Post-print of: Siebert A., Cholewiński G., Trzonkowski P., Rachoń J.: Immunosuppressive properties of amino acid and peptide derivatives of mycophenolic acid EUROPEAN JOURNAL OF MEDICINAL CHEMISTRY Vol. 189 (2020) pp. 1-17 DOI: 10.1016/j.ejmech.2020.112091

Immunosuppressive properties of amino acid and peptide derivatives of mycophenolic acid

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Abstract

Mycophenolic acid (MPA) was coupled with amino acids and biologically active peptides including derivatives of tuftsin to modify its immunosuppressive properties. Both amino acid unit in the case of simple MPA amides and modifications within peptide moiety of MPA - tuftsin conjugates influenced the observed activity. Antiproliferative potential of the obtained conjugates was investigated *in vitro* and MPA amides with threonine methyl ester and conjugate of MPA with retro-tuftisin occurred to be more selective against PBMC in comparison to parent MPA. Both amino acid and peptide derivatives of MPA acted as inosine-5'-monophosphate dehydrogenaze (IMPDH) inhibitors.

Keywords

Mycophenolic acid; amino acids; peptides; tuftsin; IMPDH inhibitors; immunosuppressants.

1. Introduction

Mycophenolic acid (MPA) **1** (Fig. 1) was isolated first time by B. Gosio from *Penicilium brevi-compactum*. It possesses antibacterial, antiviral, anticancer and immunosuppressive properties [1,2].

Fig. 1 Structure of mycophenolic acid.

MPA is an uncompetitive and reversible inhibitor of inosine-5'-monophosphate dehydrogenase (IMPDH) and its prodrugs are applied in clinic as immunosuppressants. In 1995 morpholine ester of mycophenolic acid – mycophenolate mofetil (MMF, CellCept ®, Roche AG) was approved by FDA as drug in solid organ transplantation (kidney, liver, heart) for decrease risk of rejection. The second form of the drug is mycophenolic acid sodium salt (MPS, Myfortic ®, Novartis Farma AG). Both forms are used together with other immunosuppressants, like cyclosporine, tacrolimus in transplantation and autoimmune disorders treatment, e.g. psoriasis [1-4]. MPA inhibits IMPDH via blocking binding site of NAD⁺ cofactor placed near to active center of the enzyme. The structure of IMPDH enzyme in complex with MPA 1 was reported [5] and the role of functional group of 1 in maintenance of activity was explained, like free phenol or carboxylic groups, which are able to hydrogen bond interactions with the enzyme. Biosynthesis of lymphocytes and DNA depends on IMPDH activity, since it involves nucleotides biosynthesis de novo. Other cells use both de novo and salvage pathway, when nucleobases are recycled. As a result, MPA selectively inhibits proliferation of lymphocytes B and T [1-4]. Furthermore, IMPDH exists in the two isoforms I and II, where IMPDH I is expressed in normal cells, and IMPDH II is up-regulated in activated lymphocytes and neoplastric cells. MPA inhibits both forms with higher activity against IMPDH II [6]. As a result, numerous IMPDH inhibitors revealed not only immunosuppressive, but also anticancer properties [7-11].

Despite of the progress in immunosuppressive treatment, both the risk of rejection and serious side – effects have been not eliminated so far. As a result, numerous studies were performed to increase efficiency and diminish toxicity of novel mycophenolic acid derivatives [12-19].

In our previous work we designed amino acid MPA derivatives possessing potent immunosuppressive activity [20]. These results were in agreement with literature data, that polar group at the end of side chain in MPA is important for maintenance of anti-proliferative activity, since enables hydrogen bond interactions with Ser 276 of IMPDH [21].

On the other hand, tuftsin is an endogenous tetra-peptide with the sequence Thr-Lys-Pro-Arg, naturally occurring in human blood, which can influence immune properties [22-26]. Moreover, conjugate formation can provide substance with optimized activity, including improved



potency and reduced toxicity [27-32]. Therefore, we decided to obtain and investigate amino acid and peptide derivatives of MPA as promising immunosuppressive agents.

2. Results and discussion

2.1 Chemistry

Synthesis of amino acid derivatives **2a-g** was performed *via* coupling of amino acid esters with MPA. Then, respective methyl esters were hydrolyzed to free acids **2h-m** according to previously published procedure [20,25].

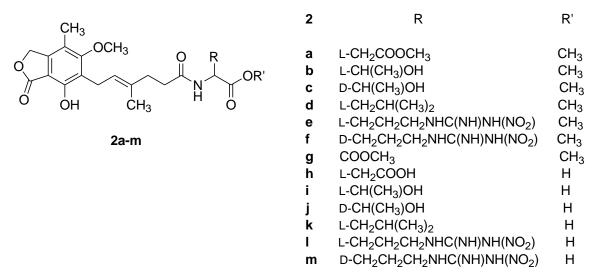


Figure 2. Amino acid derivatives of MPA 2a-m.

The next series of amide derivatives of MPA consisted peptide analogs, based on tuftsin **3a-g** and retro-tuftsin **3h-n** (Fig. 3). However, tuftsin is not stable and its half – time in plasma is 16 minutes. Therefore, we included also retro tuftsin, a peptide with reversed sequence Arg-Pro-Lys-Thr. Moreover, tuftsin can be stabilized by bonding with biological active compounds or modification at ε -amine group in lysine moiety, where next amino acid can be attached [22-24]. As a results, we designed pentapeptides **5a-n** (Fig. 4) through acylation of ε -amine group of lysine with amino acids like α -alanine, β -alanine, valine, leucine, isoleucine with methods described by Dzierzbicka and co-workers [22-24].

Then, Fmoc-protected peptides **5a-n** were combined with MPA by means of condensing agents such as EDCI, EEDQ or T₃P. The coupling of Fmoc-protected tuftsin analogs **6a-g** (Fig. 5) with MPA **1** was optimized with T₃P procedure, whereas the most effective coupling reagent in the case of retro-tuftsin conjugates **6h-n** occurred to be EDCI in the presence of DMAP. The reaction carried out in anhydrous DMF gave the F-moc protected derivatives **6a-n** in 39-65 % yield. In order to produce the final compounds **3a-n**, Fmoc protecting group was removed with 20-30% solution of diethylamine in chloroform in 80-85% yield.



The structures and purities was confirmed with ¹H NMR, ¹³C NMR, MS, MS-HPLC techniques. Additionally, characteristics were extended with COESY, HMBC, HSQC, ROESY, and TOCSY measurements. Examples of NMR, MS spectra, HPLC chromatograms are available in the supplementary material.

Figure 3. Peptides derivatives of MPA 3a-n.

Figure 4. Synthesis of peptide derivatives of MPA 3a-n.



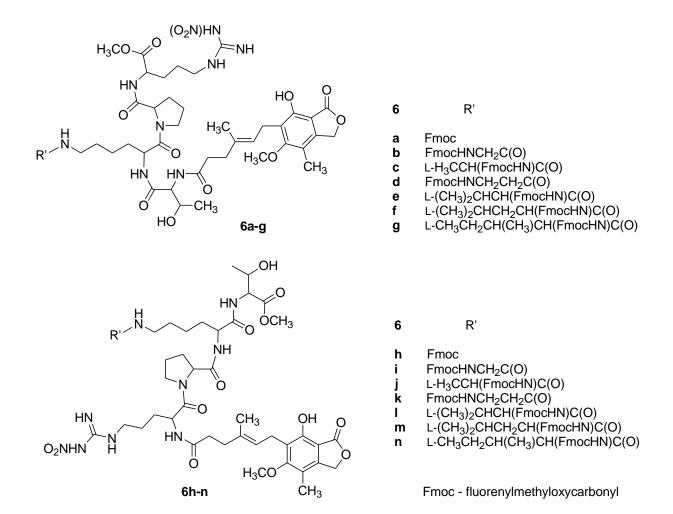


Figure 5. Fmoc – protected conjugates of MPA with tuftsin 6a-g and retro-tuftsin 6h-n.

2.2. Immunosuppressive evaluation

To assess the immunosuppressive activity, we performed both MTT cytotoxicity assay and a proliferation test using the VPD450 dye on the Jurkat lymphoidal cell line and peripheral blood mononuclear cells (PBMCs). Jurkat cells are used as a model line of human lymphocytes, whereas PBMCs are an *in vitro* model of the immune response. These tests were performed in triplicate.

Compounds were dissolved in 20 μ L of DMSO, then refilled with media to 1 mL, resulting in a starting concentration of 1 mg/mL. Then, further solutions were prepared by serial dilution, obtaining the lowest tested concentration of 10^{-7} mg/mL.

2.2.1. Determination of cytotoxic activity of tested compounds against Jurkat and PBMC cell lines using the MTT method

MPA derivatives at various concentrations were incubated with Jurkat or PBMC cells. In addition, the proliferation of human lymphocytes was stimulated with anti-CD3 / anti-CD-28 antibodies. Then, MTT was used and IC_{50} established from the obtained data (Tables 1,2).



2.2.1.1. Amino acid analogs of MPA

Table 1 shows the results of the MTT cytotoxicity assay, which was performed for the amino acid analogs of MPA **2a-m** on the T-Jurkat cell line and on PBMCs. In the case of PBMCs, all new amino acid analogs of the MPA were found to be less cytotoxic than the parent MPA **1**.

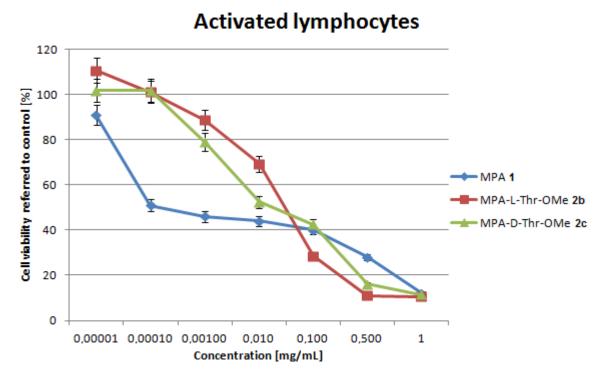


Figure 6. Comparison of cytotoxicity of compounds **2b** and **2c** – amino acid derivatives with the best SI (see Table 5) to MPA **1** against PBMC.

In contrast, compounds 2a, b, d, i gave higher toxicity against T-Jurkat cell line in comparison to 1. Analyzing the results it can be seen that the amino acid methyl esters 2a-g were more cytotoxic than their counterparts with the free carboxyl group 2h-m. This is noticeable in each case for both the T-Jurkat cell line and PBMC cells. The absolute configuration in amino acid moiety also influenced cytotoxicity. Both in the case of methyl esters 2a-g and derivatives with a free carboxyl group 2h-m, it can be seen that D enantiomers are less cytotoxic than L enantiomers for both types of tested cells.

MPA derivatives with arginine bearing free carboxyl group, both L **2l** and D **2m** enantiomer, showed the lowest toxicity. In the case of the T-Jurkat cell line for these derivatives, the IC₅₀ values in at used concentrations could not be determined. Thus, the viability at the highest tested concentration (1 mg/mL) was calculated. The cell viability was 59.12% for compound **2l** and 75.14% for **2m**. These analogs indicated also low cytoxicity to PBMCs, and their IC₅₀ values were 283.35 μ M for **2l** and 422.2 μ M in case of **2m**.



The MPA-L-Ile-OMe **2d** was found to be the most cytotoxic to both types of cells, its IC₅₀ for the Jurkat cell line was 8 μ M (approximately 8 times lower than for MPA **1**), whereas against PBMC cells **2d** gave 2.2 μ M. Figure 6 shows a comparison of the cytotoxicity of compounds **2b** and **2c** to mycophenolic acid **1** at the tested concentrations against activated lymphocytes. Both threonine esters **2b**,**c** were less cytotoxic than MPA **1** at lower concentrations, but in the range of 0.01 to 1g/mL observed activity was similar.

Table 1. IC_{50} [μM] values of amino acid derivatives of MPA **2a-m** based on MTT test.

		T-JURKAT				PBMC		
Compound	No	IC ₅₀ [μM]	Viability [%]*	р	F	IC ₅₀ [μM]	р	F
MPA	1	60.6 ± NAN				0.14 ± 0.06		
MPA-L-Asp(OMe)-OMe	2a	20 ± 3		< 0.05	1.54	18 ± 1.8	< 0.05	33.5
MPA-L-Thr-OMe	2b	17 ± 9		< 0.05	188	57 ± 31.4	< 0.05	30.1
MPA-D-Thr-OMe	2 c	115 ± 3.6		< 0.05	266	69 ± 58.2	< 0.05	25.9
MPA-L-Ile-OMe	2d	8 ± 5.7		< 0.05	8775	2.2 ± 2	< 0.05	46.6
MPA-L-Arg-OMe	2 e	346.98 ± NAN		0.2	14.1	14 ± 4.8	< 0.05	40.2
MPA-D-Arg-OMe	2f	572 ± 92.9		< 0.05	186	71 ± 63.1	< 0.05	29.5
MPA-Mal-(OMe) ₂	2g	101 ± 20.8		< 0.05	3128	3.5 ± NAN	< 0.05	28.5
MPA-L-Asp(OH)-OH	2h	539.67 ± NAN		< 0.05	236	50 ± 25.7	< 0.05	42.1
MPA-L-Thr-OH	2i	43 ± 5.4		< 0.05	420	82 ± 58.2	< 0.05	25.5
MPA-D-Thr-OH	2j	229 ± 35.9		0.105	50.5	118 ± 114.9	< 0.05	3.99
MPA-L-Ile-OH	2k	73.74 ± NAN		0.083	80.9	20.87 ± NAN	< 0.05	10.0
MPA-L-Arg-OH	21		59.12%			283.35 ± NAN	< 0.05	25.5
MPA-D-Arg-OH	2m		75.14%			422.2 ± NAN	< 0.05	50.3

 $\overline{\text{IC}_{50}}$ higher than for MPA, p – statistical significance of the difference, F – Fisher test to MPA.

