

# Spontaneous Electron Emission Versus Dissociation in Internally Hot Silver Dimer Anions

P. Jasik,<sup>1,2, a)</sup> J. Franz,<sup>1,3</sup> D. Kędziera,<sup>4</sup> T. Kilich,<sup>1,3</sup> J. Kozicki,<sup>1,3</sup> and J.E. Sienkiewicz<sup>1,3</sup>

<sup>1)</sup>*Faculty of Applied Physics and Mathematics, Gdańsk University of Technology, 80-233 Gdańsk, Poland*

<sup>2)</sup>*BioTechMed Center, Gdańsk University of Technology, 80-233 Gdańsk, Poland*

<sup>3)</sup>*Advanced Materials Center, Gdańsk University of Technology, 80-233 Gdańsk, Poland*

<sup>4)</sup>*Faculty of Chemistry, Nicolaus Copernicus University, 87-100 Toruń, Poland*

Referring to a recent experiment, we theoretically study the process of a two-channel decay of the diatomic silver anion ( $\text{Ag}_2^-$ ), namely the spontaneous electron ejection giving  $\text{Ag}_2 + e^-$  and the dissociation leading to  $\text{Ag}^- + \text{Ag}$ . The ground state potential energy curves of the silver molecules of diatomic neutral and negative ion were calculated using proper pseudo-potentials and atomic basis sets. We also estimated the non-adiabatic electronic coupling between the ground state of  $\text{Ag}_2^-$  and the ground state of  $\text{Ag}_2 + e^-$ , which in turn allowed us to estimate the minimal and mean values of the electron autodetachment lifetimes. The relative energies of the rovibrational levels allow the description of the spontaneous electron emission process, while the description of the rotational dissociation is treated with the quantum dynamics method as well as time-independent methods. The results of our calculations are verified by comparison with experimental data.

## I. INTRODUCTION

The problem of coupling nuclear and electronic motions is still an area where much work needs to be done<sup>1</sup>. Solving this problem fully nonadiabatically, without the separation of electronic and nuclear motions, is challenging even for the simplest molecule  $\text{H}_2$ , although for hydrogen isotopologues it has been solved with extraordinary accuracy<sup>2</sup> including both relativistic<sup>3</sup> and QED<sup>4</sup> effects. Currently, less accurate nonadiabatic calculations can be done for slightly heavier molecules, such as few-electron diatomics<sup>5,6</sup>. Treating electrons and nuclei on the same footing seems rather impossible for many-electron polyatomic molecules, but there is significant progress in incorporating nuclear quantum effects and non-Born-Oppenheimer effects, for specified nuclei, into calculations using nuclear-electronic orbital approach<sup>7</sup>. Nevertheless, most of the progress is made using the well-established concept of interatomic potentials, usually calculated using the Born-Oppenheimer approximation with added nonadiabatic couplings between different electronic states<sup>8</sup>. This approach has proven to be quite effective in describing ultrafast reorganization of electronic density during and after excitation by intense ultra-short electromagnetic pulses<sup>9</sup>. This kind of process should be studied in the attosecond regime. Already, a great deal of effort is being put into the experimental and theoretical research which is leading to the emergence of a new field called attosecond science<sup>10</sup>.

Quite the opposite situation takes place, when the nonadiabatic coupling is minimal, which is the case when the Born-Oppenheimer potential curves are energetically far apart for all values of  $R$ , i.e. outside the areas of avoided crossings. In the absence of an electromagnetic field, the spontaneous

process may last much longer and even reach seconds. Recently such a process has been measured in the cryogenic ion-beam storage ring DESIREE for small copper and silver clusters of anions<sup>11,12</sup> and very recently for silver dimer anions<sup>13</sup>. During the first direct observation of the spontaneous single-molecule decay of an internally hot diatomic silver anion molecule ( $\text{Ag}_2^-$ ), two reaction pathways were discovered. The first path is the fragmentation of the molecular anion into a neutral atom ( $\text{Ag}$ ) and an atomic anion ( $\text{Ag}^-$ ), the second is the spontaneous emission of an electron from the molecular anion. The decay process was explained by the Morse potential energy curves generated from the experimental parameter values for neutral and anion silver dimers.

Our goal is to theoretically describe the spontaneous decay process of silver anion dimer. The first step is to calculate the potential energy ground states of the anion and neutral silver dimers. Based on the rovibrational levels, we are able to characterise the decay channels and compare our results with the experimental data. We focus on the critical value of the rotational quantum number  $J_c$  related to a certain oscillating quantum number  $v$ . Having the ground state potential energy curve of an anionic system, we calculate the quantum dynamics (QD) of the dissociation process, which allows us to calculate the widths and lifetimes of high-lying rovibrational levels. The predissociation lifetimes for quasibond states with small widths of levels are treated with a time-independent approach. We also present the non-adiabatic coupling matrix element between chosen initial and final vibrational states of  $\text{Ag}_2^-$  and  $\text{Ag}_2$ , which allow us to estimate the spontaneous electron emission lifetimes.

Our calculation methods are described in Sec. II. In Sec. III, we discuss the results and compare them with the available experimental data. Conclusions are given in Sec. IV.

<sup>a)</sup>Electronic mail: patryk.jasik@pg.edu.pl

## II. COMPUTATION METHODS

### A. Potential energy curves

We treat the calculations of the interaction between the silver atom and its negative ion analogously to the interaction between two neutral silver atoms. The same electric axial field of the nuclei affects all electrons. The only significant change resulting from the different number of electrons lies in the different configurations of the electronic states. Since the effective electrical action on the valence electron in the anion is relatively weak, some convergence problems can be expected. The adiabatic states with which we are concerned are computed with the Born-Oppenheimer approximation, i.e. as solutions of the following time-independent Schrödinger equation

$$H^{el}\Psi_i^{el}(\vec{r};R) = E_i^{el}(R)\Psi_i^{el}(\vec{r};R), \quad (1)$$

where the internuclear distance  $R$  is kept fixed, vector  $\vec{r}$  represents all electronic coordinates,  $H_{el}$  is the electronic Hamiltonian of a diatomic system,  $\Psi_i^{el}(\vec{r};R)$  describes the  $i$ -th eigenstate of the Hamiltonian,  $E_i^{el}(R)$  is the corresponding eigenvalue, i.e. the  $i$ -th adiabatic potential. The Hamiltonian of the system can be written as

$$H^{el} = H_A + H_B + V_{AB}, \quad (2)$$

where  $H_A$  and  $H_B$  are the Hamiltonians of the isolated atoms and  $V_{AB}$  is the interaction between them<sup>14,15</sup>. In the current approach, electrons from the 4s, 4p, 4d, and 5s subshells are treated explicitly, while the Ag core, containing the nucleus and 28 electrons occupying inner shells, is represented by the pseudo-potential. The total Hamiltonian in Eq.(1, 2) can be expressed as

$$H^{el} = T + V. \quad (3)$$

Here  $T$  represents the kinetic energy operator of the valence electrons, and  $V$  is the interaction operator between valence electrons and Ag cores. The latter can be written in the following form:

$$V = V^A + V^B + \sum_{j>1}^N \frac{1}{r_{ij}} + V_{cc}. \quad (4)$$

Here  $V^A$  describes Coulomb and exchange interaction as well as the Pauli repulsion between the valence electrons and the Ag core. We use the following semi-local energy-consistent pseudopotentials<sup>16</sup>:

$$V^A = \sum_{i=1}^N \left( -\frac{Q_A}{r_{Ai}} + \sum_{l,k} B_{l,k}^A \exp(-\beta_{l,k}^A r_{Ai}^2) P_l^A \right), \quad (5)$$

where  $Q_A = 19$  denotes the net charge of the Ag-core,  $P_l^A$  is the projection operator onto the Hilbert subspace of angular symmetry  $l$  with respect to the Ag<sup>19+</sup>-core and the number of the valence electrons is  $N = 19$  or 20 for neutral or anion molecules, respectively. The parameters  $B_{l,k}^A$  and  $\beta_{l,k}^A$  define the semi-local energy-consistent pseudo-potentials<sup>16</sup>.

