± 0.04 to 18.3 ± 1.1 $\text{mgP.g}_{\text{VSS}}^{-1}.\text{h}^{-1}$, from 0.60 ± 0.04 to 2.10 ± 0.06 $\text{mgP.g}_{\text{VSS}}^{-1}.\text{h}^{-1}$, and from 1.40 ± 0.07 to 3.7 ± 0.1 $\text{mgN.g}_{\text{VSS}}^{-1}.\text{h}^{-1}$, respectively. In both non-acclimated and acclimated biomass, the activity of DPAOs corresponded to approximately 20% of the total NUR. The PCR-DGGE analysis showed a stable structure of microorganism consortia involved in the studied bench-scale bioreactor supported by the addition of fusel oil during acclimation stage. The occurrence of single novel band after 53 days of system operation along with the significantly increased PRRs, PURs, and NURs, reflects no selective pressure on the processed biomass in response to adding fusel oil, while the metabolic stimulation of heterotrophic bacteria associated with the fusel oil conversion to VFA was obtained. The denitrifying subpopulation was dominated by *Alpha* and *Betaproteobacteria* members. Furthermore, a greater diversity of *nirS* gene variants was observed which supports suppositions that bacteria capable of synthesizing nitrite reductase type S obtain the evolutionary advantage in wastewater treatment systems over *nirK* gene harboring microorganisms. The dominant DPAOs were implicitly *Dechloromonas* sp.-related bacteria.

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Compliance with ethical standards

Conflict of interest Agnieszka Tuszyńska, Magdalena Kaszubowska, Przemysław Kowal, Sławomir Ciesielski, and Jacek Makinia declare that they have no conflict of interest.

Research involving human or animal participants This article does not contain any studies with human or animal subjects performed by the any of the authors.

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References

- Wang XL, Peng YZ, Wang SY, Fan J, Cao X (2006) Influence of wastewater composition on nitrogen and phosphorus removal and process control in A2O process. *Bioprocess Biosyst Eng* 28(6):397–404
- Guerrero J, Taya C, Guisasola A, Baeza JA (2012) Understanding the detrimental effect of nitrate presence on EBPR systems:

- effect of the plant configuration. *J Chem Technol Biotechnol* 87:1508–1511
3. Xiao-Mei L, Ming-Fei S, Ji L, Chao-Lin L (2015) Metagenomic analysis of the sludge microbial community in a lab-scale denitrifying phosphorus removal reactor. *Appl Biochem Biotechnol* 175(7):3258–3270
 4. Díez-Montero R, De Florio L, González-Viar M, Volcke EIP, Tejero I (2015) Feasibility of hydraulic separation in a novel anaerobic-anoxic upflow reactor for biological nutrient removal. *Bioprocess Biosyst Eng* 38(1):93–103
 5. Gebremariam SY, Beutel MW, Christian D, Hess TF (2012) Effects of glucose on the performance of enhanced biological phosphorus removal activated sludge enriched with acetate. *Biore-sour Technol* 121:19–24
 6. Kazasi A, Boardman GD, Bott CB (2013) Evaluation of gasoline-denatured ethanol as a carbon source for denitrification. *Water Environ Res* 85(6):549–557
 7. Kapagiannidis AG, Zafiriadis I, Aivasidis A (2013) Comparison between aerobic and anoxic metabolism of denitrifying-EBPR sludge: effect of biomass poly-hydroxyalkanoates content. *New Biotechnol* 30(2):227–237
 8. Gupta PL, Choi HJ, Pawar RR, Jung SP, Lee SM (2016) Enhanced biomass production through optimization of carbon source and utilization of wastewater as a nutrient source. *J Environ Manag* 184:585–595
 9. Zaman AU (2016) A comprehensive study of the environmental and economic benefits of resource recovery from global waste management systems. *J Clean Prod* 124:41–50
 10. Fernández-Nava Y, Marañón E, Soons J, Castrillón L (2010) Denitrification of high nitrate concentration wastewater using alternative carbon sources. *J Hazard Mater* 173:682–688
 11. Fernández FJ, Castro MC, Villaseñor J, Rodríguez L (2011) Agro-food wastewaters as external carbon source to enhance biological phosphorus removal. *Chem Eng J* 166:559–567
 12. Hashemi SE, Heidarpour M, Mostafazadeh-Fard B (2011) Nitrate removal using different carbon substrates in a laboratory model. *Water Sci Technol* 63(11):2700–2706
 13. Czerwionka K, Makinia J, Kaszubowska M, Majtacz J, Angowski M (2012) Distillery wastes as external carbon sources for denitrification in municipal wastewater treatment plants. *Water Sci Technol* 65(9):1583–1590
 14. Lee S, Moon T, Park S, Choi M, Kim C (2013) Evaluation of industrial organic waste as an alternative external carbon source for denitrification in the biological nutrient removal process. *J Chem Eng* 30:1911–1917
 15. Fowdar HS, Hatt BE, Breen P, Cook PLM, Deletic A (2015) Evaluation of sustainable electron donors for nitrate removal in different water media. *Water Res* 85:487–496
 16. Yang XL, Jiang Q, Song HL, Gu TT, Xia MQ (2015) Selection and application of agricultural wastes as solid carbon sources and biofilm carriers in MBR. *J Hazard Mater* 283:186–192
 17. Wang Q, Chen Q (2016) Simultaneous denitrification and denitrifying phosphorus removal in a full-scale anoxic–oxic process without internal recycle treating low strength wastewater. *J Environ Sci* 39:175–183
 18. Zhang Y, Wang XC, Cheng Z, Li Y, Tang J (2016) Effect of fermentation liquid from food waste as a carbon source for enhancing denitrification in wastewater treatment. *Chemosphere* 144:689–696
 19. Gu A, Onnis-Hayden A (2010) Protocol to evaluate alternative external carbon sources for denitrification at full-scale wastewater treatment plants. Water Environment Federation (WERF) Alexandria, USA
 20. Zala SL, Ayer J, Desai AJ (2004) Nitrate removal from the effluent of a fertilizer industry using a bioreactor packed with immobilized cells of *Pseudomonas stutzeri* and *Comamonas testosteroni*. *World J Microbiol Biotechnol* 20:661–665
 21. Makinia J, Czerwionka K, Kaszubowska M, Majtacz J (2014) Distillery fusel oil as an alternative carbon source for denitrification—from laboratory experiments to full-scale applications. *Water Sci Technol* 69(8):1626–1633
 22. Hu X, Wisniewski K, Czerwionka K, Zhou Q, Xie L, Makinia J (2016) Modeling the effect of external carbon source addition under different electron acceptor conditions in biological nutrient removal activated sludge systems. *Environ Sci Technol* 50(4):1887–1896
 23. Puig S, Coma M, Monclus H, van Loosdrecht MCM, Colprim J, Balaguer MD (2008) Selection between alcohols and volatile acids as external carbon sources for EBPR. *Water Res* 42:557–566
 24. APHA (American Public Health Association) (2002) Standard Methods for the examination of water and wastewater. Washington, D.C., USA
 25. Swinarski M, Makinia J, Stensel HD, Czerwionka K, Drewnowski J (2012) Modeling external carbon addition in biological nutrient removal processes with an extension of the International Water Association Activated Sludge Model. *Water Environ Res* 84(8):646–655
 26. Ciesielski S, Bulkowska K, Dabrowska D, Kaczmarczyk D, Kowal P, Mozejko J (2013) Ribosomal intergenic spacer analysis as a tool for monitoring methanogenic Archaea in anaerobic digester. *Curr Microbiol* 67:240–248
 27. Nübel U, Garcia-Pichel F, Muyzer G (1997) PCR primers to amplify 16S rRNA genes from cyanobacteria. *Appl Environ Microb* 63:3327–3332
 28. Throbäck IN, Enwall K, Jarvis A, Hallin S (2004) Reassessing PCR primers targeting *nirS*, *nirK* and *nosZ* genes for community surveys of denitrifying bacteria with DGGE. *FEMS Microbiol Ecol* 49(3):401–417
 29. Hallin S, Lindgren PE (1999) PCR detection of genes encoding nitrite reductase in denitrifying bacteria. *Appl Environ Microbiol* 65(4):1652–1657
 30. Rice P, Longden I, Bleasby A (2000) EMBOSS: the european molecular biology open software suite. *Trends Genet* 16(6):276–277
 31. Tamura K, Stecher G, Peterson D, Filipowski A, Kumar S (2013) MEGA6: molecular evolutionary genetics analysis version 6.0. *Mol Biol Evol* 30(12):2725–2729
 32. Saitou N, Nei M (1987) The neighbor-joining method: a new method for reconstructing phylogenetic trees. *Mol Biol Evol* 4(4):406–425
 33. Mayer FD, Feris LA, Marcilio NR, Staudt PB, Hoffmann R, Baldo V (2015) Influence of fusel oil components on the distillation of hydrous ethanol fuel (HEF) in bench column. *Braz J Chem Eng* 32(2):585–593
 34. Zhang M, Yang Q, Zhang J, Wang C, Wang S, Peng Y (2016) Enhancement of denitrifying phosphorus removal and microbial community of long-term operation in an anaerobic anoxic–biological contact oxidation system. *J Biosci Bioeng* 122(4):456–466
 35. Monclus H, Sipma J, Ferrero G, Rodriguez-Roda I, Comas J (2010) Biological nutrient removal in an MBR treating municipal wastewater with special focus on biological phosphorus removal. *Biore-sour Technol* 101:3984–3991
 36. Coats ER, Watkins DL, Brinkman CK, Loge FJ (2011) Effect of anaerobic HRT on biological phosphorus removal and the enrichment of phosphorus accumulating organisms. *Water Environ Res* 83(5):461–469
 37. Onnis-Hayden A, Majed N, McMahon KD, Gu AZ (2008) Phosphorus removal and PAOs populations at a full-scale integrated fixed-film activated sludge (IFAS) plant, 81st Annual Water