^{*} cell viability calculated at highest tested concentration 1 mg/mL

2.2.1.2. Peptide derivatives of MPA

Table 2 presents the IC₅₀ values obtained for cytotoxicity studies of peptide analogs of MPA **3a-n**. All peptide analogs of MPA are much less toxic than mycophenolic acid **1**.

In the case of more than half of these compounds, determination of IC₅₀ against Jurkat cell line values at the concentration under investigation failed. Therefore, cell viability in the presence of these compounds at the highest concentration of 1 mg/mL was established and gave 52.48% for compound 3n to 85.82% for 3a. Peptide derivatives exhibited IC₅₀ values against the Jurkat cell line are over 13 times if compare to MPA 1. Contrary, in the case of PBMCs, IC₅₀ values are over 3500 times higher than for MPA 1. This indicates a minimal, in comparison to MPA 1, cytotoxic activity of peptide analogs against PBMCs.

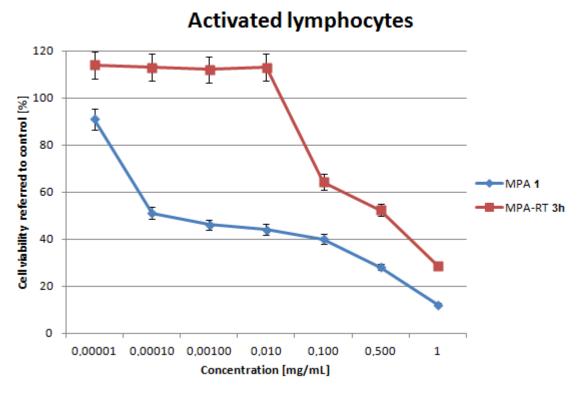


Figure 7. Comparison of cytotoxicity of compound 3h – peptide derivative with the best SI (see Table 5) to MPA 1 against PBMC.



Table 2. IC_{50} [μM] values of peptide MPA derivatives 3a-n based on MTT test.

		T-JURKAT				PBMC			
Compound	No	IC ₅₀ [μM]	Viability [%]*	р	F	IC ₅₀ [μM]	р	F	Viability [%]*
MPA	1	60.56 ± NAN				0.14 ± 0.056			
MPA-T	3 a		85.82%				-		99.09%
MPA-T-Gly	3b		53.40%			628 ± 57.5	< 0.05	21.4	
MPA-T-αAla	3c	1001 ± 110		0.099	57.0	560 ± 118.9	< 0.05	42.2	
MPA-T-βAla	3d	1020 ± 51		< 0.05	4674	526 ± 7.9	< 0.05	27.6	
MPA-T-Val	3e		66.56%						56.81%
MPA-T-Leu	3f		67.04%			687 ± 152.8	< 0.05	12.0	
MPA-T-Ile	3g	808.73 ± NAN		30.8	0.134	461 ± 199	< 0.05	11.2	
MPA-RT	3h	971 ± 108.9		0.116	41.0	633.9 ± 0.13	< 0.05	63.6	
MPA-RT-Gly	3i	1096 ± 83.9		< 0.05	284	865 ± 492.7	< 0.05	40.4	
MPA-RT-αAla	3 j		53.90%			750 ± 88.6	< 0.05	36.0	
MPA-RT-βAla	3k	786 ± 75.7		0.114	42.9	587.64 ± NAN	< 0.05	42.8	
MPA-RT-Val	31		63.35%			548 ± 56.5	< 0.05	108.5	
MPA-RT-Leu	3m		61.34%			519 ± 390.6	< 0.05	86.6	
MPA-RT-Ile	3n		52.48%			533 ± 182.6	< 0.05	45.1	

IC₅₀ higher than for MPA, p – statistical significance of the difference, F – Fisher test to MPA.

^{*} cell viability calculated at highest tested concentration 1mg/mL

We observed higher toxicity of MPA analogs with retro-tuftsin **3h-n** than compounds bearing tuftsin **3a-g**. An exception constituted two pentapeptide derivatives of MPA possessing α -alanine and isoleucine units against T-Jurkat cell line and one pentapeptide analog MPA with glycine against PBMCs, where tuftsin conjugates were more toxic. According to these results, we conclude that MPA conjugates hold retro-tuftsin showed slightly higher cytotoxicity than tuftsin counterparts. The MPA-T 3a derivative was not toxic against both types of cells. Figure 7 shows a comparison of the cytotoxicity of the MPA conjugate having retro-tuftsin **3h** to the parent MPA 1 in the range of considered concentrations against the activated lymphocytes. Conjugate 3h gave cytotoxicity lower than parent MPA 1, especially within the range of concentration 0.00001 - 0.1 g/mL.

2.2.2. Determination of antiproliferative activity of tested compounds 2a-m, 3a-g and 3h-n against Jurkat and PBMC cell lines using VPD450.

In this method VPD450 dye undergoes distribution between child cells uniformly, so that each progeny cell retains about half of the fluorescent intensity of the VPD450 of its parent cell. Fluorescence measurements were then made using flow cytometry to check cell proliferation. After that, the EC_{50} value was determined from the received data (Tables 3,4).

2.2.2.1. Amino acid MPA derivatives

In the Table 3 are collected EC₅₀ values of amino acid analogues of MPA. None of these compounds showed such a good anti-proliferative effect as mycophenolic acid 1. The results of tests carried out on the Jurkat cell line and on PBMCs are consistent. In both cases, the most active compounds occurred to be 2a,b,c,d,k. The MPA derivative with isoleucine gave the highest activity, where methyl ester 2d (EC₅₀ = 5.12 μ M for PBMC) and free acid 2k (EC₅₀ = 9 μ M for PBMCs) were characterized by the lowest EC₅₀ value. In contrast to that, the least active derivatives of MPA in relation to both types of cells were found to be 2h,l,m.



Activated lymphocytes

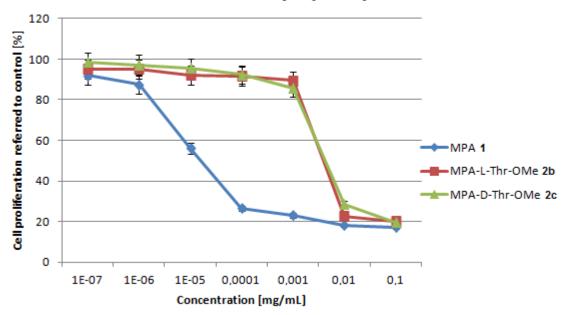


Figure 8. Comparison of antiproliferative activity of compounds **2b**, **c** – amino acid derivatives with the best SI (see Table 5) to MPA **1** against PBMC.

Again, the deprotection of methyl esters influenced the observed activity. Methyl esters **2,a,b,c** were mostly much more active than their counterparts with a free carboxyl group **2h,i,j**. It suggests the significant role of cell membrane penetration [11]. On the other hand, MPA-L-Ile-OH derivative **2k** ($EC_{50} = 2.3 \mu M$) was characterized by a lower EC_{50} value than MPA-L-Ile-OMe **2d** ($EC_{50} = 9 \mu M$) for the T-Jurkat cell line. According to these results, bulky *sec*-butyl substituent in isoleucine moiety was advantageous for antiproliferative activity and gave together with polar carboxylic group the lowest EC_{50} within amino acid MPA derivatives **2a-m**.

The effect of configuration in amino acid unit on anti-proliferative activity was lower if compared with the cytotoxicity test. However, only in the case of the MPA-Thr-OH derivative (against Jurkat cell line), the D enantiomer **2j** was more active than L **2i**. According to the other results, EC₅₀ values for L enantiomers are lower or very close to their D counterparts. Figure 8 shows the comparison of antiproliferative activity of compounds **2b** and **2c** to MPA **1** against activated lymphocytes. According to these data, MPA **1** was more active than threonine esters **2b,c** significantly in the range of concentrations 0.000001 - 0.01 mg/mL. Histograms (Figure 9) depict cell divisions containing VPD450 dye. Shifting the median fluorescence peak to the left (increasing number of cell divisions) correlates with decreasing concentration of compounds. The blue line is assigned to the fluorescence peak of cells stained with VPD450 in the presence



of the highest concentration of compound. The peak moves to the left together with decreasing compound concentrations, which indicates more intensive cell proliferation.

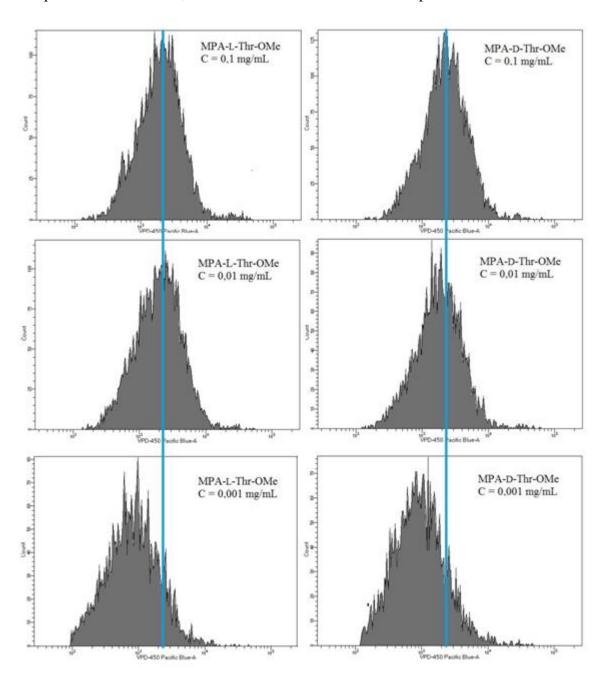


Figure 9. Histograms showing the division of cells containing dye VPD450 against PBMC in the case of tyrosine analogs **2b** and **2c**.



Table 3. EC_{50} [μ M] values of amino acid MPA derivatives 2a-m based on proliferative test with VPD450.

		T-JURKAT			РВМС			
Compound	No	EC ₅₀ [μM]	р	F	EC ₅₀ [μM]	р	F	
MPA	1	0.3 ± 0.2			0.0301 ± 0.0003			
MPA-L-Asp(OMe)-OMe	2 a	16.7 ± 0.23	0.738	0.190	8 ± 2.1	0.121	27.0	
MPA-L-Thr-OMe	2b	10 ± 7.1	0.293	4.05	10 ± 7.8	0.091	48.0	
MPA-D-Thr-OMe	2c	14 ± 5.4	0.379	2.18	9 ± 4.8	0.099	41.0	
MPA-L-Ile-OMe	2d	9 ± 5.1	0.105	36.4	5.12 ± NAN	0.667	0.33	
MPA-L-Arg-OMe	2e	22.3 ± 0.51	< 0.05	3114	77 ± 2.9	< 0.05	162	
MPA-D-Arg-OMe	2f	69 ± 40.9	< 0.05	723	95.57 ± NAN	< 0.05	492	
MPA-Mal-(OMe) ₂	2g	65 ± 7.8	0.091	48.1	11 ± 5.5	0.124	25.7	
MPA-L-Asp(OH)-OH	2h	396 ± 66.7	< 0.05	1366	211.0166 ± 3.6952	< 0.05	1310	
MPA-L-Thr-OH	2i	83 ± 70.5	< 0.05	834	118.63 ± NAN	< 0.05	17638	
MPA-D-Thr-OH	2j	21.41 ± NAN	< 0.05	672	121.51 ± NAN	< 0.05	5007	
MPA-L-Ile-OH	2k	2.3 ± 0.52	0.182	11.5	9 ± 5.4	0.118	28.3	
MPA-L-Arg-OH	21	1190 ± 474.8	< 0.05	6110	132 ± 2.8	< 0.05	1729	
MPA-D-Arg-OH	2m	1037 ± 16.2	< 0.05	3616	588 ± 11.2	< 0.05	7042	

p – statistical significance of the difference, F – Fisher test to MPA.



2.2.2.2. Peptide derivatives of MPA

Table 4 presents the results of anti-proliferative activity of peptide derivatives of MPA 3a-n. For some compounds, in the case of the T-Jurkat cell line, it was unable to determine the EC₅₀ value. The most active compounds were **3h** (EC₅₀ = 663.93 μ M) and **3l** (EC₅₀ = 556 μ M). However, even these analogs showed over 2000 times lower activity than MPA 1.

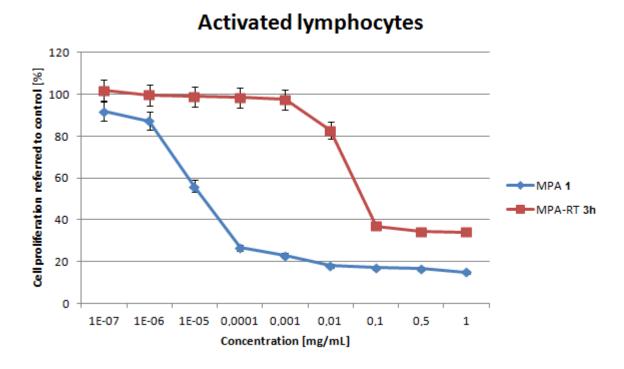


Figure 10. Comparison of antiproliferative activity of compound 3h – peptide derivative with best SI (see Table 5) do MPA 1 against PBMC.