The second term in equation (4) is fully analogous to the first but concerns the second Ag-core. The third term stands for the repulsion between the valence electrons, whereas the last term describes the interaction between Ag cores.

The core electrons of the Ag atom are represented by the energy-consistent pseudo-potential<sup>16</sup> which was formed from basis set converged towards the complete basis set limit<sup>17</sup>. Note that the spin-orbit (SO) effect makes a very small contribution to the ground energies of our systems, as it only increases the depth of the well by only 24.5 cm<sup>-1</sup>, so we do not include it in our calculations.

The first goal of our calculations is to recover the electron affinity (EA) of silver, knowing that the experimental value given by Bilodeau<sup>18</sup> is 10521 cm<sup>-1</sup>. We perform CCSD(T) calculations with Peterson's aug-cc-pV5Z-PP basis set<sup>17</sup>, but the obtained result equals 10215 cm<sup>-1</sup> is 306 cm<sup>-1</sup> below the experimental value. Then we decide to take the core-valence correlation into account and switch cc-pwCV5Z-PP basis set, augmented with the aug-cc-pV5Z-PP exponents. The obtained values of 10608 cm<sup>-1</sup> is much closer to the experimental one but overestimates it by 87 cm<sup>-1</sup>. Interestingly, the aug-cc-pV5Z-PP basis set with core-valence correlation turned on leads to values almost the same as for the aug-cc-pwCV5Z-PP basis sets.

The problem may lie in the objective function (OF) used in the optimisation of Peterson's basis set, which was set as average singles-doubles configuration interaction (CISD) energy of  $ns^2(n-1)d^9$  and  $ns^1(n-1)d^{10}$  states of a neutral atom. In our case, we are only interested in the ground states of Ag and Ag<sup>-</sup>, so we decide to optimize our basis set using the sum of the lowest energy of the neutral state Ag and its anion with the following objective function:

$$OF = E_{FC_{CCSD},Ag} + E_{FC_{CCSD},Ag^-} + 0.5(E_{CV_{CCSD},Ag} + E_{CV_{CCSD},Ag^-}) + 0.01(E_{CC_{CCSD},Ag} + E_{CC_{CCSD},Ag^-}), \quad (6)$$

where  $E_{X,Y}$  is the electronic energy component  $X$ :  $FC_{CCSD}$ -frozen-core CCSD energy,  $CV_{CCSD}$  core-valence correlation CCSD energy and  $CC_{CCSD}$  core-core correlation CCSD energy of the system  $Y = \{Ag, Ag^-\}$ . Such an objective function may be seen as a more convenient alternative to the three-step method presented by Puzzarini<sup>17</sup>, in which the HF energy and weighted correlation energy components were optimized separately. We keep the number of exponents of the augmented cc-pwCV5Z-PP basis. Instead of directly optimizing the Gaussian exponents, according to Peterson<sup>19</sup>, we expand them into orthonormal Legendre polynomials, up to six terms, and optimize the expansion coefficients. After finalizing the optimization step, we contract the  $s$ ,  $p$  and  $d$  exponents using atomic natural orbital (ANO) coefficients. In addition, we obtain uncontracted mid-bond (MB) 8s6p4d2f exponents optimized on the sum of the frozen-core CISD energy of Ag<sub>2</sub> (at the experimental  $R_e=2.5303$  Å)<sup>20</sup> and Ag<sub>2</sub><sup>-</sup> (at the experimental  $R_e=2.604$  Å)<sup>21</sup> calculated with augmented cc-pwCVQZ-PP basis set.

To calculate adiabatic potential energy curves of Ag<sub>2</sub> and Ag<sub>2</sub><sup>-</sup> diatomic molecules we use the multiconfigurational self-consistent field/complete active space self-consistent field

(MCSCF/CASSCF) method followed by the multi-reference configuration interaction (MRCI) method and single reference coupled cluster method with singles and doubles, and perturbative triples (CCSD(T)). The corresponding active space includes the molecular counterparts of the 4s, 4p, 4d, and 5s orbitals of the Ag atoms. The interaction energies are corrected for the basis set superposition error (BSSE) by counterpoise (CP) correction. For comparison, we also perform calculations with the original basis sets of Peterson and Puzzarini<sup>17</sup> and we assign the obtained potential curves as PP. All calculations are performed by means of the MOLPRO program package<sup>22</sup>. Using these computational methods, we obtain adiabatic potential energy curves for the ground state  $^2\Sigma_u^+$  of  $\text{Ag}_2^-$  and the ground state  $^1\Sigma_g^+$  of  $\text{Ag}_2$ . The new sPYtroscope<sup>23</sup> program is used to calculate rovibrational levels, spectroscopic parameters and the non-adiabatic coupling matrix element between the selected initial and final vibrational states of  $\text{Ag}_2^-$  and  $\text{Ag}_2$ .

## B. Non-adiabatic coupling elements

Before the ejection of an electron, the molecular anion  $\text{Ag}_2^-$  is in the electronic state  $1^2\Sigma_u^+$  with the electronic wave function  $\Psi_i^{(N+1)}(\vec{r}_1, \dots, \vec{r}_{N+1}; R)$ . After the electron is ejected, the wave function is given by

$$\Psi_f(\vec{r}_1, \dots, \vec{r}_{N+1}; R; E_{\text{cont}}) = \mathcal{A} \left( \Psi_f^N(\vec{r}_1, \dots, \vec{r}_N; R) \cdot \phi_{\text{cont}}(\vec{r}_{N+1}; R; E_{\text{cont}}) \right).$$

Here,  $\mathcal{A}$  is the antisymmetrisation operator and  $E_{\text{cont}}$  is the energy of the ejected electron.  $\Psi_f^N(\vec{r}; R)$  is the wave function of the the electronic ground state  $1^1\Sigma_g^+$  of the neutral molecule  $\text{Ag}_2$  and  $\phi_{\text{cont}}(\vec{r}_{N+1}; R; E_{\text{cont}})$  is the continuum wave function of the ejected electron.

The non-adiabatic coupling element between two electronic states is given by

$$\Lambda_{fi}(R, E_{\text{cont}}) = \int \Psi_f^*(\vec{r}_1, \dots, \vec{r}_{N+1}; R; E_{\text{cont}}) \frac{\partial}{\partial R} \Psi_i(\vec{r}_1, \dots, \vec{r}_{N+1}; R) d\vec{r}_1 \dots d\vec{r}_{N+1}, \quad (7)$$

where integration includes all electrons. The coupling elements depend on the internuclear distance  $R$  and the energy  $E_{\text{cont}}$  of the ejected electron. Here we neglect the terms with second derivatives. The non-adiabatic coupling matrix element between initial and final vibrational states is calculated as

$$\Lambda_{v'f,vi}(E_{\text{cont}}) = \frac{\hbar^2}{\mu_{\text{Ag}_2}} \int \chi_{v'f}^*(R) \Lambda_{fi}(R, E_{\text{cont}}) \frac{\partial}{\partial R} \chi_{vi}(R) dR, \quad (8)$$

where  $\mu_{\text{Ag}_2}$  is the reduced mass of the  $\text{Ag}_2$  molecule.  $\chi_{vi}$  is the initial vibrational wave function of the anion and  $\chi_{v'f}$  is the final vibrational wave function of the neutral molecule.

The probability of the electron autodetachment per unit time is proportional to the square of the element of the non-adiabatic vibrational coupling matrix<sup>24</sup> and can be expressed

as follows

$$\Gamma_{ad} \sim \frac{2\pi}{\hbar} |\Lambda_{v'f,vi}|^2. \quad (9)$$

The lifetimes of the spontaneous electron emission  $\tau$  are then estimated as the inverse of  $\Gamma_{ad} = \hbar/\tau$ .