- Environment Federation Technical Exhibition and Conference, Chicago, Illinois
38. Wang Z, Meng Y, Fan T, Du Y, Tang J, Fan S (2015) Phosphorus removal and N₂O production in anaerobic/anoxic denitrifying phosphorus removal process: long-term impact of influent phosphorus concentration. *Bioresour Technol* 179:585–594
 39. Guerrero J, Guisasola A, Baeza JA (2011) The nature of the carbon source rules the competition between PAO and denitrifiers in systems for simultaneous biological nitrogen and phosphorus removal. *Water Res* 45(16):4793–4802
 40. Wang DB, Zheng W, Li XM, Yang Q, Liao DX, Zeng GM (2013) Evaluation of the feasibility of alcohols serving as external carbon sources for biological phosphorus removal induced by the oxic/extended-idle regime. *Biotechnol Bioeng* 110(3):827–837
 41. Zhang H, Jiang J, Li M, Yan F, Gong Ch, Wang Q (2016) Biological nitrate removal using a food waste-derived carbon source in synthetic wastewater and real sewage. *J Environ Manag* 166:407–413
 42. Carvalho G, Lemos PC, Oehmen A, Reis MAM (2007) Denitrifying phosphorus removal: linking the process performance with the microbial community structure. *Water Res* 41(19):4383–4396
 43. Zhou S, Zhang X, Feng L (2010) Effect of different types of electron acceptors on the anoxic phosphorus uptake activity of denitrifying phosphorus removing bacteria. *Bioresour Technol* 101:1603–1610
 44. Wong PY, Cheng KY, Kaksonen AH, Sutton DC, Ginige MP (2013) A novel post denitrification configuration for phosphorus recovery using polyphosphate accumulating organisms. *Water Res* 47:6488–6495
 45. Hu ZR, Wentzel MC, Ekama GA (2002) The significance of denitrifying polyphosphate accumulating organisms in biological nutrient removal activated sludge systems. *Wat Sci Technol* 46(1–2):129–138
 46. Makinia J, Rosenwinkel KH, Swinarski M, Dobiegala E (2006) Experimental and model-based evaluation of the role of denitrifying PAO at two large scale WWTPs in northern Poland. *Water Sci Technol* 54(8):73–81
 47. Lv XM, Shao MF, Li CL, Li J, Gao XL, Sun FY (2014) A comparative study of the bacterial community in denitrifying and traditional enhanced biological phosphorus removal processes. *Microbes Environ* 17(3):261–268