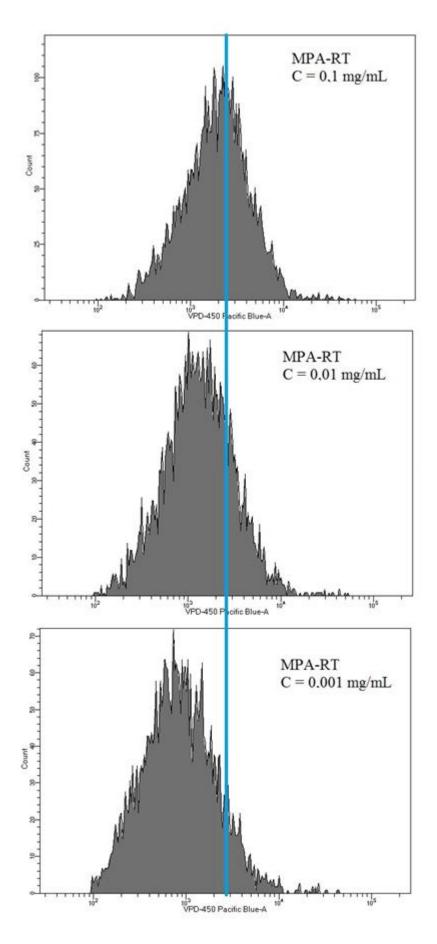


Figure 11. Histograms showing division of cells containing dye VPD450 against PBMC for 3h.

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Table 4. EC_{50} [μM] values of peptide MPA derivatives 3a-n based on proliferative test with VPD 450.

		T-JURKAT				PBMC		
Compound	No	EC ₅₀ [μM]	Viability [%]*	р	F	EC ₅₀ [μM]	р	F
MPA	1	0.3 ± 0.2				0.0301 ± 0.0003		
MPA-T	3 a		60.78%			130 ± 68.4	< 0.05	1204
MPA-T-Gly	3b	1030 ± 50.9		< 0.05	19448	590 ± 266.1	< 0.05	30012
MPA-T-αAla	3c		55.31%			259 ± 54.9	< 0.05	2129
MPA-T-βAla	3d	941 ± 367.5		< 0.05	8494	211 ± 98.2	< 0.05	23935
MPA-T-Val	3 e		55.10%			432 ± 108.4	< 0.05	33685
MPA-T-Leu	3f	801 ± 54.3		< 0.05	13790	551 ± 3.2	< 0.05	30012
MPA-T-Ile	3g	924 ± 175.2		< 0.05	12455	357 ± 107.1	< 0.05	19110
MPA-RT	3h	663.93 ± NAN		< 0.05	7500	35.43 ± NAN	< 0.05	29410
MPA-RT-Gly	3i		56.72%			565.92 ± NAN	< 0.05	37509
MPA-RT-αAla	3j	839 ± 85.5		< 0.05	17956	168 ± 75.3	< 0.05	19776
MPA-RT-βAla	3k		75.89%			154 ± 92.9	< 0.05	34031
MPA-RT-Val	31	556 ± 113		< 0.05	11781	129 ± 77.4	< 0.05	15052
MPA-RT-Leu	3m		63.04%			161 ± 62.4	< 0.05	17072
MPA-RT-Ile	3n		52.77%			152.23 ± NAN	< 0.05	19219

p – statistical significance of the difference, F – Fisher test to MPA.

^{*} cell viability calculated at highest tested concentration 1mg/mL

Analyzing the results of tests carried out on PBMCs, the most potent was derivative 3h, which gave $EC_{50} = 35.43 \mu M$. However, it is more than 1000 times lower than for MPA 1 $(EC_{50} = 0.0301 \mu M)$. MPA conjugates from retro-tuftsin **3h-n** indicated higher activity than MPA derivatives with tuftsin **3a-g**, which is clearly visible against PBMCs. Figure 10 shows a comparison of the antiproliferative activity of the MPA conjugate with retro-tuftsin 3h to mycophenolic acid 1 against PBMCs, and conjugate 3h increased clearly its activity from concentration of 0.001 g/mL. Figure 11 presents the number of cell divisions depending on the concentration of 3h. Again, the blue line indicates the fluorescence peak of cells stained with VPD450 in the presence of the highest concentration of compound. In this case, as the concentration of compound decreases, it shifts to the left, which indicates more intensive cell proliferation.

2.2.3. Selection of MPA derivatives with the most favorable antiproliferative properties

Determination of the selectivity index (SI), which is the ratio of the cytotoxicity of the compound (IC₅₀) to its antiproliferative activity (EC₅₀) enables to compare the obtained derivatives with one another and to the parent MPA 1 as well as. On this basis, we selected analogue with the best activity and the lowest toxicity. With the increase of SI, the toxicity of the compound decreases.

Table 5 shows the selectivity coefficients SI for the obtained amino acid and peptide analogs of mycophenolic acid. In the case of the T-Jurkat cell line, none of MPA derivatives was characterized by a higher selectivity index than MPA 1. In contrast, when analyzing the results of studies carried out on PBMCs, three compounds: 2b,c and 3h indicated significant activities and were selected for further studies as potential immunosuppressants. Although the EC₅₀ value of these analogues is higher than in the case of MPA 1 (Table 4), they were characterized by much lower cytotoxicity if compared to mycophenolic acid 1. The peptide derivatives 3j,l had a similar selectivity ratio as the MPA 1, but a high dose of substance would be needed to inhibit cell proliferation (168 μ M for 3j and 129 μ M for 3l).



Table 5. Selectivity index (SI) against Jurkat and PBMC cell lines.

Compound	No	$SI = IC_{50}/EC_{50}$		Compound	No	$SI = IC_{50}/EC_{50}$	
		JURKAT	PBMC			JURKAT	PBMC
MPA	1	228	4.6	MPA	1	228	4.6
MPA-L-Asp(OMe)-OMe	2 a	1.2	2.2	MPA-T	3a	-	-
MPA-L-Thr-OMe	2b	1.8	5.7	MPA-T-Gly	3b	-	1.1
MPA-D-Thr-OMe	2 c	8.3	7.3	MPA-T-αAla	3c	-	2.2
MPA-L-Ile-OMe	2d	0.9	0.4	MPA-T-βAla	3d	1.1	2.5
MPA-L-Arg-OMe	2 e	16	0.2	MPA-T-Val	3e	-	
MPA-D-Arg-OMe	2f	8.2	0.7	MPA-T-Leu	3f	-	1.2
MPA-Mal-(OMe) ₂	2g	1.6	0.3	MPA-T-Ile	3g	0.9	1.3
MPA-L-Asp(OH)-OH	2h	1.4	0.2	MPA-RT	3h	1.5	18
MPA-L-Thr-OH	2i	0.5	0.7	MPA-RT-Gly	3i	-	1.5
MPA-D-Thr-OH	2j	11	1	MPA-RT-aAla	3ј	-	4.5
MPA-L-Ile-OH	2k	33	2.3	MPA-RT-bAla	3k	-	3.8
MPA-L-Arg-OH	21	-	2.2	MPA-RT-Val	31	-	4.3
MPA-D-Arg-OH	2m	-	0.7	MPA-RT-Leu	3m	-	3.2
				MPA-RT-Ile	3n	-	3.5

SI higher than for MPA

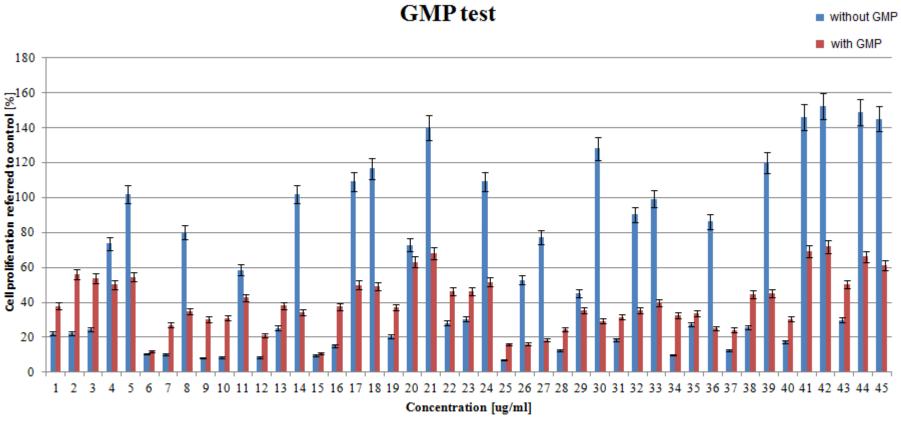


2.2.4. Determination of antiproliferative activity of compounds **2a-m**, **3d**, **e**, **3h**, **j**, **l** on PBMCs using the VPD450 dye in the presence of GMP

The inosine 5'-monophosphate dehydrogenase enzyme, inhibits nucleotide biosynthesis *de novo*. Data in the literature indicate that suppression of cell proliferation under the influence of IMPDH inhibitors is reversible by the addition of guanosine, GMP, GTP or deoxy-GMP [6]. Therefore, antiproliferative tests with the addition of GMP was performed to check whether the antiproliferative activity of MPA derivatives was based on the same mechanism of action as mycophenolic acid **1**. For this purpose, we carried out an antiproliferation test using the VPD450 dye with the addition of 50 µM GMP (concentration of GMP in the well).

Figures 12, 13 present the proliferative activity of cells relative to controls at respective concentrations of examined compounds. The mechanism of action of MPA derivatives was consistent with the action of MPA 1. In the range of concentrations in which inhibition of proliferation occurs without the addition of GMP, it can be seen that after the addition of GMP, proliferation increases definitely. In contrast, at concentrations in which proliferation without GMP was higher than about 50%, it decreased rapidly after GMP addition. In other words, the reversal of proliferation was observed again. At lower concentrations, in which the compounds did not significantly inhibit lymphocyte proliferation, the IMPDH enzyme was likely inhibited by GMP (product), which might be due to substrate-product imbalance caused by an excess of guanosine-5'-monophosphate. At these concentrations, it was also possible the toxic effects of GMP known from the netosis or its effects on other metabolic pathways (as a third transmitter or substrate for other enzymes).

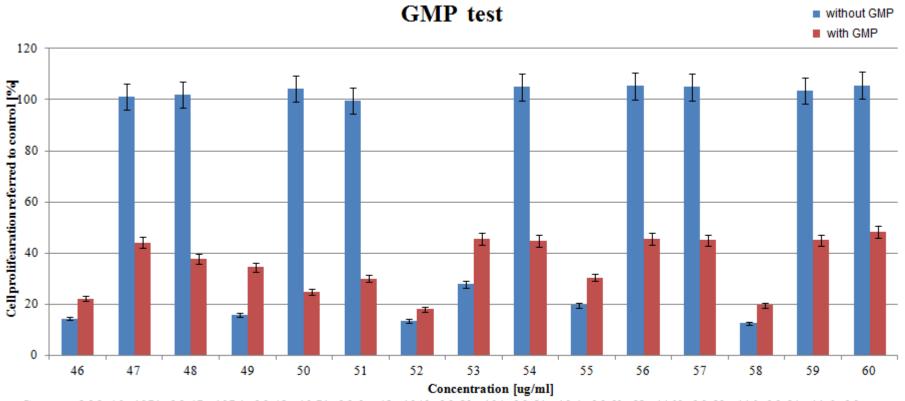




Compound 1: $1 - 312.2 \,\mu\text{M}$, $2 - 31.22 \,\mu\text{M}$, $3 - 3.122 \,\mu\text{M}$, $4 - 0.3122 \,\mu\text{M}$, $5 - 0.03122 \,\mu\text{M}$, $2\mathbf{a}$: $6 - 215.8 \,\mu\text{M}$, $7 - 21..8 \,\mu\text{M}$, $8 - 2.158 \,\mu\text{M}$, $2\mathbf{b}$: $9 - 229.6 \,\mu\text{M}$, $10 - 22.96 \,\mu\text{M}$, $11 - 2.296 \,\mu\text{M}$, $2\mathbf{c}$: $12 - 229.6 \,\mu\text{M}$, $13 - 22.96 \,\mu\text{M}$, $14 - 2.296 \,\mu\text{M}$, $2\mathbf{d}$: $15 - 223.4 \,\mu\text{M}$, $16 - 22.34 \,\mu\text{M}$, $17 - 2.234 \,\mu\text{M}$, $18 - 0.2234 \,\mu\text{M}$, $2\mathbf{e}$: $19 - 186.7 \,\mu\text{M}$, $20 - 18.67 \,\mu\text{M}$, $21 - 1.867 \,\mu\text{M}$, $2\mathbf{f}$: $22 - 1867 \,\mu\text{M}$, $23 - 186.7 \,\mu\text{M}$, $24 - 18.67 \,\mu\text{M}$, $2\mathbf{g}$: $25 - 222.5 \,\mu\text{M}$, $26 - 22.25 \,\mu\text{M}$, $27 - 2.225 \,\mu\text{M}$, $28 - 229.7 \,\mu\text{M}$, $29 - 22.97 \,\mu\text{M}$, $30 - 2.297 \,\mu\text{M}$, $2\mathbf{i}$: $31 - 237.3 \,\mu\text{M}$, $32 - 23.73 \,\mu\text{M}$, $33 - 2.373 \,\mu\text{M}$, $2\mathbf{j}$: $34 - 237.3 \,\mu\text{M}$, $35 - 23.73 \,\mu\text{M}$, $36 - 2.373 \,\mu\text{M}$, $2\mathbf{k}$: $37 - 230.7 \,\mu\text{M}$, $38 - 23.07 \,\mu\text{M}$, $39 - 2.307 \,\mu\text{M}$, $21 - 1917 \,\mu\text{M}$, $41 - 191.7 \,\mu\text{M}$, $42 - 191.7 \,\mu\text{M}$, $44 - 191.7 \,\mu\text{M}$, $45 - 191.7 \,\mu\text{M}$.