The essential part of the above approach is the computation of the non-adiabatic coupling matrix elements  $\Lambda_{fi}$  between the ground states of the molecular anion and the neutral molecule. Neutral silver dimer in the electronic ground state has a closed-shell electron configuration. The ab initio wave function of the electronic ground state is dominated by the Hartree-Fock determinant. Therefore, in the computation of the electronic coupling elements, the many-electron wave functions are approximated by single Slater determinants with Kohn-Sham orbitals. The wave function of the anion is constructed by adding an extra electron into the lowest unoccupied orbital  $\phi_{\text{bound}}$  of the neutral molecule and assuming that the occupied orbitals do not change. The non-adiabatic coupling element simplifies to

$$\Lambda_{fi}(R, E_{\text{cont}}) = \int \phi_{\text{cont}}^*(\vec{r}, R; E_{\text{cont}}) \frac{\partial}{\partial R} \phi_{\text{bound}}(\vec{r}, R) d\vec{r}. \quad (10)$$

Here the integration is over the coordinates of the ejected electron.

The continuum wave function  $\phi_{\text{cont}}(\vec{r}, R; E_{\text{cont}})$  of the ejected electron is computed in a quantum scattering calculation using the single centre expansion (SCE) method<sup>25</sup> and Kohn-Sham density functional theory. The scattering potential that describes the interactions between the ejected electron and the neutral molecule is given by

$$v_{\text{tot}} = v_{st} + v_{cp} + v_{ex}, \quad (11)$$

where  $v_{st}$  is the electrostatic potential,  $v_{cp}$  is the correlation-polarisation potential, and  $v_{ex}$  is the exchange potential. All potentials are functions of the electron density of the neutral molecule. The correlation-polarisation potential is approximated using the generalized gradient approximation of Lee, Yang and Parr (LYP)<sup>26</sup>. For large distances, the potential is switched to the asymptotic form

$$v_{cp} \rightarrow -\frac{\alpha}{r^4}, \quad (12)$$

where  $\alpha$  is the polarisability of the neutral molecule. The switching between the LYP and the long-range potential is done by a switching function around the outermost crossing point  $r_c$  of the two potentials. The exchange potential is approximated by the modified semi-classical exchange potential<sup>27</sup>. For large distances ( $r \geq r_c$ ) from the target the exchange potential is represented by the form similar to the one suggested by Silkowski and Pachucki<sup>28</sup>

$$v_{ex} \rightarrow -br^{5/2}e^{-ar}, \quad (13)$$

where the parameters  $a$  and  $b$  are determined by fitting to the radial points around  $r_c$ .

All calculations have been performed for a number of nuclear geometries. In each geometry, a single centre expansion of the bound orbital  $\phi_{\text{bound}}(\vec{r}, R)$  is done with SCELlib4 of Sanna *et al.*<sup>29</sup>. The expansion is largely dominated, with a coefficient greater than 0.9, by the  $p$ -wave that is aligned along the molecular axis. For all further calculations only the  $p$ -wave is considered,  $\phi_{\text{bound}}(\vec{r}, R)$  is renormalised, and  $\frac{\partial}{\partial R} \phi_{\text{bound}}(\vec{r}, R)$  is computed by numerical differentiation using central-difference derivatives.

For each nuclear geometry the electron density is computed with density functional theory. These computations are done with the program package Gaussian<sup>30</sup> using the B3LYP (Becke<sup>31</sup>, three-parameter, Lee-Yang-Parr<sup>26</sup>) exchange-correlation functional, the effective core potentials of Hay and Wadt<sup>32,33</sup>, and the double-zeta basis set of Dunning and Hay<sup>34</sup>. The single centre expansion of the electron density is done with the program package SCELlib4<sup>29</sup>.

The electron scattering calculations to compute  $\phi_{\text{cont}}(\vec{r}, R; E_{\text{cont}})$  are done with the program package Bumblebee<sup>35</sup>. The program solves the scattering equations in the presence of the potential  $v_{\text{tot}}$  by a hybrid propagator. For short ranges (up to about 10 Å) R-matrix propagation<sup>36</sup> is used. For large ranges, the variable phase approach<sup>37</sup> is used and the equations are integrated to the distance where the phase shifts converge using the embedded Runge-Kutta Prince-Dormand (8,9) method<sup>38</sup>. In the asymptotic region the normalisation of the wave function is done and the boundary conditions are propagated inwards. The method of Kulander and Light<sup>39</sup> is used to construct the wave function from the R-matrices<sup>40</sup>. Whenever possible the program uses the algorithms implemented in the GNU scientific library (GSL)<sup>41</sup>. Only the  $s$ -wave and the  $p$ -wave (aligned along the molecular axis) are considered. For the evaluation of the integral in Eq. (10) the integrand is represented as Akima-spline<sup>42</sup>. A spline is a piecewise polynomial, which is integrated analytically, as discussed in Chapter 9 in Ueberhuber<sup>43</sup>.

### C. Quantum dynamics

The time-dependent approach which is mathematically equivalent to the time-independent one can be regarded as a complementary tool giving other insight into physical processes<sup>44,45</sup>. In our case, we use the time-dependent method to investigate the rotational predissociation of highly excited rovibrational states of the  $\text{Ag}_2^-$ , which is one of the competitive reactions in the problem under consideration. We start our consideration from the time-dependent Schrödinger equation written in the following form

$$i\hbar \frac{\partial}{\partial t} \Phi(R, t) = H_J^{\text{nuc}} \Phi(R, t), \quad (14)$$

where  $\Phi(R, t)$  is the time-dependent wave packet moving along the effective potential energy curve, and  $H_J^{\text{nuc}}$  is the normal nuclear Hamiltonian.

By definition the wavepacket is a coherent superposition of stationary states (e.g. Tannor<sup>46</sup>) which may be represented in

the following form consisting of two contributions from the discrete and continuous parts of the spectrum

$$\begin{aligned} \Phi(R; t) = & \sum_{v, J} c_{v, J} \Psi_{v, J}^{\text{nuc}}(R) e^{-iE(v, J)t/\hbar} + \\ & + \int c_J(E) \Psi_{E, J}(R) e^{-iEt/\hbar} dE, \end{aligned} \quad (15)$$

where

$$c_{v, J} = \int_0^\infty \Psi_{v, J}^{\text{nuc}}(R)^* \Phi(R; 0) dR$$

and

$$c_J(E) = \int_0^\infty \Psi_{E, J}(R)^* \Phi(R; 0) dR$$

are the energy-dependent coefficients, squares of these coefficients form the spectral distribution of  $\Phi$  normalized to 1,  $e^{-iE(v, J)t/\hbar}$  and  $e^{-iEt/\hbar}$  are the time evolution factors,  $\Psi_{v, J}^{\text{nuc}}(R)$  and  $\Psi_{E, J}(R)$  are eigenfunctions of  $H_J^{\text{nuc}}(R)$ . The wavepacket  $\Phi(R; t)$  is a solution of Eq.14 and its initial shape at  $t = 0$  is taken as a Gaussian function of arbitrary half-width placed on the effective potential energy curve. The wavepacket moves away from its starting location due to the Newtonian force  $-dU_J/dR$ . This process is described by the time-dependent autocorrelation function

$$S(t) = \int \Phi(R; t = 0)^* \Phi(R; t) dR. \quad (16)$$

In our case, the autocorrelation function describes evolution of the initial nuclear wavepacket in the ground electronic state of the molecular anion  $\text{Ag}_2^-$ . The time-dependent population in the range till  $R_{\text{max}}$  for the particular state labeled by  $J$ , in accordance with the effective potential energy  $U_J$ , is calculated as

$$P(t) = \int_0^{R_{\text{max}}} |\Phi(R; t)|^2 dR. \quad (17)$$

We determine the spectrum by the inverse Fourier transform of  $S(t)$ <sup>47,48</sup> as follows

$$\sigma(E(v, J)) = \int_{-\infty}^\infty e^{iE(v, J)t/\hbar} S(t) dt. \quad (18)$$

In our calculations, the above integral is estimated in the range  $\langle 0, T \rangle$  using Fast Fourier Transform (FFT) procedures<sup>49</sup>.

The propagation time is assumed to be 250 ps, which is sufficient to estimate the integral in Eq. 17. In Eq. 16, we set  $R_{\text{max}}$  to be equal to 100  $a_0$ . There are  $2^{14}$  points in the integration grid. To avoid the diffraction between the outgoing and the incoming waves as a result to bouncing off the boundary at  $R_{\text{max}}$ , the negative imaginary potential is placed at 60  $a_0$ . This potential smoothly absorbs the wavepacket near the boundary. All quantum-dynamic computations are performed using parts of the YADE platform<sup>50-53</sup>.

TABLE I. Spectroscopic parameters  $R_e$  [Å],  $D_e$ ,  $D_0$ ,  $\omega_e$ ,  $B_e$ , and EA [ $\text{cm}^{-1}$ ] of the  $1^1\Sigma_g^+$  state of  $\text{Ag}_2$ . PP represents the original aV5Z/awCV5Z basis set of Peterson and Puzzarini<sup>17</sup>, CP represents the counterpoise correction, MB represents the mid-bonds, and MB+CP represents the mid-bonds with counterpoise correction.

Author	$R_e$	$D_e$	$D_0$	$\omega_e$	$B_e$	EA $\text{Ag}_2$
present, MRCI	2.570	11165	11078	175.6	0.04765	8377
present, CCSD(T)	2.523	13743	13645	195.8	0.04946	8094
present, CCSD(T) CP	2.523	13695	13597	195.6	0.04941	8065
present, CCSD(T) MB	2.517	14091	13992	197.2	0.04962	8088
present, CCSD(T) MB+CP	2.523	13825	13727	195.9	0.04944	8053
present PP, CCSD(T)	2.523	13825	13727	196.1	0.04947	8159
present PP, CCSD(T) CP	2.523	13750	13653	195.4	0.04938	8163
present PP, CCSD(T) MB	2.517	14273	14174	198.5	0.04972	8120
present PP, CCSD(T) MB+CP	2.523	13844	13746	196.0	0.04945	8145

TABLE II. Spectroscopic parameters  $R_e$  [Å],  $D_e$ ,  $D_0$ ,  $\omega_e$ , and  $B_e$  [ $\text{cm}^{-1}$ ] of the  $1^2\Sigma_u^+$  state of  $\text{Ag}_2^-$ . PP represents the original aV5Z/awCV5Z basis set of Peterson and Puzzarini<sup>17</sup>, CP represents the counterpoise correction, MB represents the mid-bonds, and MB+CP represents the mid-bonds with counterpoise correction.

Author	$R_e$	$D_e$	$D_0$	$\omega_e$	$B_e$
present, MRCI	2.685	10407	10339	136.7	0.04372
present, CCSD(T)	2.642	11288	11215	146.2	0.04493
present, CCSD(T) CP	2.649	11211	11138	145.9	0.04485
present, CCSD(T) MB	2.636	11631	11557	148.6	0.04516
present, CCSD(T) MB+CP	2.649	11329	11256	146.3	0.04490
present PP, CCSD(T)	2.642	11351	11278	146.8	0.04495
present PP, CCSD(T) CP	2.649	11280	11207	146.1	0.04483
present PP, CCSD(T) MB	2.636	11761	11686	150.2	0.04531
present PP, CCSD(T) MB+CP	2.642	11356	11283	146.5	0.04492

TABLE III. The comparison of present spectroscopic parameters  $R_e$  [Å],  $D_e$ ,  $D_0$ ,  $\omega_e$ ,  $B_e$ , and EA [ $\text{cm}^{-1}$ ] of the  $1^1\Sigma_g^+$  state of  $\text{Ag}_2$  with other experimental and theoretical results.

Author	$R_e$	$D_e$	$D_0$	$\omega_e$	$B_e$	EA $\text{Ag}_2$
present, CCSD(T) CP	2.523	13695	13597	195.6	0.04941	8065
Laher (exp.) <sup>20</sup>	2.530		13308	192.4	0.04881	
Morse (exp.) <sup>54</sup>	2.480		$13308 \pm 242$	192.4	0.05121	
Peterson and Puzzarini (theory) <sup>17</sup> aV5Z/awCV5Z	2.524	13714		196.0		
Peterson and Puzzarini (theory) <sup>17</sup> CBS	2.523	13766		196.2		
Hay et al. (theory) <sup>55</sup>	2.617	10703		172.0		
Zhang et al. (theory) <sup>56</sup>	2.644			160.0		
Stoll et al. (theory) <sup>57</sup>	2.688	13147		142.0		
Bonacic-Koutecky et al. (theory) <sup>58</sup>						8630
Ho et al. (exp.) <sup>21</sup>						$8251 \pm 56.5$

### III. RESULTS AND DISCUSSION

In Tab. I and Tab. II, we present all our spectroscopic parameters for  $\text{Ag}_2$  and  $\text{Ag}_2^-$  molecules, respectively. All calculations are performed using our optimised 5Z-type basis set and the original basis set aV5Z/awCV5Z provided by Peterson and Puzzarini<sup>17</sup> (PP) and the different computational approaches described in Section II A.

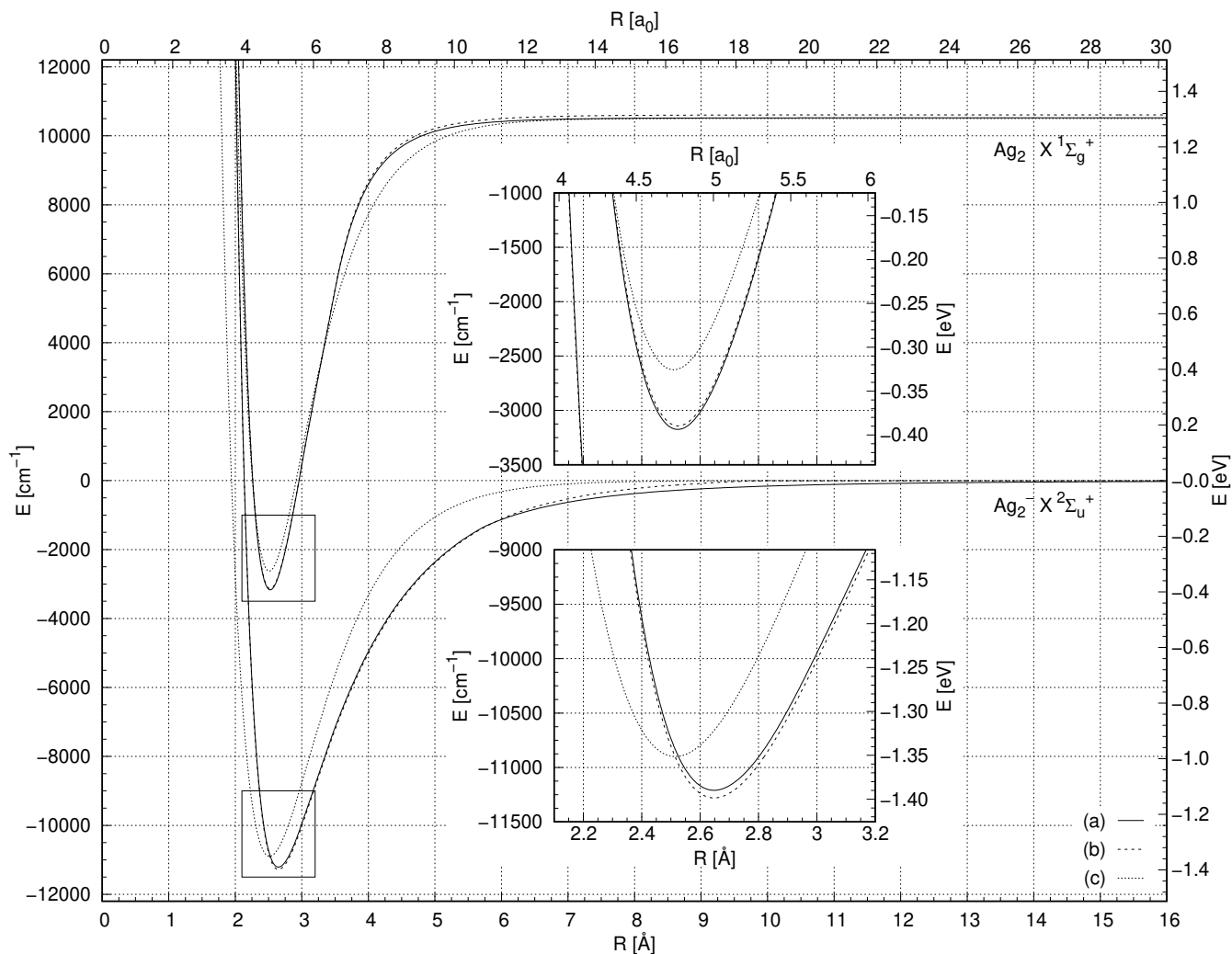
An in-depth analysis of the calculated spectroscopic parameters and potential energy curves carried out in relation to all available experimental results (see Tab. III and Tab. IV) al-