Figure 12. Antiproliferative activity of amino acid derivatives of MPA 2a-m in the presence of GMP.





Compound 3d: $46 - 1071 \,\mu\text{M}$, $47 - 107.1 \,\mu\text{M}$, $48 - 10.71 \,\mu\text{M}$, 3e: $49 - 1040 \,\mu\text{M}$, $50 - 104 \,\mu\text{M}$, $51 - 10.4 \,\mu\text{M}$, 3h: $52 - 1160 \,\mu\text{M}$, $53 - 116 \,\mu\text{M}$, $54 - 11.6 \,\mu\text{M}$, 3j: $55 - 1071 \,\mu\text{M}$, $56 - 107.1 \,\mu\text{M}$, $57 - 10,71 \,\mu\text{M}$, 3l: $58 - 1040 \,\mu\text{M}$, $59 - 104 \,\mu\text{M}$, $60 - 10.4 \,\mu\text{M}$.

Figure 13. Antiproliferative activity of peptide derivatives of MPA 3d, e, h, j, l in the presence of GMP.



2.2.5. Determination of stability of selected compounds against Jurkat cells using HPLC-MS

In order to check the stability of the obtained MPA analogs against cell cultures, an HPLC-MS analysis was carried out to determine the amount of compounds within incubation with Jurkat cells. Compounds 2a, i, 3c, h at EC₅₀ concentrations were incubated with Jurkat cells for 5 days, a sample was taken each day for HPLC-MS.

Stability of compounds 2a, 2i, 3c, 3h against Jurkat cell line Concentration referred to control [%] 100 80 MPA-L-Thr-OMe 2a MPA-L-Thr-OH 2i 40 MPA-T-aAla 3c MPA-RT 3h 20 0 1 2 3 5 Days

Figure 14. The rate of consumption of compounds 2a,i, 3c,h against Jurkat during a five-days incubation.

Figure 14 presents the stability of compounds 2a, i, 3c, h against Jurkat cells during a five-day incubation. The area of the peak of the first day was marked as 100%. Analyzing the data in the graph can be seen, that the derivative possessing threonine 3i with a free carboxyl group was characterized by considerable stability, the amount of compound during

the five-day incubation did not fall below 90%. The MPA analogue modified with threonine with a carboxyl group protected with methyl ester 2b also showed good stability towards the tested cell line, however on the fifth day of incubation, the amount of derivative 2b was about 40%. Therefore, addition of appropriate portion of the analogue **2b** could be advantageous for incubation for longer than 5 days. The peptide derivatives of MPA turned out to be much less stable, which can be explained by their extensive structure. After the third day of incubation, slightly more than 50% of the 3c and 3h analogs remained in the well. Thus, addition of another portion of compounds 3c and 3h during incubation over 4 days would be necessary.

3. Summary

In the article we presented the synthesis and immunosuppressive studies of amino acid and peptide analogs of mycophenolic acid. The most favorable conditions for the synthesis included



EDCI / DMAP and T3P / TEA procedures. We performed cytotoxicity (MTT) and antiproliferation tests (VPD 450, flow cytometry) to select compounds possessing improved parameters to MPA 1. The tests on the T-Jurkat lymphoidal cell line and peripheral blood mononuclear cells (PBMCs) were carried out. The obtained compounds fulfilled the assumption and proved to be less toxic than mycophenolic acid (against PBMC), whereas higher toxicity showed only four amino acids derivatives 2a, 2b, 2d and 2i (T-Jurkat). The highest cytotoxicity against both types of cells indicated MPA-L-Ile-OMe 2d, and the all peptide analogs of MPA were much less toxic than mycophenolic acid.

None of the designed compounds showed such a good antiproliferative effect as MPA 1. The most active occurred to be derivatives 2a-d and 2k (T-Jurkat and PBMC). The MPA derivative with isoleucine both in the form of methyl ester 2d (EC₅₀ = 5.12 μ M for PBMC) and deprotected 2k (EC₅₀ = 9 μ M for PBMC) had the lowest EC₅₀ value making it the most active MPA analog. The lowest activity against both types of cells was found in the case of 2h, 2l and 2m. The highest activity among peptide derivatives indicated compound 3h, whose EC₅₀ value was 663.93 μ M against T-Jurkat cell line and 35.43 μ M against PBMCs.

The cytotoxicity and antiproliferative activity of amide MPA derivatives depended both on the presence of methyl ester and the absolute amino acid configuration. The MPA analogs protected with the methyl ester were more active than their counterparts with the free carboxyl group, which can be due to better cell membrane penetration [11]. In addition, L-enantiomers occurred to be more active than the enantiomers with the D-configuration. Most peptide derivatives negatively affected the action of MPA, but MPA conjugates with retro-tuftsin exhibited slightly higher activity than those with tuftsin.

According to the SI selectivity coefficient, we chose three analogs with a higher SI value than MPA (SI = 4.6). These compounds include MPA derivatives with threonine of L and D configuration in the form of methyl esters, 2b and 2c (SI = 5.7 for 2b and 7.3 for 2c) and MPA conjugate with retro-tuftsin 3h (SI = 18). The two peptide derivatives are characterized by a selectivity coefficient similar to mycophenolic acid 1, they are conjugates of MPA modified with α -alanine 3j and valine 3l. Despite the lower antiproliferative activity of selected compounds, they were less toxic. Therefore, it would be necessary to use a higher dose of the drug to achieve the same antiproliferative effect as in case of MPA 1.

Proliferation studies with the addition of GMP, performed on the obtained MPA derivatives, indicated inhibitory activity against IMPDH similarly to mycophenolic acid.



Acknowledgments

This work was financially supported by the Gdansk University of Technology under grant DS/031946.

4. Experimental section

All reactions were performed in inert atmosphere with magnetic stirring. DMF was purified by distillation from benzene/water. The reactions were monitored by TLC on Merck F254 silica gel precoated plates. The following solvent systems (by volume) were used for TLC development: (A), CHCl₃:MeOH:NH₃ (5:1:0.2, v/v/v) (B). A: CHCl₃:MeOH (9:1, v/v), B: CHCl₃:MeOH:NH₃ (9:1:0.2, v/v/v), C; CHCl₃:MeOH:NH₃ (3:1:0.2, v/v/v).

¹H NMR and ¹³C NMR spectra were taken on the spectrometer Varian Unity 500 Plus in CD₃OD or DMSO. Mass spectra were performed at the Laboratory of Mass Spectrometry MALDI-TOF on the matrix DHB (BIFLEX III Bruker).

Conditions of chromatographic HPLC separation and detection of examined compounds: column - Poroshell ECC18 (3.0 × 150 mm), 2.7 mm, Agilent Technologies; column temperature -40°C; injection volume - 2 mL; flow rate - 0.4 mL/min; eluents: (A) 0.1% HCOOH in water, (B) 0.1% HCOOH in ACN/MeOH (1:1, v/v); gradient program:

Gradient program for HPLC

Time [min]	% A	% B
0	90	10
20	0	100
30	0	100

Post time – 10 min; UV-Vis detection; wavelenghts UV: 254 nm; Vis: 580 nm; peak width > 0.1 min (2s); ESI MS detection

4.1. General procedure for the preparation of MPA amino acid analogues 2a-m

The amino acid analogues of the MPA, were obtained according to general procedure for preparation amino acid derivatives described in our last work [20,25].

4.2. Preparation of tuftsin/retro-tuftsin derivatives 3a-n

The procedure for the synthesis of compounds 3a and 3h by the mixed anhydride method has been published [22-25]. The Fmoc-protecting groups was removed by treatment with diethylamine in chloroform.



4.2.1. Conjugates of MPA with tuftsin derivatives **3a-h**.

4.2.1.1 Coupling of MPA 1 with peptides 5a-g

Derivative TFA×Thr-Lys(Fmoc)-Pro-Arg(NO₂)-OCH₃ **5a** 0.025 g (0.0289 mmol), MPA 0.0068 g (0.0212 mmol), TEA 0.0014 ml (0.0112 mmol) were dissolved in 1 ml anhydrous DMF, under nitrogen. Then, the reaction mixture was coolded to 0°C, followed by addition of T3P 0.030 ml (0.052 mmol, 50 % solution in DMF) and stirred for 2 h. Subsequently, stirring was continued for 48 h at room temperature. The progress of the reaction was monitored with TLC (solvent system A). When the reaction was complete, the solvent was distilled of under vacuum, and the product isolated with preparative thin layer chromatography (PTLC), solvent system A.

4.2.1.1.1. Compound MPA-Thr-Lys(Fmoc)-Pro-Arg(NO₂)-OMe **6a**:

Product 6a was obtained with yield 65% as white powder.

MS m/z calculated for C₅₄H₆₉N₉O₁₅ 1084.1; found 1084.5 [M+H]⁺

 $R_f = 0.512$ (solvent system A).

4.2.1.1.2. Compound MPA-Thr-Lys(FmocGly)- $Pro-Arg(NO_2)$ - $OMe\ {\it 6b}$:

Product **6b** was obtained with yield 48 % as white powder.

MS m/z calculated for C₅₆H₇₂N₁₀O₁₆ 1141.2; found 1141.5 M⁺

Rf = 0.589 (solvent system A).

4.2.1.1.3. Compound MPA-Thr-Lys($Fmoc\alpha Ala$)- $Pro-Arg(NO_2)$ -OMe **6c**:

Product **6c** was obtained with yield 53 % as white powder.

MS m/z calculated for $C_{58}H_{76}N_{10}O_{16}$ 1155.2; found: 1155.5 M^+

Rf = 0.542 (solvent system A).

4.2.1.1.4. Compound MPA-Thr-Lys($Fmoc\beta Ala$)-Pro-Arg(NO_2)-OMe **6d**:

Product **6d** was obtained with yield 51 % as white powder.

MS m/z calculated for C₅₈H₇₆N₁₀O₁₆ 1155.2; found: 1155.5 M⁺

Rf = 0.502 (solvent system A).

4.2.1.1.5. Compound MPA-Thr-Lys(FmocVal)-Pro-Arg(NO₂)-OMe **6e**:

Product **6e** was obtained with yield 49 % as white powder.

MS m/z calculated for $C_{59}H_{78}N_{10}O_{16}$ 1183.3; found 1183.6 M^+

Rf = 0.583 (solvent system A).



4.2.1.1.6. Compund MPA-Thr-Lys(FmocLeu)-Pro-Arg(NO₂)-OMe **6**f:

Product 6f was obtained with yield 42 % as white powder.

MS m/z calculated for $C_{60}H_{80}N_{10}O_{16}$ 1197.3; found 1197.6 M^+

Rf = 0.602 (solvent system A).

4.2.1.1.7. Compound MPA-Thr-Lys(FmocIle)-Pro-Arg(NO₂)-OMe **6g**:

Product 6g was obtained with yield 45 % as white powder.

MS m/z calculated for $C_{60}H_{80}N_{10}O_{16}$ 1197.3; found 1197.6 M^+

Rf = 0.571 (solvent system A).

4.2.1.2. Deprotection of MPA conjugates with tuftsin to 3a-g

Derivative MPA-Thr-Lys(Fmoc)-Pro-Arg(NO₂)-OMe **6a** 0.05 g (0.046 mmol) was dissolved in 1 ml of chloroform and 0.2-0.3 ml of diethylamine was added. Then, the reaction mixture was stirred overnight and controlled with TLC (solvent system A). Subsequently, solvent was evaporated under vacuum, and the product isolated with preparative this layer chromatography (PTLC), solvent systems B and C. The conjugate **3a** was given in yield 84% as a white powder.

4.2.1.2.1. Compound MPA-Thr-Lys-Pro-Arg(NO_2)-OMe 3a:

Product 3a was obtained with yield 84 % as white powder.