lows us to conclude that for both systems the best agreement is obtained using the optimised basis set and coupled-cluster (CC) method with counterpoise correction. For the selected computational approach, the estimated root mean squared error (RMSE) and mean absolute error (MAE) are the smallest with respect to the experimental data and are respectively 126 and 71. The same errors calculated for potentials obtained by using PP basis set and CC method with CP correction are larger reaching values equal to 138 and 74. Other combinations of basis sets and methods lead to much larger errors.

In Fig. 1 we show our potential energy curves calculated with the use of selected approaches (optimised and PP basis

TABLE IV. The comparison of spectroscopic parameters  $R_e$  [ $\text{\AA}$ ],  $D_e$ ,  $D_0$ ,  $\omega_e$ , and  $B_e$  [ $\text{cm}^{-1}$ ] of the  $1^2\Sigma_u^+$  state of  $\text{Ag}_2^-$  with other experimental and theoretical results.

Author	$R_e$	$D_e$	$D_0$	$\omega_e$	$B_e$
present, CCSD(T) CP	2.649	11211	11138	145.9	0.04485
Bonacic-Koutecky et al. (theory) <sup>58</sup>	2.780				
Spasov et al. (theory) <sup>59</sup>					0.04220
Ho et al. (exp.) <sup>21</sup>	$2.604 \pm 0.007$		$11050 \pm 1291$	$145 \pm 10$	

FIG. 1. Potential energy curves for the  $\text{Ag}_2^-$  and  $\text{Ag}_2$  ground states: (a) current work; (b) current work PP; (c) Morse potentials generated by experimental and theoretical parameters<sup>13</sup>.

sets + CC with CP) for the ground states of both systems. A visual comparison can be made with the Morse potential curves<sup>13</sup> plotted for  $\text{Ag}_2^-$  with the experimental parameters<sup>21</sup> and for  $\text{Ag}_2$  with the theoretical parameters<sup>55</sup>. In the case of  $\text{Ag}_2^-$ , the divergence between the two curves is quite pronounced, especially for the attractive part of the potential and smaller for the repulsive part. In the case of  $\text{Ag}_2$ , the differences in the positions of the potential wells between the parameterized Morse potential<sup>13</sup> and our theoretical results remain within a reasonable range. On the other hand, for  $\text{Ag}_2^-$ ,

the analogous difference is much greater, but it should not seriously affect the interpretation of the experimental data.

For  $\text{Ag}_2$  spectroscopic parameters (see Tab. III), our results show better agreement with more recent Laher experimental data<sup>20</sup> than those reported by Morse<sup>54</sup>. As for  $\text{Ag}_2^-$  (see Tab. IV), the agreement between our results and the only available experimental data of Ho et al.<sup>21</sup> is quite satisfactory.

The electron affinity (EA) value plays a key role. In fact, the spontaneous emission of electrons is allowed because EA of  $\text{Ag}_2$  is less than the dissociation energy of  $\text{Ag}_2^-$ . Then in

Tab. III and V, we show our calculated molecular and atomic EA. Although the atomic values are fully in line with the experimental data, the molecular values remain around  $200\text{ cm}^{-1}$  below the experimental data.

TABLE V. Electron affinity EA [ $\text{cm}^{-1}$ ] of the Ag atom.

Author	EA Ag
Bilodeau et al. (exp.) <sup>18</sup>	10521
present, CCSD(T) CP	10524
present PP, CCSD(T) CP	10608
present, MRCI	9116
Peterson and Puzzarini <sup>17</sup> CBS	10598
Peterson and Puzzarini <sup>17</sup> aV5Z/awCV5Z	10573
Stoll et al. <sup>57</sup>	9033

The analysis of the effective potential energy curves along with the increase of the rotational quantum number  $J$  (Fig. 2) shows that from  $J = 254$ , spontaneous emission of electrons from  $\text{Ag}_2^-$  is not possible. Then we define  $J_c = 253$  as the

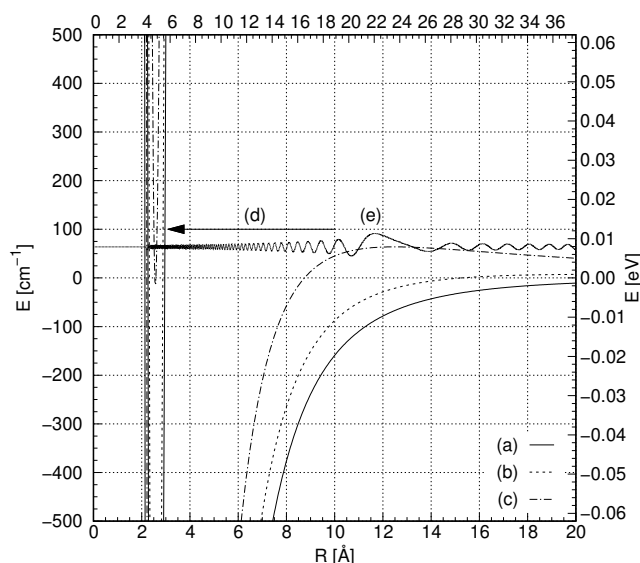


FIG. 2. Effective potential energy curves of  $\text{Ag}_2^-$  and  $\text{Ag}_2$  ground states; (a)  $J=0$ ; (b)  $J=150$ ; (c)  $J=254$ ; (d) spontaneous electron emission is no longer possible for  $J \geq 254$ ; (e) vibrational wave function for  $\nu=155$ ,  $J=254$ .

critical value, i.e. the highest  $J$  value for which spontaneous emission is allowed. In this case, only the lowest vibration level of the neutral system lies below the highest vibration levels of the anionic system. Our map of the rovibrational levels of  $\text{Ag}_2^-$  and  $\text{Ag}_2$  molecules (Fig. 3) shows the regions defined by the rotational quantum numbers  $J$  and energies. A much more accurate division of the energy scale above  $-200\text{ cm}^{-1}$  allows for better recognition of transitions between different decay paths. What is not visible everywhere, each line has a discrete structure. Each part of these lines corresponds to an energy of a given rovibrational level defined by a pair of quantum numbers  $(\nu, J)$ .

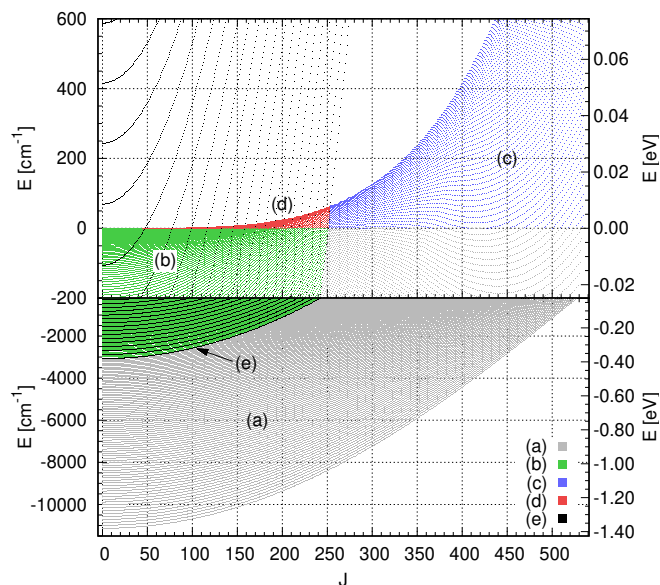


FIG. 3. A map of rovibrational levels of  $\text{Ag}_2^-$  and  $\text{Ag}_2$ : (a) no decay possible, the gray points are rovibrational levels of  $\text{Ag}_2^-$ , but only those lying below the lowest level of  $\text{Ag}_2$ ; (b) spontaneous electron emission only, the green points are rovibrational levels of  $\text{Ag}_2^-$ , but only those lying above the lowest levels of  $\text{Ag}_2$  and below zero; (c) fragmentation only, the blue points are quasi-bound rovibrational levels of  $\text{Ag}_2^-$ , no mixing with rovibrational levels of  $\text{Ag}_2$ ; (d) both decay modes possible, the red points are quasi-bound rovibrational levels of  $\text{Ag}_2^-$ , but only those mixing with the rovibrational levels of  $\text{Ag}_2$ ; (e) the black points are rovibrational levels of  $\text{Ag}_2$ .