MPA-T 3a: ¹HNMR (400 MHz, DMSO-d₆) δ ppm: 0.97 (d, J=6.3 Hz, 3H, γ -T4), 1.36 (m, 2H, γ -K4), 1.53 (m, 5H, δ -K5, β -K3b, γ -R4), 1.63 (m, 2H, β -K3a, β -R3a), 1.74 (s, 3H, f), 1.86 (m, 3H, β -P3b, γ -P4b, γ -P4a), 2.03 (m, 1H, β -P3a), 2.07 (s, 3H, e), 2.14 (m, 2H, g), 2.125 (m, 2H, h), 2.88 (m, 2H, ϵ -K6), 3.17 (m, 2H, δ -R5), 3.29 (d, J=6,8 Hz, 2H, d), 3.52 (m, 1H, δ -P5b), 3.62 (s, 3H, COOMe), 3.65 (m, 1H, δ -P5a), 3.69 (s, 3H, c), 3.92 (m, 1H, β -T3), 4.17 (m, 1H, α -T2), 4.21 (m, 1H, α -R2), 4.35 (m, 1H, α -P2), 4.50 (m, 1H, α -K2), 5.14 (t, J=6.4 Hz, 1H, a), 5.23 (s, 2H, b), 7.81 (m, 2H, α -KNH, α -TNH), 8.34 (d, J=8.34 Hz, 1H, α -RNH);

¹³CNMR (CD₃OD, 400 MHz,CD₃OD-d₄): 10.02 (e), 14.93 (f), 18.83 (γ -T4), 21.73 (γ -K4), 22.25 (d), 24.79 (γ -P4), 26.7 (γ -R4), 28.19 (β -R3), 29.11 (δ -K5), 30.28 (β -P3), 34.26 (h), 35.13 (g), 39.10 (ϵ -K6), 40.15 (δ -R5), 42.15 (δ -P5), 48.46 (α -R2), 50.83 (α -K2), 51.46 (OMe), 56.93 $(\alpha-T2)$, 58.87 $(\alpha-P2)$, 61.17 (c), 66.99 ($\beta-T3$), 69.39 (b), 106.36 (l), 116.21 (r), 122.31 (o), 123.11 (a), 133.70 (j), 145.27 (m), 153.82 (p), 163.43 (k), 170.83 (T1), 171.31 (K1), 172.12 (n), 172.47 (R1), 173.10 (P1), 174.75 (i);

MS m/z calculated for C₃₉H₅₉N₉O₁₃ 861.4; found 861.9 M⁺

Rf = 0.458 (solvent system B);

Purity HPLC 100%.



4.2.1.2.2. Compound MPA-Thr-Lys(Gly)-Pro-Arg(NO_2)-OMe 3b:

Product **3b** was obtained with yield 81 % as white powder.

MPA-T-Gly 3b: ¹HNMR (400 MHz, DMSO-d₆) δ ppm: 0.94 (d, J=6,3 Hz, 3H, γ -T4), 1.24 (m, 2H, γ -K4), 1.33 (m, 2H, δ -K5), 1.46 (m, 5H, β -K3b, γ -R4, β -K3a, β -K3a), 1.67 (m, 4H, f, β -R3a), 1.75 (m, 3H, β -P3b, γ -P4b, γ -P4a), 2.01 (m, 4H, β -P3a, e), 2.10 (m, 2H, g), 2.17 (m, 2H, hb, ha), 3.03 (m, 2H, ε -K6), 3.10 (m, 4H, δ -R5, α -G2), 3.23 (d, J=6.8 Hz, 2H, d), 3.48 (m, 1H, δ-P5b), 3.61 (s, 3H, COOMe), 3.63 (m, 1H, δ-P5a), 3.65 (s, 3H, c), 3.89 (m, 1H, β-T3), 4.14 $(m, 1H, \alpha-T2), 4.18 (m, 1H, \alpha-R2), 4.32 (m, 1H, \alpha-P2), 4.44 (m, 1H, \alpha-K2), 5.09 (t, J=6.5 Hz,$ 1H, a), 5.15 (s, 2H, b), 6.70 (m, 2H, α -KNH, α -TNH), 7.86 (t, J=5.5 Hz, 1H, ϵ -NHK), 8.22 (d, J=7.4 Hz, 1H, α -RNH);

¹³CNMR (CD₃OD, 400 MHz,CD₃OD-d₄): 11.48 (e), 16.51 (f), 20.19 (γ -T4), 22.52 (γ -K4), 22.95 (d), 24.90 (γ -P4/ γ -R4), 28.46 (β -R3), 29.17 (δ -K5), 39.45 (β -P3), 31.24 (β -K3), 34.58 (h), 35.50 (g), 38.54 (ϵ -K6), 40.63 (δ -P5), 43.71 (α -G2), 50.63 (α -R2), 52.02 (α -K2), 52.29 (COOMe), 58.44 (α -T2), 59.43 (α -P2), 60.99 (c), 67.03 (β -T3), 68.86 (b), 105.90 (l), 107.27(r), 123.01 (o), 123.23 (a), 134.18 (j), 146.11 (m), 159.78 (p), 162.98 (k), 170.29 (T1), 170.42 (K1), 170.91 (n), 171.93 (G1), 172.24 (P1), 172.56 (R1), 172.80 (i);

MS m/z calculated for C₄₁H₆₂N₁₀O₁₄ 919.0; found 919.4 M⁺

 $\mathbf{Rf} = 0.372$ (solvent system B)

Purity HPLC 100%.

4.2.1.2.3. Compound MPA-Thr-Lys(α Ala)-Pro-Arg(NO_2)-OMe 3c:

Product 3c was obtained with yield 79% as white powder.

MPA-T-αAla **3c**: 1 HNMR (400 MHz, DMSO-d₆) δ ppm: 0.97 (d, J=6.3 Hz, 3H, γ -T4), 1.18 (d, J=6.3 Hz, 3H, β-A3), 1.29 (m, 2H, γ -K4), 1.39 (m, 3H, δ-K5, β-K3b), 1.48 (m, 2H, γ -R4) 1.56 $(m, 1H, \beta-R3b), 1.62 (m, 1H, \beta-K3a), 1.68 (m, 1H, \beta-R3a), 1.74 (s, 3H, f), 1.79 (m, 2H, \beta-P3b, f)$ γ -P4b), 1.89 (m, 1H, γ -P4a), 2.03 (m, 1H, β -P3a), 2.07 (s, 3H, e), 2.15 (m, 2H, g), 2.27 (m, 2H, h), 3.05 (m, 2H, ϵ -K6), 3.17 (m, 2H, δ -R5), 3.28 (d, J=6.8 Hz, 2H, d), 3.05 (m, 1H, δ -P5b), 3.41 (dd, J=13.9 Hz, J=6.9 Hz, 1H, α -A2), 3.62 (s, 3H, COOMe), 3.65 (m, 1H, δ -P5a), 3.69 (s, 3H, c), 3.92 (m, 1H, β -T3), 4.17 (m, 1H, α -T2), 4.22 (m, 1H, α -R2), 4.35 (m, 1H, α -P2), 4.47 (m, 1H, α -K2), 5.15 (t, J=6.5 Hz, 1H, a), 5.21 (s, 2H, b), 7.74 (m, 2H, α -KNH, α -TNH), 7.97 (m, 1H, ε -KNH), 8.30 (d, J=7.4 Hz, 1H, α -RNH);

¹³CNMR (CD₃OD, 400 MHz,CD₃OD-d₄) δ ppm: 10.05 (e), 14.98 (f), 18.86 (γ -T4), 18.92 $(\beta-A3)$, 22.17 (γ-K4), 22.29 (d), 24.81 (γ-P4), 28.22 (γ-R4), 28.50 (β-R3), 29.04 (δ-K5), 30.49 (β-P3), 31.28 (β-K3), 34.32 (g), 35.15 (h), 38.55 (ε-K6), 40.15 (δ-R5), 46.89 (δ-P5), 48.48



 $(\alpha-R2)$, 49.81 $(\alpha-A2)$, 51.1 $(\alpha-K2)$, 51.48 (OMe), 58.67 $(\alpha-T2)$, 59.76 $(\alpha-P2)$, 61.17 (c), 69.03 (β-T3), 69.33 (b), 106.37 (l), 115.91 (r), 122.40 (o), 123.17 (a), 133.65 (j), 145.26 (m), 154.20 (p), 163.40 (k), 171.12 (T1), 171.30 (K1, A1), 172.07 (n), 172.51 (R1), 173.01 (P1), 174.67 (i); MS m/z calculated for C₄₂H₆₄N₁₀O₁₄ 933.0; found 933.4 M⁺

Purity HPLC 100%

 $R_f = 0.442$ (solvent system B).

4.2.1.2.4. Compound MPA-Thr-Lys(β Ala)-Pro-Arg(NO_2)-OMe 3d:

Product 3d was obtained with field 78% as white powder.

MPA-T-βAla **3d**: ¹HNMR (400 MHz, DMSO-d₆) δ ppm: 0.97 (d, J=6.3 Hz, 3H, γ -T4), 1.33 (m, 2H, γ -K4), 1.38 (m, 3H, δ -K5, β -K3b), 1.47 (m, 2H, γ -R4) 1.55 (m, 1H, β -R3b), 1.63 $(m, 1H, \beta-K3a), 1.71 (m, 1H, \beta-R3a), 1.74 (s, 3H, f), 1.78 (m, 1H, \beta-P3b), 1.88 (m, 2H, \gamma-P4b, f)$ γ -P4a), 2.04 (m, 1H, β -P3a), 2.08 (s, 3H, e), 2.16 (m, 2H, g), 2.27 (m, 2H, h), 2.46 (m, 2H, α -A2), 2.89 (q, J=7.3 Hz, 2H, β -A3), 3.02 (m, 2H, ϵ -K6), 3.15 (m, 2H, δ -R5), 3.30 (d, J=7.0 Hz, 2H, d), 3.5 (m, 1H, δ -P5b), 3.65 (s, 3H, COOMe), 3.65 (m, 1H, δ -P5a), 3.70 (s, 3H, c), 3.91 (m, 1H, β -T3), 4.18 (m, 2H, α -T2, α -R2), 4.35 (m, 1H, α -P2), 4.47 (m, 1H, α -K2), 5.13 (t, J=6.4 Hz, 1H, a), 5.25 (s, 2H, b), 7.79 (m, 2H, α -KNH, α -TNH), 7.81 (m, 1H, ϵ -KNH), 8.33 (d, J=7.2 Hz, 1H, α -RNH);

 13 CNMR CD₃OD, 400 MHz,CD₃OD-d₄) δ ppm: 10.07 (e), 14.95 (f), 18.82 (γ-T4), 22.18 $(\gamma-K4)$, 22.24 (d), 24.81 $(\gamma-P4)$, 28.21 $(\gamma-R4)$, 28.45 $(\beta-R3)$, 29.03 $(\delta-K5)$, 30.46 $(\beta-P3)$, 31.39 $(\beta-K3)$, 34.33 (h), 35.16 (g), 35.88 (α-A2), 38.54 (β-A3), 40.18 (ε-K6), 42.15 (δ-R5), 46.92 $(\delta-P5)$, 48.48 (α -R2, α -K2), 51.49 (OMe), 58.63 (α -T2), 60.08 (α -P2), 60.22 (c), 67.10 (β -T3), 69.47 (b), 106.33 (l), 116.53 (r), 122.24 (o), 123.01 (a), 133.82 (j), 145.28 (m), 153.28 (p), 163.44 (k), 170.77 (T1), 171.13 (K1), 171.25 (A1), 172.15 (n), 172.41 (R1), 173.04 (P1), 174.67 (i);

MS m/z calculated for C₄₂H₆₄N₁₀O₁₄ 933.0; found 933.5 M⁺

Rf = 0.370 (solvent system B)

Purity HPLC 94.8 %.

4.2.1.2.5. Compound MPA -Thr-Lys(Val)-Pro-Arg(NO₂)-OMe 3e:

Product 3e was obtained with yield 82% as white powder.

MPA-T-Val **3e**: 1 HNMR (400 MHz, DMSO-d₆) δ ppm: 0.81 (d, J=6.8 Hz, 3H, γ -V4), 0.86 (d, J=6.9 Hz, 3H, γ -V4), 0.96 (d, J=6.3 Hz, 3H, γ -T4), 1.22 (m, 5H, γ -K4, δ -K5, β -K3b), 1.48 $(m, 5H, \gamma-R4, \beta-K3a, \beta-R3a, \beta-R3b), 1.68 (s, 3H, f), 1.77 (m, 4H, \beta-P3b, \beta-V2, \gamma-P4b, \gamma-P4a),$ 2.01 (m, 4H, β -P3a, e), 2.10 (m, 4H, g,h), 2.99 (m, 5H, ϵ -K6, δ -R5, α -V2), 3.25 (d, J=6.9 Hz,



2H, d), 3.48 (m, 1H, δ -P5b), 3.59 (s, 3H, COOMe), 3.64 (m, 1H, δ -P5a), 3.66 (s, 3H, c), 3.88 $(m, 1H, \beta-T3), 4.15 (m, 2H, \alpha-T2, \alpha-R2), 4.33 (m, 1H, \alpha-P2), 4.44 (m, 1H, \alpha-K2), 5.10$ $(t, J=5.14 \text{ Hz}, 1H, a), 5.20 (s, 2H, b), 7.72 (m, 2H, \alpha-KNH, \alpha-TNH), 7.98 (m, 1H, \epsilon-NHK), 8.29$ (d, J=8.3 Hz, 1H, α -RNH);

¹³CNMR (CD₃OD, 400 MHz,CD₃OD-d₄): 10.03 (e), 14.92 (f), 16.70 (γ -V4), 18.02 (γ -T5), 18.83 (γ -T4), 22.25 (d), 24.81 (γ -R4), 24.87 (γ -P4), 28.22 (β -R3), 28.52 (δ -R5), 29.03 (β -P3), 30.48 (β -K3), 31.31 (β -V3), 34.34 (δ -K5), 35.17 (β -P3), 38.55 (β -R3), 40.14 (β), 51.11 $(\alpha-K2)$, 51.46 (COOMe), 58.67 $(\alpha-T2)$, 59.76 $(\alpha-P2)$, 60.18 (c), 67.04 $(\beta-T3)$, 69.40 (b), 106.34 (l), 116.34 (r), 122.29 (o), 123.07 (a), 133.74 (j), 145.25 (m), 159.46 (p), 163.43 (k), 171.12 (K1), 171.30 (T1), 172.05 (n/V1), 172.44 (P1), 173.02 (R1), 174.66 (i);

MS m/z calculated for C₄₃H₆₆N₁₀O₁₄ 947.0; found 947.5 M⁺

Rf = 0.58 (solvent system B)

Purity HPLC 99.3%.

4.2.1.2.6. Compound MPA-Thr-Lys(Leu)-Pro-Arg(NO₂)-OMe **3f**:

Product **3f** was obtained with yield 83% as white powder.

MPA-T-Leu **3f**: ¹HNMR (400 MHz, DMSO-d₆) δ ppm: 0.81 (m, 6H, δ-L5, ε-L6), 0.96 (d, J=6.3) Hz, 3H, γ -T4), 1.24 (m, 7H, γ -K4, β -L2, δ -K5, β -K3b, β -R3b), 1.53 (m, 4H, γ -R4, β -K3a, β -R3a,), 1.74 (s, 3H, f), 1.77 (m, 4H, β -P3b, γ -L4, γ -P4b, γ -P4a), 2.05 (m, 4H, β -P3a, e), 2.13 $(m, 4H, g,h), 3.01 (m, 4H, \epsilon-K6, \delta-R5), 3.28 (d, J=6.8 Hz, 2H, d), 3.44 (m, 1H, \delta-P5b), 3.62$ (m, 4H, COOMe, δ -P5a), 3.69 (s, 3H, c), 3.89 (m, 1H, β -T3), 4.16 (m, 2H, α -T2, α -R2), 4.33 (m, 1H, α -P2), 4.44 (m, 1H, α -K2), 5.12 (t, J=6.5 Hz, 1H, a), 5.24 (s, 2H, b), 7.73 (m, 2H, α -KNH, α -TNH), 8.13 (m, 1H, ϵ -NHK), 8.29 (d, J=8.3 Hz, 1H, α -RNH);

 13 CNMR (CD₃OD, 400 MHz,CD₃OD-d₄): 10.60 (e), 14.97 (f), 18.07 (γ -T4), 21.10 (d/ γ -K4), 21.68 (ϵ -L6) 22.26 (δ -L5), 24.34 (γ -L4), 24.82 (γ -R4/ γ -P4), 28.24 (β -R3), 28.48 (δ -R5), 29.03 (β-P3), 30.48 (β-K3), 34.35 (δ-K5), 35.17 (β-P3), 38.55 (β-R3), 40.16 (g), 42.35 (ε-K6), $48.47 (\alpha-R2)$, $51.11 (\alpha-K2)$, $51.48 (\alpha-L2)$, 52.63 (COOMe), $58.69 (\alpha-T2)$, 60.19 (c), 67.05(β-T3), 69.41 (b), 106.34 (l), 116.31 (r), 122.29 (o), 123.07 (a), 133.75 (j), 145.25 (m), 159.44 (p), 163.44 (k), 171.13 (K1), 171.31 (T1), 172.05 (n/L1), 172.44 (P1), 173.03 (R1), 174.65 (i); MS m/z calculated for C₄₅H₇₀N₁₀O₁₄ 975.1; found 975.3 M⁺

Rf = 0.548 (solvent system B)

Purity HPLC 98.7%.



4.2.1.2.7. Compound MPA-Thr-Lys(Ile)-Pro-Arg(NO_2)-OMe 3g:

Product **3g** was obtained with yield 77.7% as a white powder.

MPA-T-Ile **3g**: ¹HNMR (400 MHz, DMSO-d₆) δ ppm: 0.81 (m, 6H, δ -I5, ϵ -I6), 0.96 (d, J=6.3 Hz, 3H, γ -T4), 1.26 (m, 6H, γ -K4, β -I3, δ -K5, β -R3b), 1.51 (m, 5H, γ -R4, β -K3b, β -R3a, β -K3a), 1.70 (s, 4H, f, γ-I4b), 1.77 (m, 4H, β -P3b, γ-I4a, γ-P4b, γ-P4a), 2.00 (m, 1H, β -P3a), 2.08 (s, 3H, e), 2.15 (m, 4H, g, h), 2.99 (m, 4H, ε -K6, δ -R5), 3.22 (d, J=3.2 Hz, 1H, α -I2), 3.28 $(d, J=6.8 Hz, 2H, d), 3.48 (m, 1H, \delta-P5b), 3.62 (m, 3H, COOMe), 3.64 (m, 1H, \delta-P5a), 3.69$ (s, 3H, c), 3.90 (m, 1H, β -T3), 4.16 (m, 2H, α -T2, α -R2), 4.33 (m, 1H, α -P2), 4.44 (m, 1H, α-K2), 5.12 (t, J=6.4 Hz, 1H, a), 5.23 (s, 2H, b), 7.73 (m, 2H, α-KNH, α-TNH), 8.13 (m, 1H, ϵ -NHK), 8.29 (d, J=8.3 Hz, 1H, α -RNH);

 13 CNMR (CD₃OD, 400 MHz,CD₃OD-d₄): 11.52 (e), 11.81 (δ-I5), 15.68 (ε-I6), 16.52 (f), 20.15 $(\gamma-T4)$, 22.67 $(\gamma-K4)$, 22.90 (d), 24.48 $(\gamma-L4)$, 24.90 $(\gamma-P4)$, 28.44 $(\beta-R3)$, 29.15 $(\delta-P5)$, 29.47 $(\beta-P3)$, 31.29 $(\beta-K3)$, 34.56 $(\delta-K5)$, 35.52 $(\beta-P3)$, 37.77 (h), 38.66 (g), 47.24 $(\alpha-P2)$, 49.04 $(\alpha-12)$, 50.70 $(\alpha-K2)$, 52.61 (COOMe), 58.45 $(\alpha-P2)$ 59.42 $(\alpha-T2)$, 61.06 (c), 67.13 ($\beta-T3$), 69.02 (b), 107.35 (l), 115.86 (r), 122.96 (o), 123.05 (a), 134.40 (j), 146.20 (m), 159.88 (p), 163.03 (k), 170.31 (K1), 170.46 (T1), 170.76 (n/I1), 172.26 (P1), 172.49 (R1), 172.82 (i); MS m/z calculated for C₄₅H₇₀N₁₀O₁₄ 975.1 found 975.5 M⁺

Rf = 0.532 (solvent system B)

Purity HPLC 98.5%.

4.2.2. MPA conjugates with retro-tuftsin 3h-n

4.2.2.1. Coupling of MPA 1 with peptides 5h-n

Compound TFA×Arg(NO₂)-Pro-Lys(Fmoc)-Thr-OCH₃ **5h** 0.01 g (0.011 mmol), MPA 0.0026 g (0,008 mmola), DMAP 0.015 g (0.011 mmola) were dissolved in 1 ml of anydrous DMF under nitrogen. Then, the reaction mixture was cooled to 0 °C, followed by addition of EDCI 0.0017 g (0.0107 mmol), and after 2 h stirred for 48 h at room temperature. The progress of the reaction was monitorem with TLC (solvent system A). When the reaction was complete, solvent was distilled of under vacuum and the product isolated with preparative thin layer chromatography (PTLC), solvent system A.

4.2.2.1.1. Compound MPA-Arg(NO_2)-Pro-Lys(Fmoc)-Thr-OCH₃ **6h**:

Product **6h** was obtained with yield 64% as a white powder.

MS m/z calculated for C₅₄H₆₉N₉O₁₅ 1084.1; found 1084.5 M⁺

 $R_f = 0.489$ (solvent system A).



4.2.2.1.2 Compound MPA-Arg(NO₂)-Pro-Lys(FmocGly)-Thr-OCH₃ 6i:

Product 6i was obtained with yield 28% as white powder.

MS m/z was calculated for $C_{56}H_{72}N_{10}O_{16}$ 1141.2; found 1141.6 M^+

 $R_f = 0.474$ (solvent system A).

4.2.2.1.3. Compound MPA-Arg(NO₂)-Pro-Lys(FmocαAla)-Thr-OCH₃ **6j**:

Product 6j was obtained with yield 28% as white powder.

MS m/z calculated for C₅₈H₇₄N₁₀O₁₆ 1155.3; found 1155.5 M⁺

 $R_f = 0.480$ (solvent system B).

4.2.2.1.4. Compound MPA-Arg(NO₂)-Pro-Lys(Fmoc β Ala)-Thr-OCH₃ **6k**:

Product **6k** was obtained with yield 28% as white powder.

MS m/z calculated for C₅₈H₇₄N₁₀O₁₆ 1155.2; found 1155.6 M⁺

 $R_f = 0.420$ (solvent system B).

4.2.2.1.5. Compound MPA-Arg(NO₂)-Pro-Lys(FmocVal)-Thr-OCH₃ 61:

Product **61** was obtained with yield 28% as white powder.

MS m/z calculated for C₅₉H₇₈N₁₀O₁₆ 1183.3; found 1183.6 M⁺

Rf = 0.465 (solvent system A).

4.2.2.1.6. Compound MPA-Arg(NO₂)-Pro-Lys(FmocLeu)-Thr-OCH₃ 6m:

Product 6m was obtained with yield 28% as white powder.

MS m/z calculated for C₆₀H₈₀N₁₀O₁₆ 1197.3; found 1197.6 M⁺

 $\mathbf{Rf} = 0.512$ (solvent system A).

4.2.2.1.7. Compound MPA-Arg(NO₂)-Pro-Lys(FmocIle)-Thr-OCH₃ 6n:

Product **6n** was obtained with yield 28% as white powder.

MS m/z calculated for C₆₀H₈₀N₁₀O₁₆ 1197.3; found 1197.6 M⁺

Rf = 0.585 (solvent system A).

4.2.2.2. Deprotection of MPA conjugates with retro-tuftsin to 3h-n

Derivative MPA-Arg(NO₂)-Pro-Lys(Fmoc)-Thr-OMe 6h 0.05 g (0.046 mmol) was dissolved in 1 ml of chloroform and 0.2-0.3 ml of diethylamine was added. Then, the reaction mixture was stirred overnight and controlled with TLC (solvent system A). Subsequently, solvent was evaporated under vacuum, and the product isolated with preparative this layer chromatography (PTLC), solvent systems B and C. The conjugate **3h** was given in yield 86% as a white powder.



4.2.2.2.1. Compound MPA-Arg(NO_2)-Pro-Lys-Thr-OMe **3h**:

Product **3h** was obtained with yield 86% as white powder.

MPA-RT **3h**: ¹HNMR (400 MHz, DMSO-d₆) δ ppm: 1.057 (m, 3H, γ -T4), 1.348 (m, 2H, γ -K4), 1.487 (m, 3H, δ -K5, β -R3b), 1.532 (m, 3H, β -K3b, γ -R4), 1.640 (m, 2H, β -K3a, β -R3a), 1.728 (s, 3H, f), 1.810 (m, 2H, γ -P4b, β -P3b), 1.908 (m, 1H, γ -P4a), 1.992 (m, 1H, β -P3b), 2.083 (s, 3H, e), 2.132 (m, 2H, g), 2.171 (m, 2H, h), 2.735 (m, 2H, ϵ -K6), 3.135 (m, 2H, δ -R5), 3.283 (d, J=6.7 Hz, 2H, d), 3.548 (m, 1H, δ -P5b), 3.621 (s, 3H, COOMe), 3.649 (m, 1H, δ -P5a), 3.696 (s, 3H, c), 4.106 (m, 1H, β-T3), 4.237 (m, 1H, α-T2), 4.338 (m, 2H, α-K2, α-P2), 4.424 (m, 1H, α -R2), 5.099 (m, 1H, a), 5.253 (s, 2H, b), 7.972 (d, J=8.2 Hz, 1H, α -TNH), 8.045(m, 2H, ε -KNH), 8.114 (d, J=7.6 Hz, 1H, α -RNH), 8.162 (d, J=7.8 Hz, 1H, α -KNH);

¹³CNMR (CD₃OD, 400 MHz) δ ppm: 11.54 (e), 16.49 (f), 19.03 (γ-T4), 20.52 (d), 22.42 $(\gamma-K4)$, 22.87 $(\gamma-P4)$, 24.96 $(\delta-K5)$, 26.77 $(\gamma-R4)$, 28.65 $(\beta-R3)$, 29.55 $(\beta-P3)$, 31.57 $(\beta-K3)$, 34.27 (h), 35.46 (g), 38.97 (ϵ -K6), 41.67 (δ -R5), 47.33 (δ -P5), 52.30 (OMe), 52.50 (α -R2), $56.47 (\alpha-K2)$, $58.35 (\alpha-T2)$, $59.76 (\alpha-P2)$, 61.12 (c), $66.82 (\beta-T3)$, 69.10 (b), 107.44 (o), 116.45(r,l), 122.96 (a), 134.54 (j), 146.27 (m), 153.17 (p), 163.05 (k), 170.61 (n), 170.69 (T1), 171.46 (P1), 171.88 (i), 172.23 (K1), 172.54 (R1);

MS m/z calculated for C₃₉H₅₉N₉O₁₃ 861.4; found 862.4 M⁺

Rf = 0.583 (solvent system B);

Purity HPLC 100%.

4.2.2.2.2. Compound MPA-Arg(NO₂)-Pro-Lys(Gly)-Thr-OMe 3i:

Product 3i was obtained with yield 81% as white powder.