The spectrum of the effective potential energy curve for  $J_c + 1$  obtained using quantum dynamics in the YADE framework<sup>50–53</sup> (the autocorrelation function in Eq. 18) from the Gaussian wavepacket  $\Phi(R; t=0)$  is presented in Fig. 4 and compared with the energies of the vibrational levels obtained from the sPYtroscopy program<sup>23</sup>. QD calculations are very accurate for quasibond states with level widths larger than approximately  $10^{-4}\text{ cm}^{-1}$  as opposed to time-independent methods implemented in Level<sup>60</sup> or sPYtroscopy<sup>23</sup>. We calculate all the level widths and predissociation lifetimes for the rotational quantum number  $J$  from 0 to  $J_c + 1 = 254$  using QD for highly excited rovibrational levels of  $\text{Ag}_2^-$  and time-independent methods for all other rovibrational levels. Based on these calculations, and assuming that the maximal lifetime of the fragmentation reaction is less than 10 seconds, which corresponds to the duration of the experiment reported by Anderson *et al.*<sup>13</sup>, we estimate the mean value of the predissociation lifetime to be approximately 224 ms.

For quasibond states with large level widths, the lifetimes of highly-lying rovibrational levels are calculated from the vibrational wave functions using the recently described method<sup>45</sup>. The obtained exemplary lifetimes (by fitting the population to the exponential decay  $e^{-t/\tau}$ )<sup>45</sup> are  $\tau_{(J=254, \nu=154)} = 204\text{ ns}$  and  $\tau_{(J=254, \nu=155)} = 57\text{ ps}$ , where the latter one corresponds to the highest vibrational level. Their corresponding level widths (i.e. the full width at half maxi-

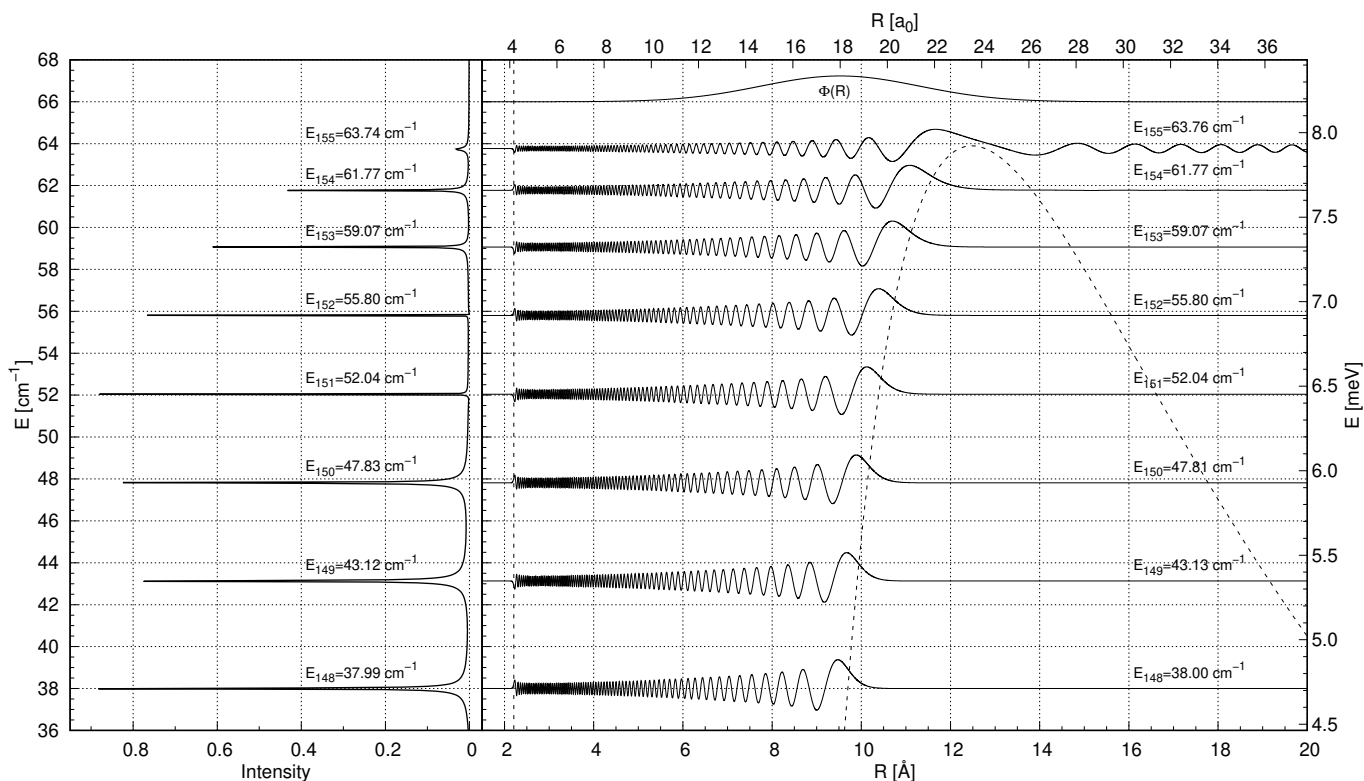


FIG. 4. (left panel) Vibrational energy levels for the effective potential with  $J_c + 1 = 254$  obtained by autocorrelation function (Eq. 18) calculated from quantum dynamics of the Gaussian packet  $\Phi(R; t = 0)$  (shown on the top of the right panel) using the YADE software<sup>45,50–53</sup> are compared with energy levels and vibrational wave functions derived from the sPYtroscope program<sup>23</sup> (right panel).

imum, FWHM) are  $\Gamma_{(J=254, v=154)}^{YADE} = 0.26 \times 10^{-4} \text{ cm}^{-1}$  and  $\Gamma_{(J=254, v=155)}^{YADE} = 0.093 \text{ cm}^{-1}$ . The respective values obtained from the LEVEL program<sup>60</sup> using time-independent treatment are  $\Gamma_{(J=254, v=154)}^{LEVEL} = 0.15 \times 10^{-4} \text{ cm}^{-1}$  and  $\Gamma_{(J=254, v=155)}^{LEVEL} = 0.072 \text{ cm}^{-1}$ . The above analysis leads us to the statement that the fragmentation reaction ( $\text{Ag}_2^- \rightarrow \text{Ag} + \text{Ag}^-$ ) for the highly-lying rovibrational levels is very fast. But for quasi-bond states with smaller level widths, it reaches milliseconds or even a few seconds.

Matrix element of an electronic non-adiabatic coupling element calculated from Eq. 10 for six different energies of the ejected electron is shown in Fig. 5. The kinetic energies of the ejected electron equal to the differences between the initial and final rovibronic states are in the range from 10 meV to 50 meV and are typical for the case under consideration. The mentioned electronic non-adiabatic couplings allow the calculation of the vibrational coupling element between selected states according to Eq. 8. Assuming that the energy of the detached electron is in the range given above, and the maximal lifetime of the electron emission reaction is less than 10 s, which corresponds to the duration of the experiment reported by Anderson *et al.*<sup>13</sup>, then the mean value of this coupling element between all possible rovibronic states of  $\text{Ag}_2^-$  and  $\text{Ag}_2$  for  $J = 0$  is  $1.4 \cdot 10^{-9}$  a.u. Averaging is performed by all possible pairs of vibrational levels, for which energy gaps are in the range related to the kinetic energy of the ejected electrons.