MPA-RT-Gly **3i**: ¹H NMR (400 MHz, DMSO-d₆) δ ppm: 1.06 (d, J = 6.6 Hz, 3H, γ -T4), 1.29 $(m, 2H, \gamma - K4), 1.38 (m, 3H, \delta - K5, \beta - R3), 1.46 (m, 3H, \beta - K3, \gamma - R4), 1.65 (m, 2H, \beta - K3, \beta - R3),$ 1.73 (s, 3H), 1.81 (m, 2H, γ -P4, β -P3), 1.88 (m, 1H, γ -P4), 1.98 (m, 1H, β -P3), 2.06 (s,3H), 2.13 (m, 2H), 2.19 (m, 2H), 3.07 (m, 2H, ϵ -K6), 3.11 (m, 2H, δ -R5), 3.26 (d, J = 6.7 Hz, 2H), 3.36 (m, 2H, a-G2), 3.51 (m, 1H, δ -P5), 3.60 (m, 1H, δ -P5), 3.63 (s, 3H, COOCH3), 3.68 $(s, 3H), 4.10 (m, 1H, \beta-T3), 4.26 (m, 1H, \alpha-T2), 4.28 (m, 1H, \alpha-K2), 4.34 (m, 2H, \alpha-P2), 4.45$ (m, 1H, α -R2), 5.12 (t, J = 6.5 Hz, 1H), 5.21 (s, 2H), 7.82 (d, J = 8.4 Hz, 1H, α -TNH), 7.97 (m, 3H, α -RNH, a-KNH, ϵ -KNH);

 13 C NMR (400 MHz,CD₃OD-d₄) δ ppm: 10.04 (e), 14.99 (f), 18.97 (γ-T4), 22.59 (d), 24.65 $(\gamma-K4)$, 28.43 $(\gamma-P4)$, 29.22 $(\gamma-R4)$, 31.05 $(\beta-R3)$, 33.94 $(\delta-K5)$, 35.05 $(\beta-P3)$, 38.65 $(\beta-K3)$, 40.53 (h), 42.17 (g), 42.78 (ϵ -K6), 48.47 (δ -R5), 50.77 (δ -P5), 51.47 (α -R2), 53.28 (OMe), $56.93 (\alpha-K2)$, $57.81 (\alpha-T2)$, $60.11 (\alpha-P2)$, 60.15 (c), $66.99 (\beta-T3)$, 69.33 (b), 106.38 (o), 122.41



(r,l), 123.16 (a), 133.57 (j), 145.29 (m), 159.56 (p), 163.41 (k), 171 (n), 171.13 (T1), 172.56 (P1), 172.97 (i,K1), 173. 25 (R1), 174.23 (G1);

MS m/z calculated for $C_{41}H_{62}N_{10}O_{14}$ 919.0; found 919.4 M⁺.

Rf = 0.454 (solvent system B).

Purity HPLC 100%.

4.2.2.2.3. Compound MPA-Arg(NO₂)-Pro-Lys(α Ala)-Thr-OMe 3j:

Product 3j was obtained with yield 80% as white powder.

MPA-RT-αAla **3j**: 1 HNMR (400 MHz, DMSO-d₆) δ ppm: 1.043 (d, J=5.7 Hz, 3H, γ-T4), 1.211 (d, J=6.9 Hz, 3H, β -A3), 1.290 (m, 2H, γ -K4), 1.380 (m, 3H, δ -K5, β -R3b), 1.5483 (m, 3H, β -K3b, γ -R4), 1.653 (m, 2H, β -K3a, β -R3a), 1.727 (s, 3H, f), 1.802 (m, 2H, γ -P4b, β -P3b), 1.894 $(m, 1H, \gamma-P4a), 1.998 (m, 1H, \beta-P3a), 2.064 (s, 3H, e), 2.128 (m, 2H, g), 2.183 (m, 2H, h),$ 3.043 (m, 2H, ϵ -K6), 3.130 (m, 2H, δ -R5), 3.266 (d, J=6.5 Hz, 2H, d), 3.488 (m, 1H, α -A2), 3.532 (m, 1H, δ -P5b), 3.617 (m, 1H, δ -P5a), 3.625 (s, 3H, COOMe), 3.682 (s, 3H, c), 4.103 $(m, 1H, \beta-T3), 4.257 (m, 1H, \alpha-T2), 4.278 (m, 1H, \alpha-K2), 4.341 (m, 2H, \alpha-P2), 4.442 (m, 1H, \alpha-T2), 4.278 (m, 1H, \alpha-K2), 4.341 (m, 2H, \alpha-P2), 4.442 (m, 1H, \alpha-R2), 4.341 (m, 2H, \alpha-P2), 4.442 (m, 1H, \alpha-R2), 4.341 (m, 2H, \alpha-P2), 4.442 (m, 1H, \alpha-R2), 4.341 (m, 2H, \alpha-P2), 4.442 (m, 2H,$ α -R2), 5.114 (t, J=6.6 Hz, 1H, a), 5.217 (s, 2H, b), 7.827 (d, J=8.4 Hz, 1H, α -TNH), 8.026 (m, 3H, α -RNH, α -KNH, ϵ -KNH);

¹³CNMR (400 MHz,CD₃OD-d₄) δ ppm: 10.06 (e), 14.98 (f), 18.56 (β-A₃), 18.98 (γ-T₄), 22.30 (d), 24.65 (γ -K4), 28.43 (γ -P4), 29.22 (γ -R4), 31.05 (β -R3), 33.96 (δ -K5), 35.07 (β -P3), 38.74 $(\beta-K3)$, 40.53 (h), 49.68 (δ -R5) 50.77 (δ -P5, α -A2), 51.48 (α -R2), 53.32 (OMe), 56.94 (α -K2), 57.81 (α -T2), 60.11 (α -P2), 60.18 (c), 67.01 (β -T3), 69.38 (b), 106.38 (o), 116.13 (l), 122.34 (r), 123.1 (a), 133.65 (j), 145.29 (m), 159.55 (p), 163.42 (k), 171.01 (n), 171.14 (T1), 172.48 (P1), 172.96 (i) 173.23 (K1), 173.80 (R1), 174.22 (A1);

MS m/z calculated for $C_{42}H_{64}N_{10}O_{14}$ 933.0; found 933.5 M⁺.

Rf = 0.480 (solvent system B)

Purity HPLC 97%.

4.2.2.2.4. Compound MPA-Arg(NO_2)-Pro-Lys(β Ala)-Thr-OMe 3k:

Product 3k was obtained with yield 88% as white powder.

MPA-RT-βAla **3k:** ¹HNMR (400 MHz, DMSO-d₆) δ ppm: 1.043 (d, J=7 Hz, 3H, γ -T4), 1.270 (m, 2H, γ -K4), 1.385 (m, 3H, δ -K5, β -R3b), 1.5484 (m, 3H, β -K3b, γ -R4), 1.653 (m, 2H, β -K3a, β -R3a), 1.727 (s, 3H, f), 1.817 (m, 2H, γ -P4b, β -P3b), 1.886 (m, 1H, γ -P4a), 2.00 (m, 1H, β -P3a), 2.068 (s, 3H, e), 2.105 (m, 2H, g), 2,162 (m, 2H, h), 2.444 (t, J=6,8 Hz, 2H, α -A2), 2.946 (t, J=5,3 Hz, 2H, β -A3), 3.026 (m, 2H, ϵ -K6), 3.128 (m, 2H, δ -R5), 3.270 (d, J=6.8 Hz, 2H, d), 3.516 (m, 1H, δ -P5b), 3.615 (m, 1H, δ -P5a), 3.623 (s, 3H, COOMe),



3.686 (s, 3H, c), 4.102 (m, 1H, β -T3), 4.253 (m, 1H, α -T2), 4.282 (m, 1H, α -K2), 4.340 (m, 2H, α -P2), 4.436 (m, 1H, α -R2), 5.114 (t, J=6.9 Hz, 1H, a), 5.222 (s, 2H, b), 7.847 (d, J=8.4 Hz, 1H, α -TNH), 8.043 (m, 3H, α -RNH, α -KNH, ϵ -KNH);

¹³CNMR (400 MHz,CD₃OD-d₄) δ ppm: 10.06 (e), 14.98 (f), 18.98 (γ-T4), 22.31 (d), 24.66 $(\gamma-K4)$, 28.30 $(\gamma-P4)$, 29.22 $(\gamma-R4)$, 31.00 $(\beta-R3)$, 31.96 $(\beta-A3)$, 33.95 $(\delta-K5)$, 35.06 $(\beta-P3)$, $36.06 (\alpha-A2)$, $38.66 (\beta-K3)$, 40.56 (h), 42.151 (g), $46.868 (\epsilon-K6)$, $48.332 (\delta-R5) 50.85 (\delta-P5)$, 51.48 (α -R2), 53.29 (OMe), 56.93 (α -K2), 57.81 (α -T2), 60.18 (c), 67.01 (β -T3), 69.37 (b), 106.37 (o), 116.03 (l), 122.37 (r), 123.12 (a), 133.63 (j), 145.31 (m), 159.56 (p), 163.42 (k), 170.96 (n), 170.99 (T1), 172.51 (P1), 172.98 (i) 173.26 (K1), 174.24 (R1,A1);

MS m/z calculated for $C_{42}H_{64}N_{10}O_{14}$ 933.0; found 933.5 M⁺.

Rf = 0.309 (solvent system B).

Purity HPLC 100%.

4.2.2.2.5. Compound MPA-Arg(NO₂)-Pro-Lys(Val)-Thr-OMe 3l:

Product 31 was obtained with yield 79% as white powder.

MPA-RT-Val **31**: ¹HNMR (400 MHz, DMSO-d₆) δ ppm: 0.807 (d, J=6.9 Hz, 3H, γ-V4), 0.858 (d, J=6.8 Hz, 3H, δ -V5), 1.026 (d, J=4.2 Hz, 3H, γ -T4), 1.279 (m, 2H, γ -K4), 1.369 (m, 3H, δ -K5, β -R3b), 1.461 (m, 3H, β -K3b, γ -R4), 1.627 (m, 2H, β -K3a, β -R3a), 1.714 (s, 3H, f), 1.792 $(m, 2H, \gamma-P4b, \beta-P3b), 1.861 (m, 2H, \gamma-P4a, \beta-V3), 1.977 (m, 1H, \beta-P3a), 2.061 (s, 3H, e),$ 2.114 (m, 2H, g), 2.183 (m, 2H, h), 3.005 (m, 5H, α -V2, ϵ -K6, δ -R5), 3.262 (d, J=6.7 Hz, 2H, d), 3.502 (m, 1H, δ -P5b), 3.604 (m, 1H, δ -P5a), 3.625 (s, 3H, COOMe), 3.676 (s, 3H, c), 4.096 $(m, 1H, \beta-T3), 4.252 (m, 1H, \alpha-T2), 4.269 (m, 1H, \alpha-K2), 4.3 (m, 1H, \alpha-P2), 4.437 (m, 1H, \alpha-P2), 4.437 (m, 1H, \alpha-P2), 4.269 (m, 1H, \alpha-K2), 4.3 (m, 1H, \alpha-P2), 4.437 (m, 1H, \alpha-P2), 4.43$ α -R2), 5.096 (t, J=6,5 Hz, 1H, a), 5.222 (s, 2H, b), 7.803 (d, J=8.4 Hz, 1H, α -TNH), 8.022 (m, 3H, α -RNH, α -KNH, ϵ -KNH);

¹³CNMR (400 MHz,CD₃OD-d₄) δ ppm: 10.05 (e), 14.96 (f), 16.99 (δ-V5), 18.22 (γ -V4), 18.97 $(\gamma-T4)$, 22.28 (d), 24.65 $(\gamma-K4)$, 28.49 $(\gamma-P4)$, 29.22 $(\gamma-R4)$, 31.05 $(\beta-R3)$, 31.56 $(\beta-V3)$, 33.96 $(\delta-K5)$, 35.07 (β -P3), 38.68 (β -K3), 40.53 (h), 42.16 (g), 46.648 (ϵ -K6), 47.904 (δ -R5) 50.74 $(\delta-P5)$, 51.46 (α -R2), 53.33 (OMe), 56.93 (α -T2), 57.80 (α -V2), 60.1 (α -P2), 60.18 (c), 67.01 (β-T3), 69.39 (b), 106.35 (o), 116.26 (l), 122.3 (r), 123.08 (a), 133.67 (j), 145.28 (m), 159.57 (p), 163.42 (k), 170.98 (n), 171.12 (T1), 172.43 (P1), 172.97 (i) 173.22 (K1), 173.41 (R1), 174.21 (V1);

MS m/z calculated for C₄₃H₆₆N₁₀O₁₄ 947.0; found 948.4 [M+H]⁺

Rf = 0.436 (solvent system B)

Purity HPLC 99%.



4.2.2.2.6. Compound MPA-Arg(NO_2)-Pro-Lys(Leu)-Thr-OMe 3m:

Product **3m** was obtained with yield 91% as white powder.