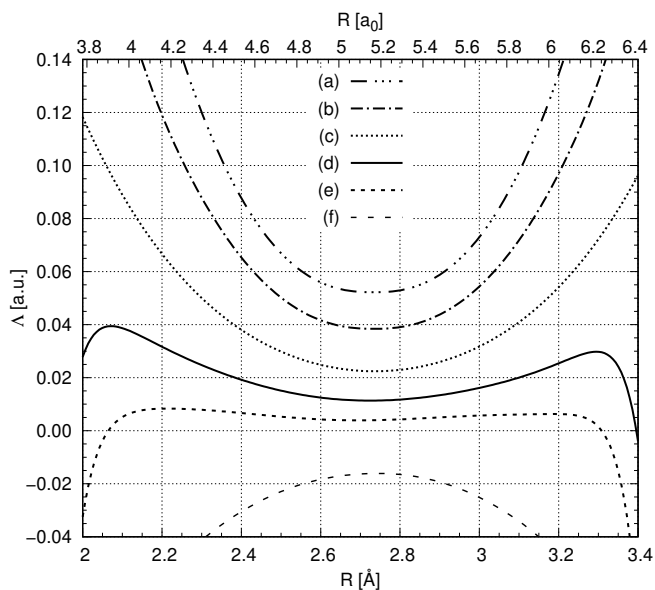


FIG. 5. Electronic non-adiabatic coupling elements for chosen energies of detached electron: (a)  $E_{\text{cont}} = 50 \text{ meV}$ ; (b)  $E_{\text{cont}} = 40 \text{ meV}$ ; (c)  $E_{\text{cont}} = 30 \text{ meV}$ ; (d)  $E_{\text{cont}} = 24 \text{ meV}$ ; (e)  $E_{\text{cont}} = 20 \text{ meV}$ ; (f)  $E_{\text{cont}} = 10 \text{ meV}$ .

As expected, the value of the vibrational coupling element is



very small. In turn, the mean lifetime of the spontaneous electron emission ( $\text{Ag}_2^- \rightarrow \text{Ag}_2 + e^-$ ) as estimated by Eq. (9), is around 3 seconds, and the shortest calculated lifetime is 262 ms.

All this confirms the experimental finding that both processes (i.e. spontaneous electron emission and rotational predissociation) compete with each other in the same timescale and their duration can be quite long, in specific cases up to a few seconds.

#### IV. CONCLUSIONS

In reference to the recent experiment, we set ourselves the goal of creating a theoretical description of the two-channel decay of a diatomic silver anion molecule containing spontaneous emission of electrons and defragmentation. We calculated the electronic ground states of anion and neutral diatomic silver molecules. Systematic calculations revealed the Morse-like shape potentials in both cases, with a much more pronounced deviation for the attractive part of the anion potential. The map of the rovibrational levels of both systems shows the dependence of the decay channels on the rotational quantum number  $J$  and the energy of the vibronic levels. This is broadly in line with the experimental conclusions. The region where both channels are open is relatively small. The highest value of  $J$ , for which both channels are still open called the critical  $J_c$ , is 253. In the case of  $J$  higher than the critical  $J_c$ , only the defragmentation channel is open.

Electronic non-adiabatic coupling elements were calculated as a function of the kinetic energy of the ejected electron. Since then, there has been no crossing or avoided-crossing of the potential curves, its value is small, which makes the calculated mean value of the vibrational non-adiabatic coupling between all considered rovibrational levels very small, of the order of  $10^{-9}$  a.u. This means that the spontaneous electron emission process can be long and has a mean lifetime estimated by our approach of a few seconds, although the shortest lifetimes are on a millisecond scale (i.e.  $\sim 250$  ms).

In turn, the mean lifetime of the predissociation reaction calculated for the considered quasibond states is smaller compared to the electron autodetachment reaction and equals around 250 ms. It means that the fragmentation reaction is generally faster than the spontaneous electron emission reaction occurring even on pico- and nanosecond scale, but some overlap in time (starting with milliseconds) of both channels can be observed, leading to vie of these two decay pathways of the  $\text{Ag}_2^-$  molecule. All the above confirms the experimental results on the timescale of reactions given by Anderson *et al.*<sup>13</sup>.

#### ACKNOWLEDGEMENTS

We acknowledge partial support from the COST Action "Attosecond Chemistry" (Grant No. CA18222). This work was supported by computer grants from the computer centres WCSS (Wrocławskie Centrum Siciowo-Superkomputerowe,

Politechnika Wroclawska) and CI TASK (Centrum Informatyczne Trójmiejskiej Akademickiej Sieci Komputerowej, Politechnika Gdańska). J. F. thanks Nico Sanna for helpful discussions.

#### DATA AVAILABILITY

The data that support the findings of this study are available from the corresponding author upon reasonable request and are openly available in MOST Wiedzy at <https://mostwiedzy.pl/en/open-research-data-series/niema,202103311024423091181-0/catalog>.

- <sup>1</sup>B. Sutcliffe, *Theoretical Chemistry Accounts* **140**, 23 (2021).
- <sup>2</sup>K. Pachucki and J. Komasa, *The Journal of Chemical Physics* **144**, 164306 (2016).
- <sup>3</sup>M. Puchalski, A. Spyszkiwicz, J. Komasa, and K. Pachucki, *Phys. Rev. Lett.* **121**, 073001 (2018).
- <sup>4</sup>M. Puchalski, J. Komasa, P. Czachorowski, and K. Pachucki, *Phys. Rev. Lett.* **122**, 103003 (2019).
- <sup>5</sup>C. E. Scheu, D. B. Kinghorn, and L. Adamowicz, *The Journal of Chemical Physics* **114**, 3393 (2001).
- <sup>6</sup>S. Bubin, M. Stanke, and L. Adamowicz, *The Journal of Chemical Physics* **131**, 044128 (2009).
- <sup>7</sup>F. Pavosevic, T. Culpitt, and S. Hammes-Schiffer, *Chemical Reviews* **120**, 4222 (2020).
- <sup>8</sup>J. R. Reimers, L. K. McKemmish, R. H. McKenzie, and N. S. Hush, *Phys. Chem. Chem. Phys.* **17**, 24641 (2015).
- <sup>9</sup>M. F. Kling and M. J. J. Vrakking, *Annual Review of Physical Chemistry* **59**, 463 (2008).
- <sup>10</sup>P. B. Corkum and F. Krausz, *Nature Physics* **3**, 381 (2007).
- <sup>11</sup>K. Hansen, M. H. Stockett, M. Kaminska, R. F. Nascimento, E. K. Anderson, M. Gatchell, K. C. Chartkunchand, G. Eklund, H. Zettergren, H. T. Schmidt, and H. Cederquist, *Phys. Rev. A* **95**, 022511 (2017).
- <sup>12</sup>E. K. Anderson, M. Kaminska, K. C. Chartkunchand, G. Eklund, M. Gatchell, K. Hansen, H. Zettergren, H. Cederquist, and H. T. Schmidt, *Phys. Rev. A* **98**, 022705 (2018).
- <sup>13</sup>E. K. Anderson, A. F. Schmidt-May, P. K. Najeeb, G. Eklund, K. C. Chartkunchand, S. Rosen, A. Larson, K. Hansen, H. Cederquist, H. Zettergren, and H. T. Schmidt, *Phys. Rev. Lett.* **124**, 173001 (2020).
- <sup>14</sup>M. Wiatr, P. Jasik, and J. E. Sienkiewicz, *Phys. Scr.* **90**, 054012 (2015).
- <sup>15</sup>M. Wiatr, P. Jasik, T. Kilich, J. E. Sienkiewicz, and H. Stoll, *Chem. Phys.* **500**, 80 (2018).
- <sup>16</sup>D. Figgen, G. Rauhut, M. Dolg, and H. Stoll, *Chem. Phys.* **311**, 227 (2005).
- <sup>17</sup>K. A. Peterson and C. Puzzarini, *Theor. Chem. Acc.* **114**, 283 (2005).
- <sup>18</sup>R. C. Bilodeau, M. Scheer, and H. K. Haugen, *Journal of Physics B: Atomic, Molecular and Optical Physics* **31**, 3885 (1998).
- <sup>19</sup>G. A. Petersson, S. Zhong, J. A. Montgomery, and M. J. Frisch, *J. Chem. Phys.* **118**, 1101 (2003).
- <sup>20</sup>R. R. Laher, M. A. Khakoo, M. Kuzmanovic, V. Bojovic, and A. Antic-Jovanovic, *J. Quant. Spectrosc. Radiat. Transfer* **112**, 786 (2011).
- <sup>21</sup>J. Ho, K. M. Ervin, and W. C. Lineberger, *J. Chem. Phys.* **93**, 6987 (1990).
- <sup>22</sup>H.-J. Werner, P. J. Knowles, G. Knizia, F. R. Manby, M. Schütz, P. Celani, W. Györfy, D. Kats, T. Korona, R. L. A. Mitrushenkov, G. Rauhut, K. R. Shamasundar, T. B. Adler, R. D. Amos, A. Bernhardsson, A. Berning, D. L. Cooper, M. J. O. Deegan, A. J. Dobbyn, F. Eckert, E. Goll, C. Hampel, A. Hesselmann, G. Hetzer, T. Hrenar, G. Jansen, C. Köppl, Y. Liu, A. W. Lloyd, R. A. Mata, A. J. May, S. J. McNicholas, W. Meyer, M. E. Mura, A. Nicklass, D. P. O'Neill, P. Palmieri, D. Peng, K. Pflüger, R. Pitzer, M. Reiher, T. Shiozaki, H. Stoll, A. J. Stone, R. Tarroni, T. Thorsteinsson, and M. Wang, "Molpro, version 2012.1 and a package of ab initio programs," (2012), <http://www.molpro.net>.
- <sup>23</sup>P. Jasik, "sPYtroscopy - the computer program, written in Python, for computing rovibrational spectra of diatomic molecules including an arbitrary number and types of couplings between electronic states," (2020), unpublished.



- <sup>24</sup>N. Douguet, S. Fonseca dos Santos, M. Raoult, O. Dulieu, A. E. Orel, and V. Kokoouline, *The Journal of Chemical Physics* **142**, 234309 (2015).
- <sup>25</sup>F. A. Gianturco and A. Jain, *Phys. Rep.* **143**, 347 (1986).
- <sup>26</sup>C. Lee, W. Yang and R. G. Parr, *Phys. Rev. B* **37**, 785 (1988).
- <sup>27</sup>F. A. Gianturco and S. Scialla, *J. Phys. B: At. Mol. Phys.* **20**, 3171 (1987).
- <sup>28</sup>M. Silkowski and K. Pachucki, *J. Chem. Phys.* **152**, 174308 (2020).
- <sup>29</sup>N. Sanna, G. Morelli, S. Orlandini, M. Tacconi, and I. Baccarelli, *Comput. Phys. Comm.* **248**, 106970 (2020).
- <sup>30</sup>M. J. Frisch *et al.*, in *Gaussian 09, Revision A.02* (Gaussian, Inc., Wallingford CT, 2016).
- <sup>31</sup>A. D. Becke, *J. Chem. Phys.* **98**, 5648 (1993).
- <sup>32</sup>P. J. Hay and W. R. Wadt, *J. Chem. Phys.* **82**, 299 (1985).
- <sup>33</sup>P. J. Hay and W. R. Wadt, *J. Chem. Phys.* **82**, 270 (1985).
- <sup>34</sup>T. H. Dunning Jr. and P. J. Hay, in *Modern Theoretical Chemistry, Vol. 3*, edited by H. F. Schaefer III (Plenum, New York, 1977) pp. 1 – 28.
- <sup>35</sup>J. Franz, “Bumblebee - the computer program package for electron-molecule scattering,” (2020), unpublished.
- <sup>36</sup>E. B. Stechel, R. B. Walker and J. C. Light, *J. Chem. Phys.* **69**, 3518 (1978).
- <sup>37</sup>F. Calogero, *Variable Phase Approach to Potential Scattering* (Academic Press, New York, 1967).
- <sup>38</sup>J. R. Dormand and P. J. Prince, *J. Comput. Appl. Math.* **6**, 19 (1980).
- <sup>39</sup>K. C. Kulander and J. C. Light, *J. Chem. Phys.* **73**, 4337 (1980).
- <sup>40</sup>J. Franz, M. Gustafsson and G. Nyman, *Mon. Not. R. Astron. Soc.* **414**, 3547 (2011).
- <sup>41</sup>M. Galassi, *et al.*, in *GNU Scientific Library Reference Manual (3rd Ed.)*, ISBN 0954612078. (GSL – version 2.6, 2020) <https://www.gnu.org/software/gsl/>.
- <sup>42</sup>H. Akima, *J. Assoc. Comput. Mach.* **17**, 589 (1970).
- <sup>43</sup>C. W. Ueberhuber, *Numerical Computation (Volume 1)* (Springer Verlag, Berlin, 1997).
- <sup>44</sup>P. Jasik, J. E. Sienkiewicz, J. Domsta, and N. E. Henriksen, *Phys. Chem. Chem. Phys.* **19**, 19777 (2017).
- <sup>45</sup>P. Jasik, J. Kozicki, T. Kilich, J. E. Sienkiewicz, and N. E. Henriksen, *Phys. Chem. Chem. Phys.* **20**, 18663 (2018).
- <sup>46</sup>D. J. Tannor, *Introduction to quantum mechanics: a time-dependent perspective* (University Science Books, Sausalito, 2007).
- <sup>47</sup>R. Schinke, *Photodissociation Dynamics* (Cambridge University Press, Cambridge, 1993).
- <sup>48</sup>P. Billingsley, *Probability and measure* (John Wiley & Sons, New York - Chichester - Brisbane - Toronto -Singapore, 1995).
- <sup>49</sup>M. Frigo and S. G. Johnson, *Proceedings of the IEEE* **93**, 216 (2005), special issue on “Program Generation, Optimization, and Platform Adaptation”.
- <sup>50</sup>V. Šmilauer *et al.*, in *Yade Documentation 2nd ed* (The Yade Project, 2015) <http://yade-dem.org/doc/>.
- <sup>51</sup>J. Kozicki and F. V. Donzé, *Computer Methods in Applied Mechanics and Engineering* **197**, 4429 (2008).
- <sup>52</sup>J. Kozicki and F. V. Donzé, *Engineering Computations* **26**, 786 (2009).
- <sup>53</sup>The Yade Project, in *Yade Publications* (The Yade Project, 2020) <https://www.yade-dem.org/doc/publications.html>.
- <sup>54</sup>M. D. Morse, *Chem. Rev.* **86**, 1049 (1986).
- <sup>55</sup>P. J. Hay and R. L. Martin, *J. Chem. Phys.* **83**, 5174 (1985).
- <sup>56</sup>H. Zhang and K. Balasubramanian, *J. Chem. Phys.* **98**, 7082 (1993).
- <sup>57</sup>H. Stoll, P. Fuentealba, P. Schwerdtfeger, J. Flad, L. v. Szentpály, and H. Preuss, *The Journal of Chemical Physics* **81**, 2732 (1984).
- <sup>58</sup>V. Bonacic-Koutecky, L. Cespiva, P. Fantucci, J. Pittner, and J. Koutecky, *J. Chem. Phys.* **100**, 490 (1994).
- <sup>59</sup>V. A. Spasov, T. H. Lee, J. P. Maberry, and K. M. Ervina, *J. Chem. Phys.* **110**, 5208 (1999).
- <sup>60</sup>R. J. L. Roy, *Journal of Quantitative Spectroscopy and Radiative Transfer* **186**, 167 (2017).