MPA-RT-Leu **3m**: ¹HNMR (400 MHz, DMSO-d₆) δ ppm: 0.77 (m, 6H, δ –L5, ε-L6), 1.014 $(m, 3H, \gamma-T4), 1.22 (m, 10H, \gamma-K4, \beta-L3b, \beta-L3a, \delta-K5, \beta-R3b, \beta-K3b, \gamma-R4), 1.59 (m, 3H, \gamma-T4), 1.22 (m, 10H, \gamma-K4, \beta-L3b, \beta-L3b, \beta-L3a, \delta-K5, \beta-R3b, \beta-K3b, \gamma-R4), 1.59 (m, 3H, \gamma \beta$ -K3a, β -R3a, γ -L4b), 1.69 (s, 3H, f), 1.79 (m, 3H, γ -P4b, β -P3b, γ -L4a), 1.98 (m, 1H, γ -P4a), 2.05 (s, 3H, e), 2.11 (m, 4H, g, h), 3.03 (m, 2H, ϵ -K6), 3.11 (m, 2H, δ -R5), 3.25 (d, 2H, J=6.2 Hz, d), 3.32 (m, 1H, α -L2), 3.52 (m, 2H, δ -P5b, δ -P5a), 3.60 (s, 3H, COOMe), 3.67 (s, 3H, c), 4.09 (m, 1H, β -T3), 4.24 (m, 2H, α -T2, α -K2), 4.32 (m, 1H, α -P2), 4.43 (m, 1H, α -R2), 5.10(t, J=6.6 Hz, 1H, a), 5.21 (s, 2H, b), 7.82 (d, J=8.4 Hz, 1H, α -TNH), 8.08 (m, 2H, α -RNH, α -KNH), 8.12 (m, 1H, ϵ -KNH);

¹³CNMR (400 MHz,CD₃OD-d₄) δ ppm: 11.49 (e), 16.44 (f), 20.48 (γ-T4), 23.39 (d/γ-K4), 23.35 (δ -L5), 24.41 (γ -L4), 24.87 (γ -P4/ γ -R4), 28.83 (β -R3), 29.17 (δ -K5), 29.52 (β -P3), 31.81 (β-K3), 34.25 (h), 35.48 (g), 38.81 (ε-K6), 40.54 (δ-R5) 43.17 (β-L3), 47.18 (δ-P5), 50.26 $(\alpha-R2)$, 52.28 (OMe), 52.74 $(\alpha-L2)$ 52.85 $(\alpha-K2)$, 58.13 $(\alpha-T2)$, 59.58 $(\alpha-P2)$, 60.95 (c), 66.76 (β-T3), 68.98 (b), 122.94 (o/a), 146.11 (m), 159.72 (p), 162.98 (k), 170.48 (n), 171.43 (T1), 171.82 (P1), 172.14 (i) 172.59 (K1);

MS m/z calculated for C₄₅H₇₀N₁₀O₁₄ 975.1; found 975.5 M⁺

Rf = 0.678 (solvent system B)

Purity HPLC 99.5%.

4.2.2.2.7. Compound MPA-Arg(NO_2)-Pro-Lys(Ile)-Thr-OMe 3n:

Product **3n** was obtained with yield 84% as white powder.

MPA-RT-Ile **3n**: ¹HNMR (400 MHz, DMSO-d₆) δ ppm: 0.806 (m, 6H, δ-I5, ε-I6), 1.041 (d, J=6.3 Hz, 3H, γ -T4), 1.078 (m, 2H, γ -I4), 1.273 (m, 2H, γ -K4), 1.393 (m, 3H, δ -K5, β -R3b), 1.459 (m, 3H, β-K3b, γ-R4), 1.634 (m, 3H, β-K3a, β-R3a, β-I3), 1.727 (s, 3H, f), 1.804 (m, 2H, γ -P4b, β -P3b), 1.894 (m, 2H, γ -P4a, β -T3), 1.995 (m, 1H, β -P3a), 2.071 (s, 3H, e), 2.129 (m, 4H, g, h), 3.007 (m, 5H, α -I2, ϵ -K6, δ -R5), 3.272 (d, J=6.5 Hz, 2H, d), 3.515 (m, 1H, δ -P5b), 3.617 (m, 1H, δ -P5a), 3.625 (s, 3H, COOMe), 3.687 (s, 3H, c), 4.103 (m, 1H, β-T3), 4.25 (m, 2H, α -T2, α -K2), 4.339 (m, 1H, α -P2), 4.444 (m, 1H, α -R2), 5.142 (t, J=6.7 Hz, 1H, a), 5.228 (s, 2H, b), 7.817 (d, J=8.2 Hz, 1H, α -TNH), 8.018 (m, 3H, α -RNH, α -KNH, ϵ -KNH); ¹³CNMR (400 MHz,CD₃OD-d₄) δ ppm: 10.07 (e), 10.45 (δ-I5), 14.45 (ε-I6), 14.93 (f), 18.97 $(\gamma-T4)$, 22.29 (d), 24.22 $(\gamma-I4)$, 24.66 $(\gamma-K4)$, 28.48 $(\gamma-P4)$, 29.22 $(\gamma-R4)$, 31.04 $(\beta-R3)$, 33.97 $(\delta-K5)$, 35.08 (β -P3), 38.24 (β -I3), 38.69 (β -K3), 40.53 (h), 42.16 (g), 46.894 (ϵ -K6), 48.357 $(\delta-R5)$ 50.76 $(\delta-P5)$, 51.47 $(\alpha-R2)$, 53.35 (OMe), 56.93 $(\alpha-T2)$, 57.81 $(\alpha-I2)$, 59.08 $(\alpha-K2)$,



60.11 (α-P2), 60.18 (c), 67.02 (β-T3), 69.39 (b), 106.35 (o), 116.2 (l), 122.32 (r), 123.09 (a), 133.66 (j), 145.29 (m), 159.56 (p), 163.42 (k), 170.99 (n), 171.14 (T1), 172.45 (P1), 172.97 (i) 173.23 (K1), 173.42 (R1), 174.21 (I1);

MS m/z calculated for C₄₅H₇₀N₁₀O₁₄ 975.1; found 975.5 M⁺

Rf = 0.635 (solvent system B)

Purity HPLC 98%.

4.3. Investigation of immunosuppressive activity

4.1.Cell lines

T-Jurkat cell line, a type of acute lymphoblastic leukemia was obtained from a cell bank.

4.2. Growing media

The T-Jurkat cell line and PBMCs were suspended in a liquid medium containing RPMI-1640, 10% FBS and P/S. Jurkat cells were grown in sterile culture bottles secured with a sterile filter cap providing free gas exchange

4.3. Jurkat cell line culture

4.3.1. Passage of the cell line

Cells were suspended in the culture medium, placed in a incubator at 37 °C, in a 5% CO₂ atmosphere. Every 2-3 days, were passed by pipetting. Then, the used medium was exchanged for a new one and the culture placed in a thermostat.

4.3.2. Thawing the cell line

Jurkat cells were stored in cryospheres in a cell bank under liquid nitrogen. Cells were thawed and suspension centrifuged. Next, the supernatant was separated and the rest suspended in nutrient solution and pipetted into sterile bottles. Finally, the culture bottles were filled up to 10 mL with culture medium containing RPMI-1640, 10% FBS and P/S.

4.3.3. Freezing the cell line

The cells suspended in the culture medium were centrifuged and the supernatant decanted. The pellet was suspended in medium with the addition (10% final volume) of FBS and 10% (10% final volume) of DMSO. Then, a cell suspension was placed in cryoamphlets and frozen at -80 °C.

4.4. Isolation of PBMCs

Peripheral blood mononuclear cells (PBMCs) were obtained from a buffy coat, which is treated as waste in the process of producing erythrocyte mass. A buffy coat was taken from the Regional Center for Blood Donation and Blood Treatment in Gdańsk, Poland (RCKiK). The



obtained test material came from anonymous and healthy donors. PBMC cells were obtained by centrifugation in a gradient of venous blood density. The blood was diluted with PBS in a 1: 1 ratio. Then, the diluted blood was layered in 15 mL vortex tubes containing 4 mL of Ficoll and Uropoline (1:1), followed by spinning for 18 min. at 1800 rpm. After this time, it a separation to individual fractions in test tubes was observed. Lymphocytes with monocytes formed a "leukocytic" coat at the border of Ficoll and plasma. A buffy coat was carefully collected and rinsed twice with PBS buffer, spinning at 1500 rpm for 5 min. After pouring the buffer, 2 mL of lysis buffer was added to the PBMCs (pH 7.4-7.5) and left at room temperature for 5 min. Subsequently, erythrocytes lysis occurred, and the lysis buffer was diluted with PBS 1: 1 and centrifuged (1500 rpm, 5 min). Depending on the test, PBMC cells were suspended the in the culture medium to obtain 10⁶ cells in 1 ml, or left in PBS.

4.5. Preparation of dilutions of MPA analogs

To prepare the appropriate concentration in eppendorf tube, 1 mg of the compound was dissolved in 20 µL of DMSO, and subsequently diluted with medium to 1 mL, resulting in a concentration of 1 mg/mL. Subsequent concentrations were prepared by serial dilution, obtaining the lowest test concentration of 10-8 mg/mL in the case of MPA and 10-5 mg/mL in the case of MPA derivatives. When performing the repetitions, a respective compound (1 mg) was dissolved each time to avoid the storage of samples in the form of solutions.

4.6. Testing used in biological investigations

4.6.1. Colorimetric MTT test

4.6.1.1. Jurkat cell line

Jurkat cells were transferred from the culture bottle to the centrifuge tube, then swirled (1500 rpm, 5 min) and the spent medium removed. Cells were suspended in culture medium (50 μ L) and placed in 96-well, flat bottom plates at 50×10^3 cells per well. Subsequently, the appropriate dilutions of compounds in triplicate were added to each well, and the plates was kept in an incubator (37 °C, 5% CO₂). After 48 hours of incubation, 20 μL of MTT solution (5mg / ml H₂O) was placed in each well. Then the plate was incubated for a further 3 hours. After the next incubation 100 µL of acidified 0.4 N HCl of isopropanol was added to each well, followed by shaking for 15 minutes to dissolve the precipitated formazan crystals. Then, the intensity of the color at 570 nm using a spectrophotometer was measured.



4.6.1.2. PBMC

The were isolated according to the procedure described in section 4.6.1.1., followed by suspension in an appropriate amount of culture medium (50 µL) and placed in a 96-well flat-bottomed plate at 50×10^3 cells per well. In addition, 1 µL of monoclonal antibodies anti-CD3 / anti-CD28 were added to each well. Then, 50 μL of the compounds were added to the plate in appropriately prepared dilutions in triplicate. The plates were incubated for 72 hours in a heating oven (37 °C, 5% CO₂). Next, 20 µL of MTT solution was added to each well and incubated for 3 hours, followed by acidification (0.4N HCl) with isopropanol. The intensity of the color was read using a wavelength of 570 nm. The test on PBMC cells was performed for three different patients, and the result was averaged.

4.6.2. Proliferation test using VPD450

4.6.2.1. Preparation of the dye solution

Solution 1 mM of VPD450 dye in DMSO was prepared in centrifuge tube DMSO. Then oneoff portions of the dye solution (2 µL) were prepared before use. VPD450 dye solutions were stored at -80 °C without light.

4.6.2.2. Cell staining

The cells $10-30 \times 10^6$ were suspended in 1 mL of PBS and 2 μ L of the previously prepared dye solution (1 mM) was added. The cells were incubated for 15 min in a water bath at 37 °C. Subsequently, the cells were rinsed with PBS and spun. The supernatant was poured off and the cell pellet was rinsed with 10 mL of culture medium and spun. The liquid was separated after rinsing. The color of the cells was checked with flow cytometer before applying.

4.6.2.2.1. Jurkat cell line

Jurkat cells were transferred from the culture bottle to the centrifuge tube (1500 rpm, 5 min) and the spent medium was removed. Jurkat cells were rinsed twice with PBS and stained according to 4.6.2.2. procedure. Next, cells were suspended in culture medium (50 µL) and placed in 96-well, round-bottom plates (50×10^3 cells per well). Subsequently, the appropriate dilutions of compounds in triplicate was added to each well, followed by incubation the plates in an incubator (37 °C, 5% CO₂, 48h). Then, cells with compounds were transferred to 5 mL tubes with rinsing the wells with cold PBS. Subsequently, the cells were centrifuged and the supernatant separated. The pellet was suspended in 300 µL of PBS, stirred and analyzed by flow cytometry.



4.6.2.2.2. PBMC

The cells isolated according to the procedure described in section 2.2.4. were suspended in PBS and stained as described in 4.6.2.2. Stained cells were suspended in an appropriate amount of culture medium (50 μ L) and placed in a 96-well round-bottom plate (50 \times 10³ cells per well). Next, 1 μ L of anti-CD3 / anti-CD28 monoclonal antibodies to each well were added. Then, 50 μ L of the compounds were added to the plate in appropriately prepared dilutions in triplicate. The plates were incubated for 72 hours in a heating oven (37 °C, 5% CO₂). After this time, PBMC cells and compounds were transferred from wells to 5 mL tubes and the wells rinsed with cold PBS. Then the cells were centrifuged and the supernatant was separated. The residue was suspended in 300 μ L of PBS, thoroughly stirred and analyzed by flow cytometry.

4.6.3. Data analysis

The results from both tests were then adjusted with SigmaPlot Software in order to obtain IC₅₀ or EC₅₀ values for each compound.

4.6.4. PBMC in the presence of GMP

The test for the determination of antiproliferative activity using flow cytometry against PBMC cell line in the presence of GMP was carried out analogously as described in 4.6.2. GMP solution (50 μ M) was added to cell culture with respective compound and antibodies.

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